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Comparison of the method of classes and the quadrature of moment for the modelling of Neodymium Oxalate Precipitation

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Introduction

- **Oxalic precipitation :**
 - to deal with radioactive waste and recover the actinides **lanthanides** → To facilitate the development of experimental methods
- **Modelling approach:**
 - Experimentation → Thermodynamics + kinetics & numerical methods

[1] [2] [7] [4] [5]

Thermodynamics et kinetic laws

- Supersaturation ratio (S) → The driving force of the precipitation process
- Kinetic laws of **nucleation**, cristal growth and **agglomeration** = f (S)

Homogeneous primary nucleation

$$R_N = 310^{31} \exp\left[-\frac{67600}{RT}\right] \exp\left[-\frac{187}{(LnS)^2}\right]$$

$S > 50$ $293K < T < 333K$

[1] [8] [9]

Crystal growth

$$G = 2,910^{-6} \exp\left[-\frac{14000}{RT}\right] (Ps)^{1/5} (S-1)$$

$293K < T < 333K$

[2] [3]

Agglomeration

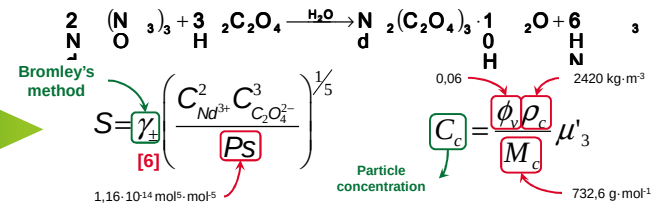
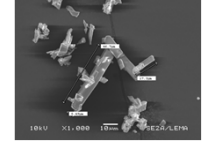
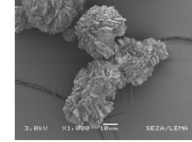
$$\beta = 2,5510^{-7} \cdot I^{-0,70} \cdot S \cdot \gamma^{-0,24} \exp\left[-\frac{40900}{RT}\right]$$

$S > 61$
 $293K < T < 333K$
 $45s^{-1} < \dot{\gamma} < 1024s^{-1}$
 $600mol \cdot m^{-3} < I < 2100mol \cdot m^{-3}$

[7]

$\dot{\gamma} = \sqrt{\frac{\bar{\epsilon}}{\nu}}$ $\bar{\epsilon} = \frac{Np \cdot N^3 \cdot D^5}{V}$

$Nd_2(C_2O_4)_3 \cdot 10 H_2O$ obtained for precipitation different conditions

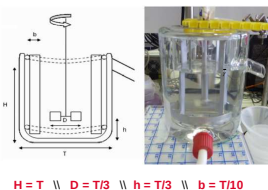


Experimental Study

MSMPR

- Continuous precipitation until steady state

- $V = 200$ ml
- Temperature = $20^\circ C$
- four stainless steel baffles
- stainless steel four 45° pitched blade turbine → $Np = 1,5$

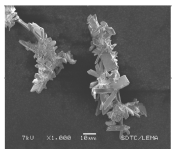


$H = T \parallel D = T/3 \parallel h = T/3 \parallel b = T/10$

Continuous experiments

- Short mean residence time ≈ 1 min → high S
- Constant agglomeration kernel : β
- Scanning Electron Microscopy → crystal size distributions

MEB observations of Loose agglomerates



	N tr/min	$C_{Nd,0}$ mol·m ⁻³	$C_{Ox,0}$ mol·m ⁻³	S^{-1}	$L_{4,3}$ μm
Run 1	1000	142,2	213,7	362	65
Run 2	2000	142,2	213,7	1024	41

Two population balances

Method of classes

Elementary particles: $\frac{dN'_k}{dt} + \frac{N'_k}{\tau} = 0^k R_N + \frac{1}{\Delta L_k} [F(G_{k+1/2}) - F(G_{k-1/2})]$

Loose agglomerates: $\frac{dN_k}{dt} + \frac{N_k}{\tau} = \frac{N'_k}{\tau} + B_k - D_k$

■ 3rd order scheme of Koren [4]

■ scheme of Litster [5]

Quadrature of Moments

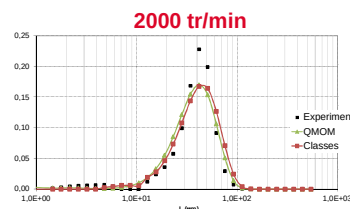
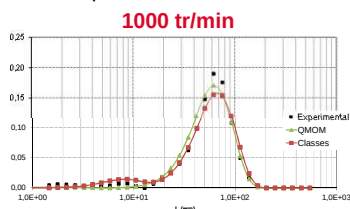
for $k = 0, 1, 2, \dots, 2n-1$

Equation for μ'_k : $\frac{d\mu'_k}{dt} + \frac{\mu'_k}{\tau} = 0^k R_N + k G \mu'_{k-1}$

Equation for μ_k : $\frac{d\mu_k}{dt} + \frac{\mu_k}{\tau} = \frac{\mu'_k}{\tau} + \frac{\beta}{2} \sum_{i=1}^n w_i \sum_{j=1}^n w_j (L_i^3 + L_j^3)^{k/3} - \frac{\beta}{2} \sum_{i=1}^n w_i L_i^k \sum_{j=1}^n w_j$

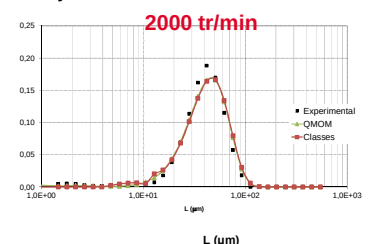
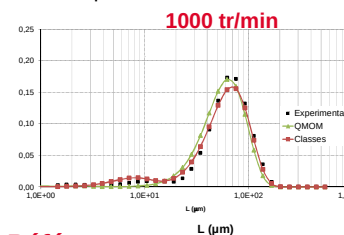
Main results

- Comparison of volume fractions at 5τ



- Crystal size distributions → Experimental vs. Predicted

- Comparison of volume fractions at steady state



Conclusions

- Kinetic laws (R_N , G et β) and loose agglomerates from experimental runs
- Two population balance models : solved by the **Method of classes** and **QMOM**
- Both methods compared well with experimental data during transient and at steady state
- **QMOM** required much less computational effort and is preferentially used with the reconstruction method detailed in [10]

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