## **Supplementary Information**

## Characterization of Lithium Phosphorus Oxide Thin Film Libraries by Laser-Induced Breakdown Spectroscopy Imaging: A Step Towards High-Throughput Quantitative Analyses

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  - Specific data on Gate and Gain for each spectrometer

	Spectro 1	Spectro 2	Spectro 3
Wavelength (nm)	164-236	380-467	390-520
Gate (ns)	5000	5000	500
Delay (us)	1	1	1
Gain	200	500	1

## Table S1. Features of each of the three spectrometers

- Calculation regarding detectable Li and P amounts ablated with one laser shot
- (i) The precise density of amorphous  $Li_{2.3}PO_{3.65}$  is unknown, but it is expected to be close of those of crystalline  $Li_4P_2O_7$  and  $Li_3PO_4$ , i.e. 2.3 and 2.5 g.cm<sup>-3</sup> respectively. Therefore, the approximate value of 2.4 was used in the calculation. Its molar mass is 2.3 x 6.94 + 30.97 + 3.65 x 16 = 105.33 g.mol<sup>-1</sup>.
- (ii) Steady state deposition rate of  $Li_{2,3}PO_{3,65}$  in the present sputtering conditions is 115-120 nm.h<sup>-1</sup>, or 2 nm.min<sup>-1</sup>
- (iii) Therefore, the amount of Li<sub>2.3</sub>PO<sub>3.65</sub> material deposited during 2 min corresponds to a 4 nmthick layer.
- (iv) The volume of  $Li_{2.3}PO_{3.65}$  material ablated on a spot of 12 µm in diameter (and 7 µm deep) is equal to  $\pi x (12 \cdot 10^{-4})^2 / 4 x (4 \cdot 10^{-7}) = 452 \cdot 10^{-15} \text{ cm}^3$
- (v) This corresponds to  $452 \cdot 10^{-15} \times 2.4 / 105.33 = 10.3 \cdot 10^{-15} \text{ mol}$
- (vi) And to 2.3 x 6.94 x  $10.3 \cdot 10^{-15}$  = 157.10<sup>-15</sup> g or 0.157 pg of Li, and  $30.97 \times 10.3 \cdot 10^{-15}$  = 319.10<sup>-15</sup> g or 0.319 pg of P