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From research to industry

Thermal analysis of a large geopolymer mortar monolith (2.7 m³) in an industrial waste container

5 april 2022

Pascal GIRARDIN

Traitement and conditionning of radioactive waste



Alternative Energies and Atomic Energy Commission (CEA) - www.cea.fr



Summary

- Mortars based on geopolymers have been proposed for the conditioning of magnesium-alloy scraps. The qualification of the immobilization process requires the confirmation that, during the setting of the geopolymer matrix, the temperature in the waste container would not exceed a threshold where adversarial consequences could be observed, such as increased scrap corrosion (accompanied with hydro-gen production) or structural matrix disorders.
- An experimental campaign has culminated in 2020 with the pouring of a large geopolymer mortar quantity (2.7 m3) in an industrial-size cubic container, and the monitoring of the ensuing temperature transient inside the mortar and on the package walls. We present a numerical thermal model based on finite elements, we justify the hypothesis and adopted material parameters, and we compare the obtained numerical simulation with the experimental result.
- The comparison between the measured temperatures and the calculated ones is rather acceptable. The agreement regarding the maximum temperature inside the mortar is especially satisfactory. Sensibility analyses have shown that the temperature peak is not dependent on the external cooling fluxes (irradiance and atmospheric convection), but only on the mortar thermal properties.



Context – Issue - Objective

- Context : More than 1600 tons of magnesium-alloy scraps are currently stored in interim repositories at Marcoule, waiting for a definitive conditioning and disposal
 - Included, 600 tons with the status {Low-Medium activity, Short Life}, to be definitely stored in a surface repository, after immobilization inside a 2.7 m³ cubic container

▶ Issue : Mg is highly reactive : contact with cementitious matrices leads to H₂ production

- Geopolymer mortars appear to be advantageous candidates
- During setting, heat generation and temperature increase, which activates corrosion and H₂ production
- Laboratory tests suggest an admissible temperature ceiling of 60 °C
- Requirement to guarantee this maximal temperature in a range of operating and incidental situations, in large monoliths

Objectives : to build and validate a numerical thermal model of the monolith setting

- Definition of a thermo-kinetic model of the setting heat generation, and to investigate its parameters
- Preparation of a 3-D finite element model
- Validation of the model though the available experimental results, and analyze of the thermal transient



Experimental program

- Experimental program addressing several operational issues
 - Many technical topics being investigated : preparation of the activation solution, mortar mixing, transfer, pouring
 and filling, rheology, sedimentation, geometrical stability, impact of the solid waste on the filling, ..., and the setting
 thermal peak
 - Outline :
 - 2012-: Solidification tests in small instrumented canisters (120 I, 220 I)
 - 2013/10 : Solidification of a 1 m³ monolith
 - 2020/09 : Manufacturing of an industrial-sized 2,7 m³ monolith, inside its container











Modelling the setting heat

The setting of the geopolymer matrix

- A mix of reactions : dissolution-rearrangement of polymers-precipitation
- As for cementitious materials, a usual way to model this reaction mix is to aggregate them :
 - Definition of an apparent reaction advancement, based on the generated heat
 - Definition of three parameters : apparent activation energy, affinity function, specific enthalpy

$$\Theta_{volumic} = \rho_{mortar} \cdot H_{specific} \cdot \frac{d\xi}{dt} = \rho_{mortar} \cdot H_{specific} \cdot \tilde{A}(\xi) \cdot e^{\left(-Ea(T,\xi)/_{RT}\right)}$$

- Observation of the elastic modulus G' and viscous modulus G" during the setting of a geopolymer paste (without sand), at different temperatures (by a team lead by Dr A. Poulesquen, CEA)
- Plotting of the time necessary to reach the maximum of the loss function $tan(\delta) = G''/G'$
- Conclusion : $Ea \sim 74.5 kJ/mol$

Evaluation of $H_{specific}$. $\tilde{A}(\xi)$ with calorimetric test

- Reference mortar : $H_{specific} \sim 78 kJ/kg$









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Apparent activation energy $Ea(T,\xi)$

Vindication of the considered value $Ea \sim 74.5 kJ/mol$

- Calorimetric measurements,

at different temperature conditions in an environmental chambe Bependance of the apparent activation energy on the reaction advancement



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Evaluation of material characteristics

- Thermal conductivity
 - Measured by R&D team at INSA-Rennes led by Aveline Darquennes (associated R&D project 'DECIMAL'), with a 'Hot disk' device
 - On the solidified mortar, but still wet $\lambda_{mortier} = 1.87 W/K/m$

Specific heat

- Measured here at the CEA/LCBC, with a microcalorimetric device
- On the solidified mortar, but still wet $: C_{p_mortier} = 1515 J/K/kg$

Specific enthalpy

- Measured here at at the CEA/LCBC, with a Langavant calorimeter, on several instances
- Nominal value $H_{specific} \sim 78 kJ/kg$, but Minimal value $H_{specific} \sim 76 kJ/kg$, and Maximal value $H_{specific} \sim 90 kJ/kg$
- Impact of input material properties, mixing method and duration, ...



Geometrical and thermal modelling of the 2020 industrial size experiment

Thermal modeling with the CEA Cast3M software

- Geometrical details (inside the box) : air gap at the apex, internal basket
- 10125 nodes, 180-hour transient, 37 thermal sensors, inside and outside the box
- Thermal conduction, heat generation, external fluxes (atmospheric convection, radiative transfer)





Simulation results ; Comparison with experiment

► Temperature peak at 09:30 Maximal temperature



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CQ

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Simulation results ; Comparison with experiment



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Time (hours) Pascal Girardin, CEA/DES/ISEC/SEAD/LCBC





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Simulation results ; Comparison with experiment







Sensibility analysis

Parameters

- Emissivity
- External temperature
- h : internal thermal transfert efficiency
- Apparent activation energy
- Specific enthalpy

| Parameter | Maximal temperature in the mortar (°C) | Deviation from the nominal temperature (simulation) (°C) |
|---|--|--|
| Experimentation measurement | 70.0052 | -0.2953 |
| Simulation, nominal values | 70.3005 | |
| Emissivity $\varepsilon = 0.75$ | 70.3074 | 0.0069 |
| Emissivity ε = 1.0 | 70.2988 | -0.0017 |
| $T_{air} = 10 \ ^{\circ}\mathrm{C}$ | 70.2638 | -0.0367 |
| <i>T_{air}</i> = 25 °C | 70.3674 | 0.0669 |
| $h_{apex} = 8 \text{ W/m}^2/\text{K}$ | 70.3009 | 0.0004 |
| $h_{apex} = 20 \text{ W/m}^2/\text{K}$ | 70.2988 | -0.0017 |
| $h_{contact}$ = 100 W/m ² /K | 70.3056 | 0.0051 |
| $h_{contact}$ = 2000 W/m ² /K | 70.3001 | -0.0004 |
| h_{pehd} = 40 W/m ² /K | 70.3114 | 0.0109 |
| h _{pehd} = 150 W/m²/K | 70.2883 | -0.0122 |
| $h_{grille} = 40 \text{ W/m}^2/\text{K}$ | 70.3087 | 0.0082 |
| $h_{grille} = 250 \text{ W/m}^2/\text{K}$ | 70.2961 | -0.0044 |
| Ea = 65 kJ/mol | 70.2184 | -0.0821 |
| Ea = 84 kJ/mol | 70.4165 | 0.116 |
| $H_{massique} = 75000 \text{ J/kg}$ | 68.5398 | -1.7607 |
| $H_{massique} = 92000 \text{ J/kg}$ | 78.6093 | 8.3088 |

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Some conclusions of the experimental/numerical campain

- ► Confirmed peak temperature in a 2.7 m³ waste-free form at 70°C
 - Peak temperature (with waste) should be smaller but close to the 60°C limit
- Industrial size heat transient : The peak temperature is independent of the outside cooling phenomena (convection, irradiance)
- ► Usefullness of the sensitivity analysis, to priorize the parameter precision needs
- Impact of the geometrical scale : issues at small scale (120 l, 220 l) can be very different than those at large scale (irradiance, ...)
 Maximal temperature







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Next steps and investigations

Thermal impact of the metallic waste

- Evaluation of an homogenized thermal conductivity (with the help of the Mori-Tanaka method) $\lambda_{mortier} = 1.87 W/K/m$, estimated $\lambda_{enrobé} = 2.47W/K/m$
- Improvement of our knowledge about the enthalpy fluctuations (depending on the input material properties)
- **•** Evaluation of the H₂ generation in the waste container due to the heat peak



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Thank you for your attention

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