



## Novel production method for traceable surface sources by aluminium functionalisation

Dilan Tuzun, V. Lourenço, Lucille Chambon, Yann Kergadallan

### ► To cite this version:

Dilan Tuzun, V. Lourenço, Lucille Chambon, Yann Kergadallan. Novel production method for traceable surface sources by aluminium functionalisation. ICRM 2023 - 23rd International Conference on Radionuclide Metrology and its Applications, Mar 2023, Bucarest, Romania. cea-04272758

HAL Id: cea-04272758

<https://cea.hal.science/cea-04272758v1>

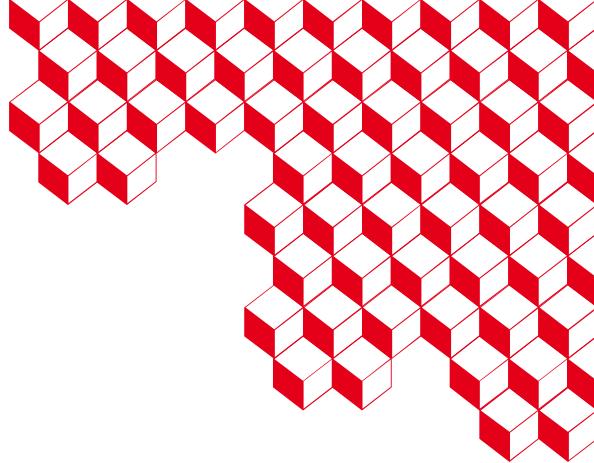
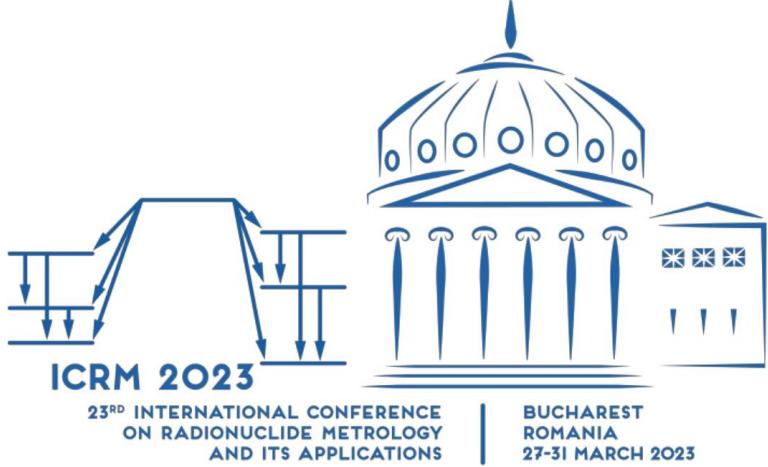
Submitted on 6 Nov 2023

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution 4.0 International License



# Novel production method for traceable surface sources by aluminium functionalisation

Dilan Tüzün, V. Lourenço, L. Chambon, Y. Kergadallan

Université Paris-Saclay, CEA, List, LNE-LNHB, Palaiseau, France



# Aim: Measuring radioactivity on complex surfaces

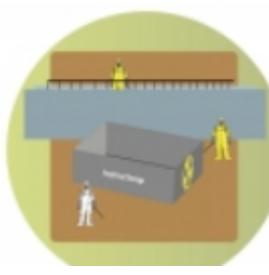
Planning



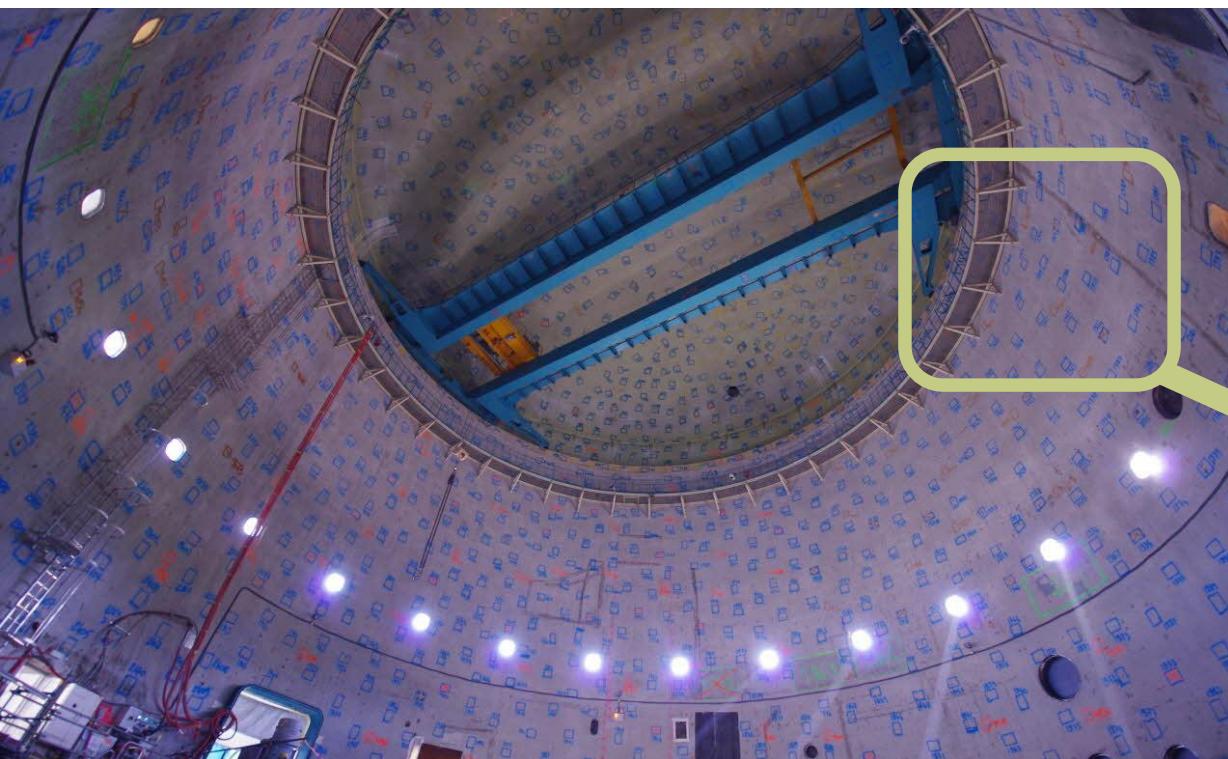
Physical and radiological characterisation



Decontamination

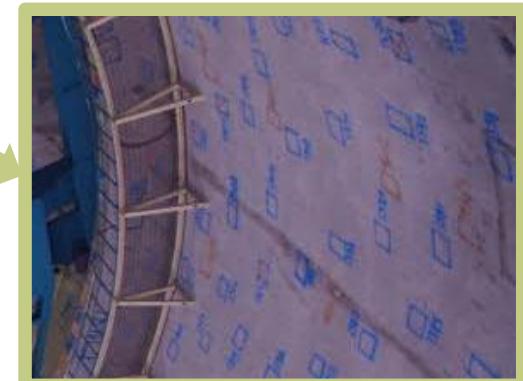
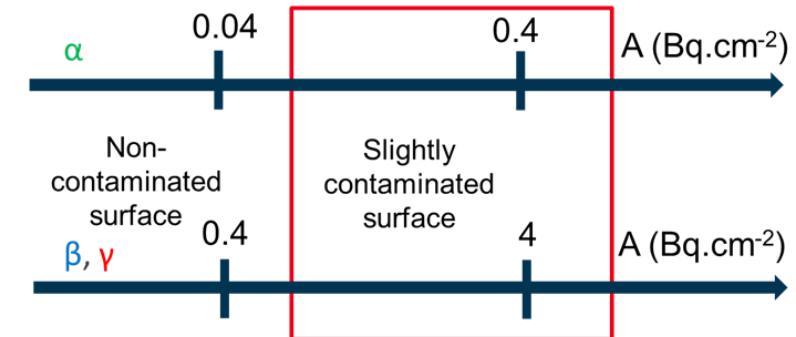


[IAEA, 2016]



**Challenges:**

- Discrimination of non-contaminated surface from low-level contaminated surface
- Very large areas



- Non-flat surfaces

# Aim: Measuring radioactivity on complex surfaces



Complex shapes

## Needs:

- Reliable contamination measurements
- Calibration of measurement systems

## Aim:

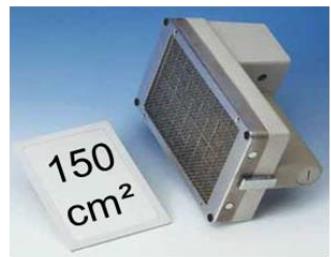
- Traceable activity and emitted flux
- Large surface areas
- Non-contaminating surface (use on-site)
- Complex shapes



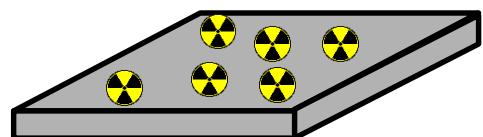
Provide metrologically traceable standard surface sources of various shapes

# Our approach: chemical attachment

Commercial source



Commercial sources :  
Radionuclide dried + protected

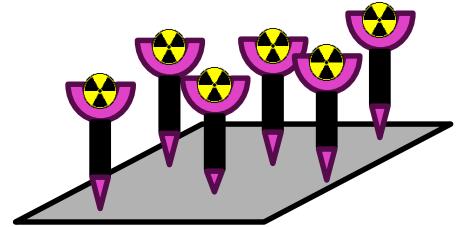


} Al thickness:  
3 mm

Adsorption / Physisorption  $E_{bonding} \sim \text{kJ/mol}$



Chemical bonds radionuclide-surface



} Al thickness: 0.3 mm

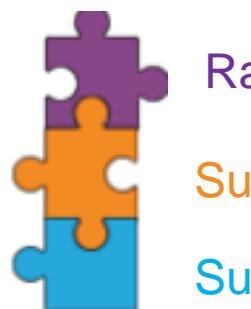
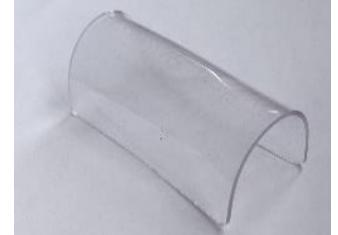
Chemisorption  $E_{bonding} > 100 \text{ kJ/mol}$

## Use of **functionalisation**

- Strong chemical bonds (non-contaminating sources, to check)  
**→ No cover layer**
- Thin sources **→ Minimise self-absorption** of low penetration radiation ( $\alpha$ ,  $\beta$ )

## **Chosen substrates:**

- Aluminium foil
- Polymeric surfaces



Radionuclide

Surface functionalisation

Substrate

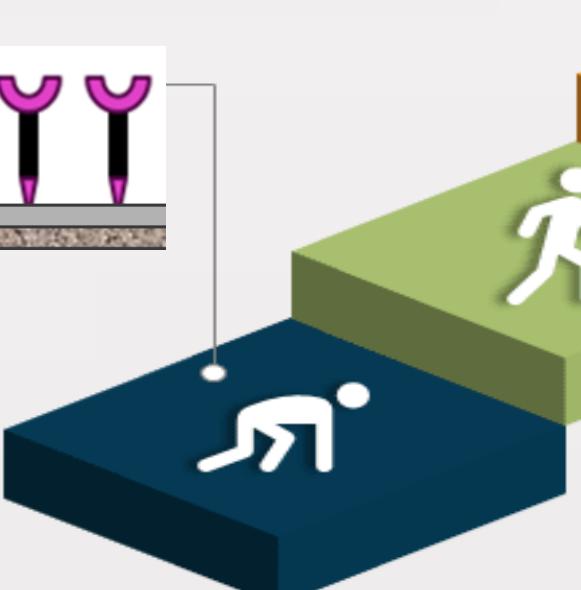
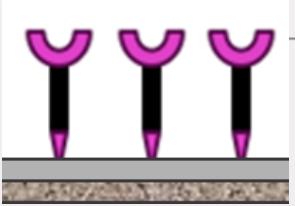
Emitted particles



# Outline

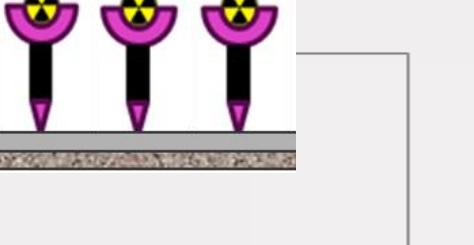
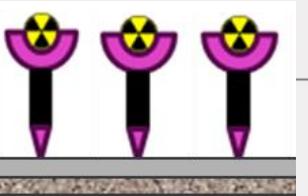
## 1) Functionalisation

- Attach functions to the surface
- 3 main methods



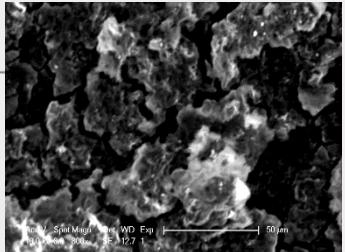
## 3) Radionuclide Binding

- Attach the radionuclide  $^{152}\text{Eu}$  and  $^{241}\text{Am}$



## 4) Radiological Characterization

- Uniformity
- Traceability
- Non-contamination

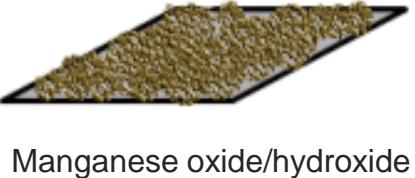


## 2) Chemical characterization

- SEM analyses (topology of the surface)
- Infrared Spectroscopy (Chemical bonds)

Uniform,  
traceable, large  
area surface  
sources ?

# Functions on the surface

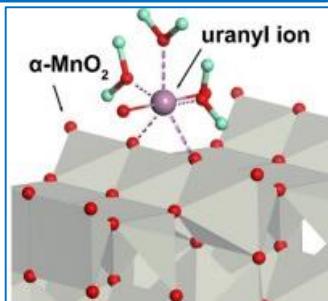


## Manganese oxide layer

$\text{Ra}^{2+}$ ,  $\text{UO}_2^{2+}$

- Reduction of Mn(VII)
- Precipitate as an oxide layer
- Not polymeric
- Brownish colored layer

(Ren et al, 2020)

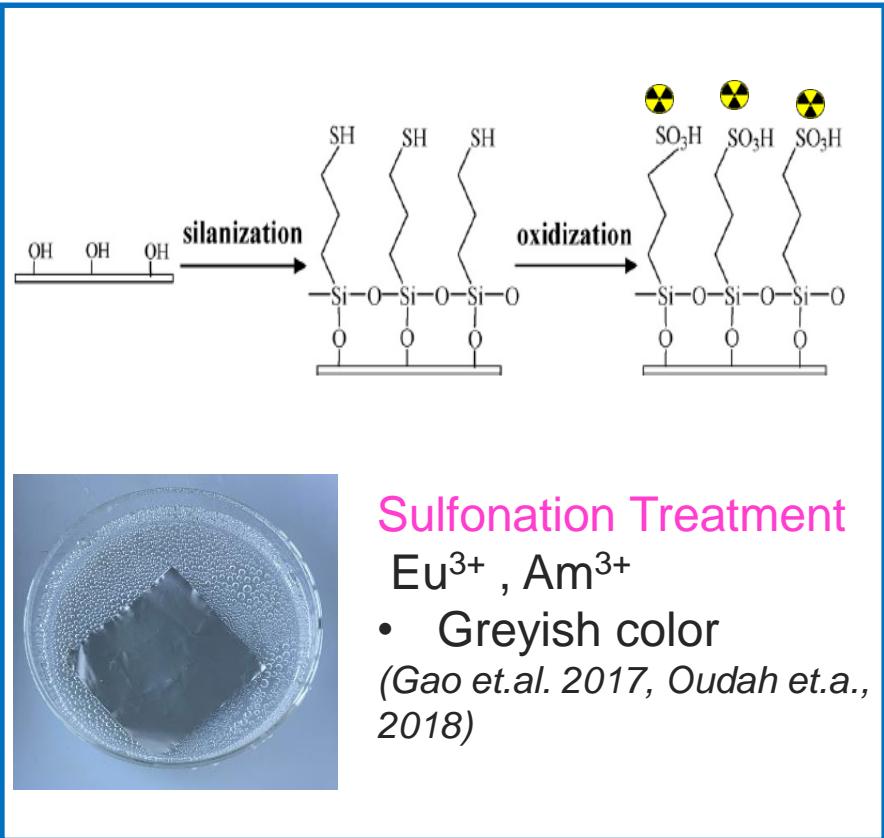


## Chromium oxide layer

$\text{Ra}^{2+}$ ,  $\text{UO}_2^{2+}$

- Reduction of Cr(VI)
- Inorganic polymer
- Soft-gel like mixture formed
- Yellow to dark brown color

(Pommiers et. al. 2014)

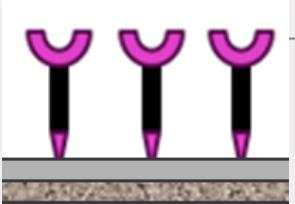


Act as cation exchangers to bind radioactive element

# 1st step

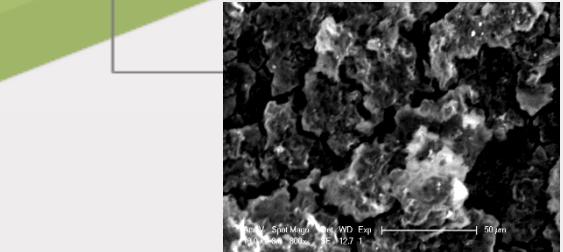
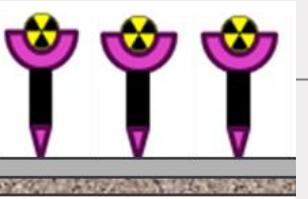
## 1) Functionalisation

- Attach functions to the surface
- 3 main methods

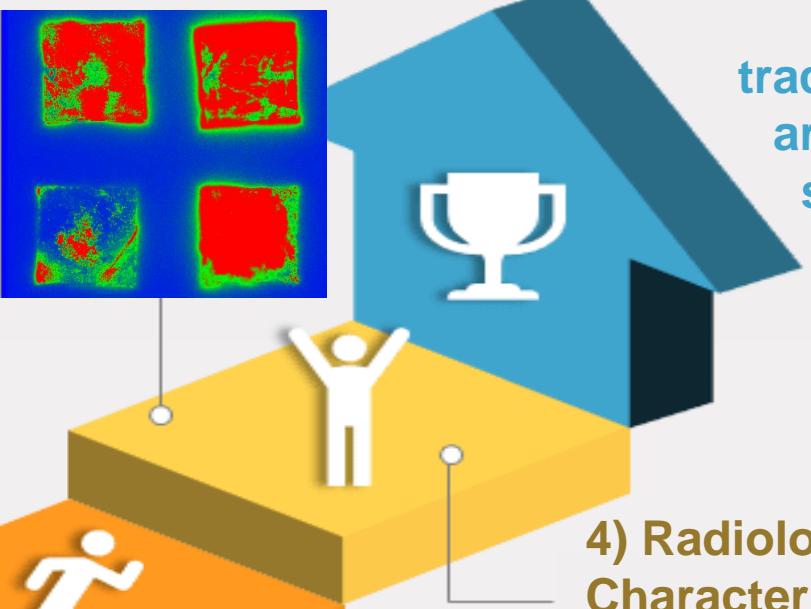
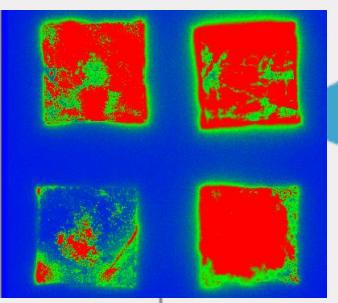


## 3) Radionuclide Binding

- Attach the radionuclide  $^{152}\text{Eu}$  and  $^{241}\text{Am}$



ICRM 2023 Bucharest



Uniform,  
traceable, large  
area surface  
sources ?

## 4) Radiological Characterization

- Uniformity
- Traceability
- Non-contamination

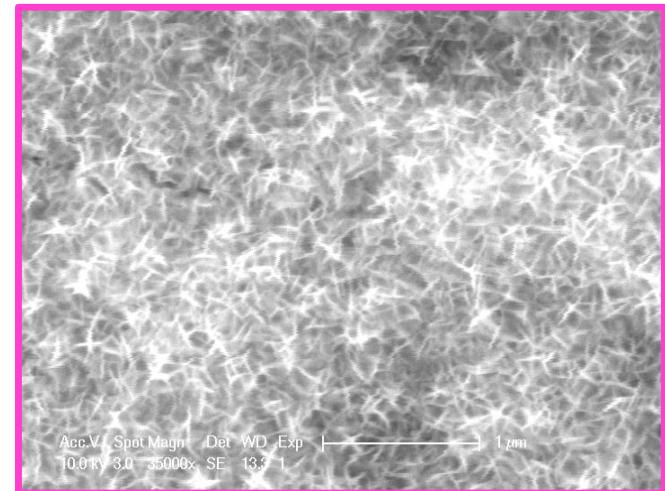
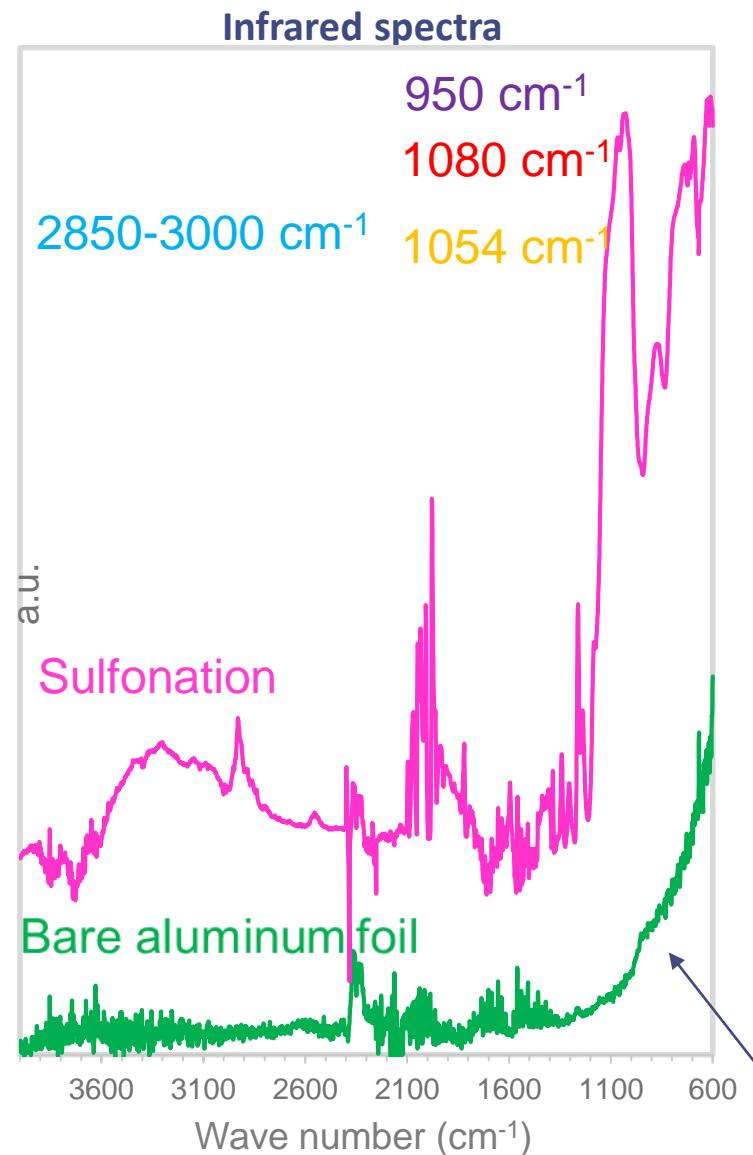
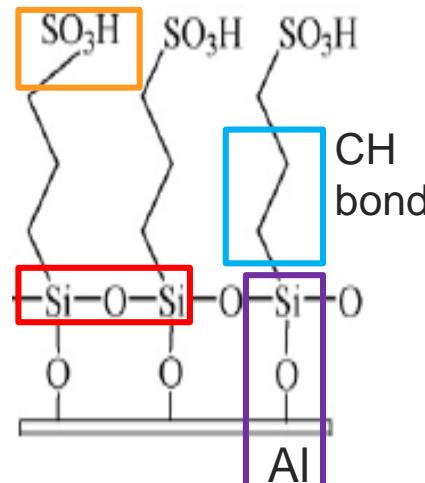
## 2) Chemical characterization

- SEM analyses (topology of the surface)
- Infrared Spectroscopy (Chemical bonds)

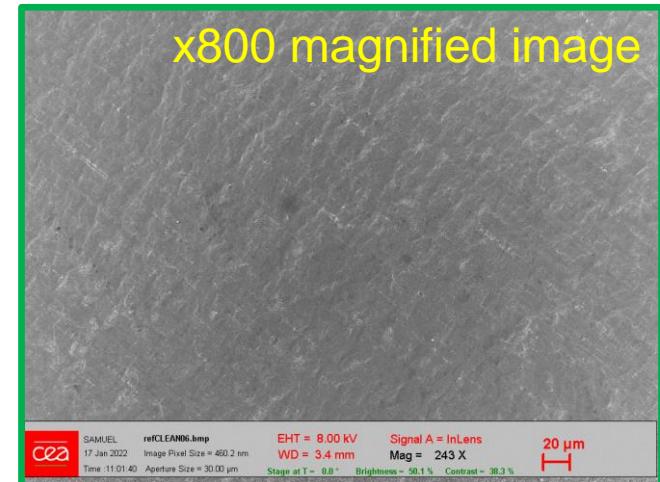


# How effective is functionalisation?

3D animation of vibrations  
in infrared spectroscopy



Magnification: x800 x35000

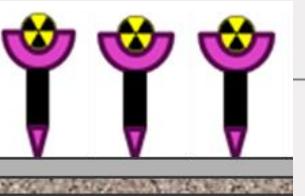
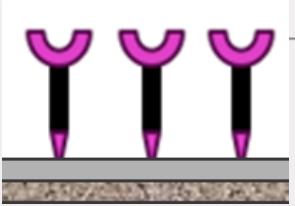


AI-OH bonds

# 2<sup>nd</sup> step

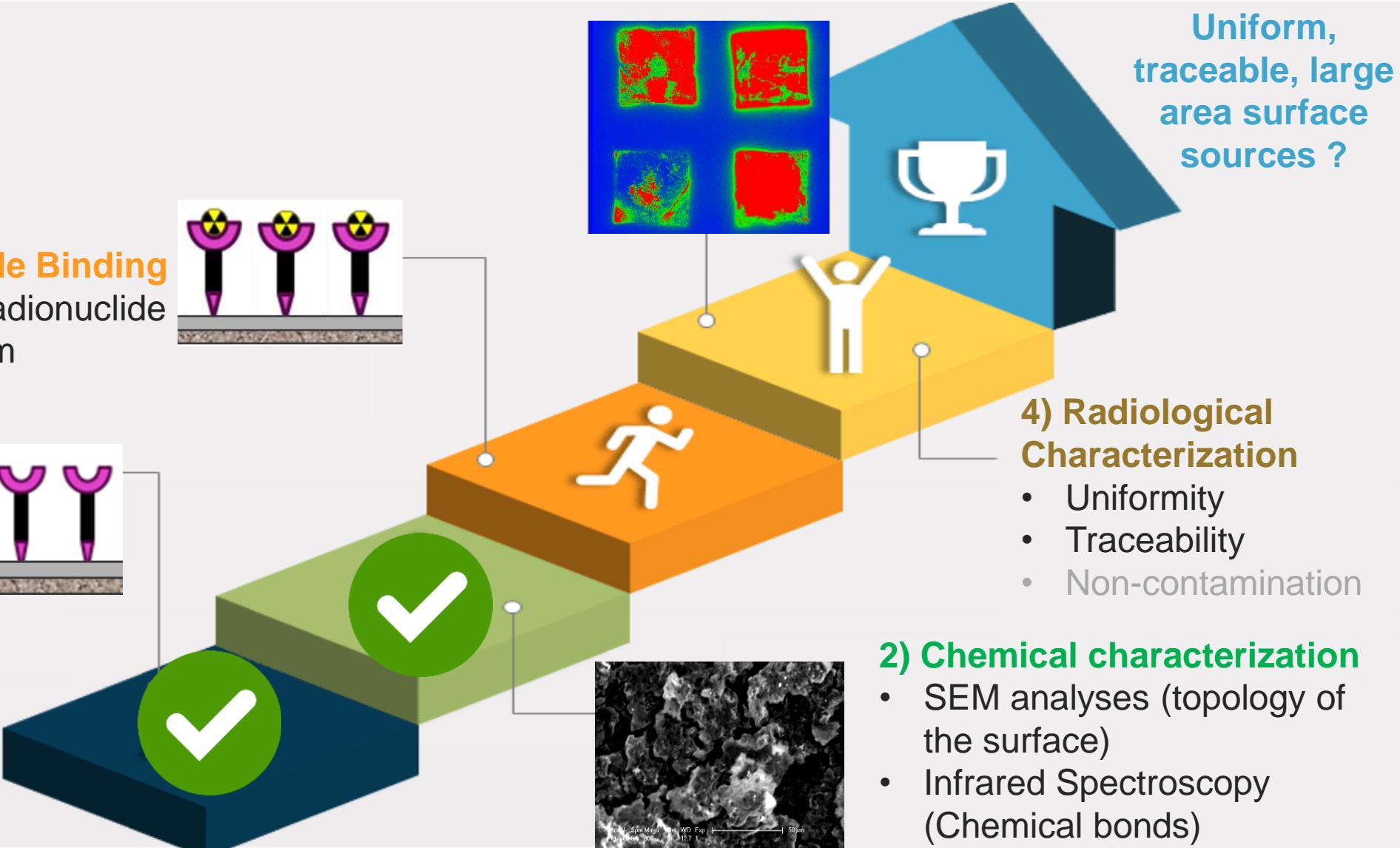
## 1) Functionalisation

- Attach functions to the surface
- 3 main methods



## 3) Radionuclide Binding

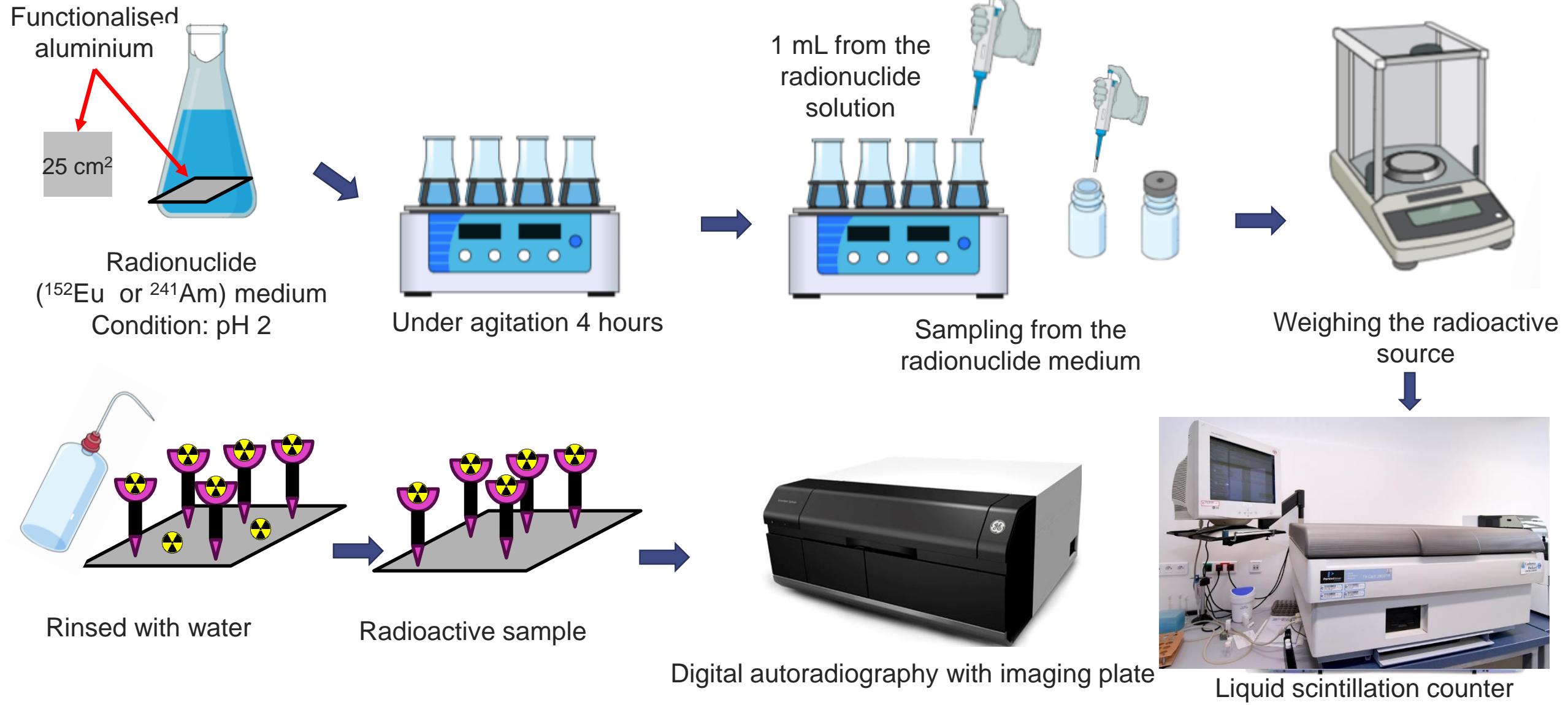
- Attach the radionuclide  $^{152}\text{Eu}$  and  $^{241}\text{Am}$



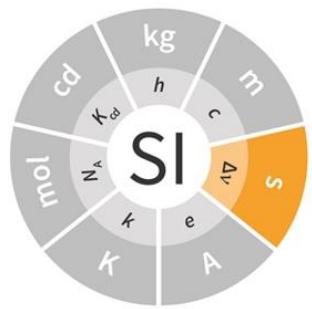
- ## 2) Chemical characterization
- SEM analyses (topology of the surface)
  - Infrared Spectroscopy (Chemical bonds)



# Radionuclide binding experiment



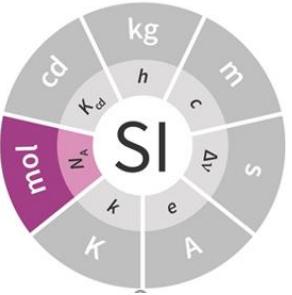
# Experimental conditions



Activity ( $\text{Bq} = \text{s}^{-1}$ )

$$n = \frac{A \cdot T_{1/2}}{N_A \cdot \ln 2}$$

Amount of substance (mol)



$^{241}\text{Am}$

$^{152}\text{Eu}$

Half-life ( $t_{1/2}$ )

432.2 y

13.54 y

Emitter type

$\alpha, \gamma$

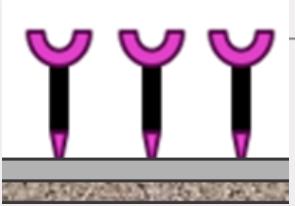
$\beta, \gamma$

Samples	Radionuclide	Stable/Active element ratio	Total amount of element in solution (picomol)
Sulfonation	$^{152}\text{Eu}$	~ 57	~ 0.4
Chromium oxide layer	$^{152}\text{Eu}$	~ 57	~ 0.4
Manganese oxide layer	$^{152}\text{Eu}$	~ 57	~ 0.4
Manganese oxide layer	$^{241}\text{Am}$	No carrier	~ 0.5

# 3<sup>rd</sup> step

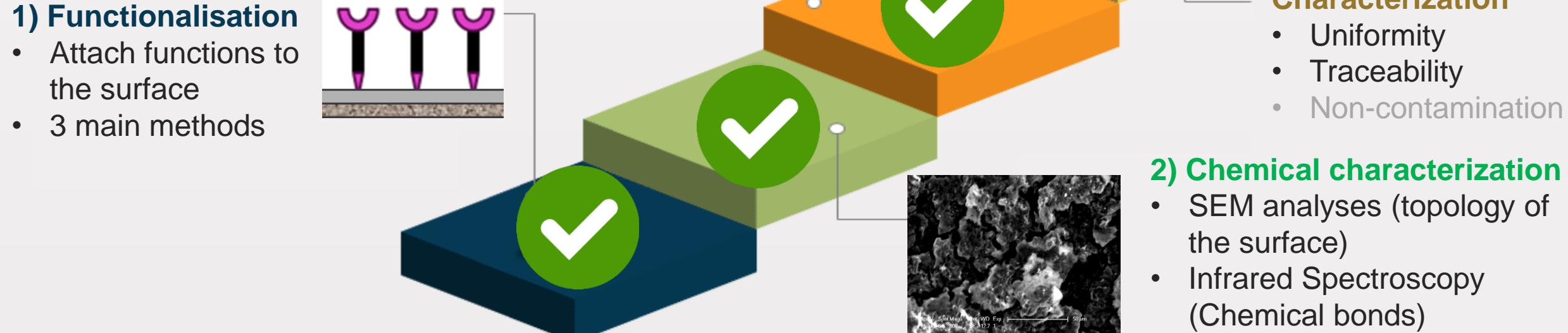
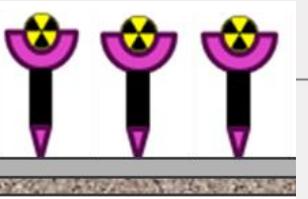
## 1) Functionalisation

- Attach functions to the surface
- 3 main methods

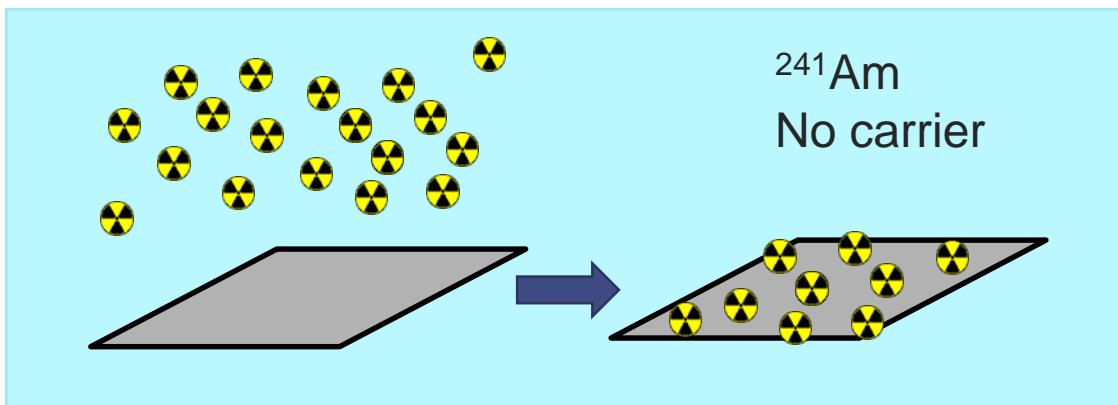
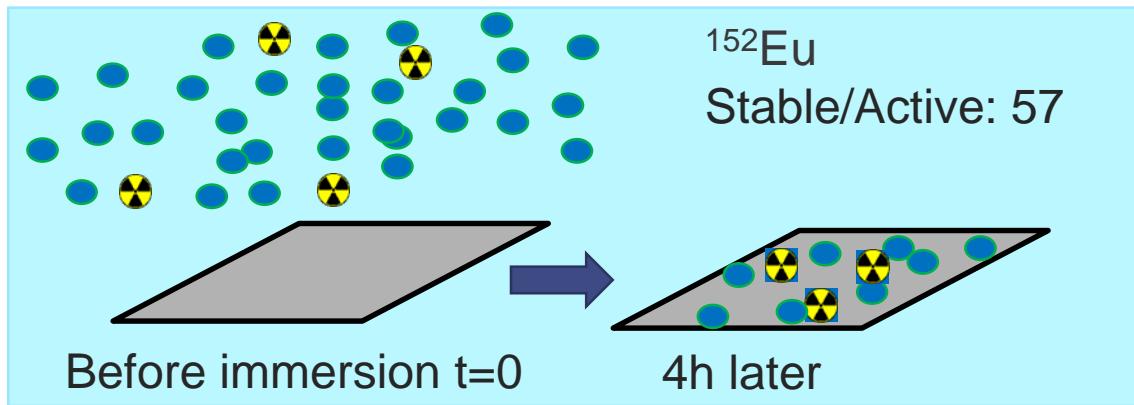


## 3) Radionuclide Binding

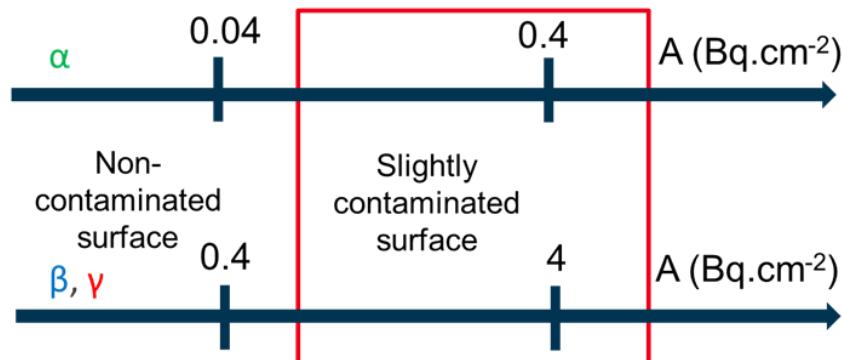
- Attach the radionuclide  $^{152}\text{Eu}$  and  $^{241}\text{Am}$



# Activity of the sources



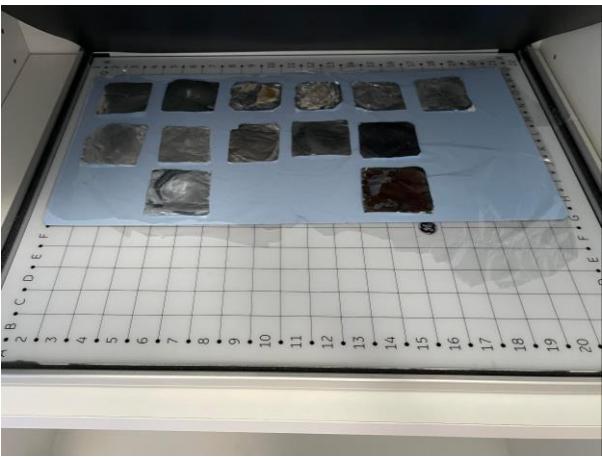
Samples	RN	% LSC activity decrease	Max nber of moles on the surface (picomol)	Max activity bound (Bq.cm <sup>-2</sup> )
Sulfonation	$^{152}\text{Eu}$	(15 ± 1)%	~ 0.07	3.5 ± 0.3
Cr oxide	$^{152}\text{Eu}$	(19 ± 1)%	~ 0.07	5.0 ± 0.5
Mn oxide	$^{152}\text{Eu}$	(8.0 ± 0.5)%	~ 0.03	2.0 ± 0.2
Mn oxide	$^{241}\text{Am}$	(70 ± 5)%	~ 0.4	18.5 ± 2



# Autoradiography imaging

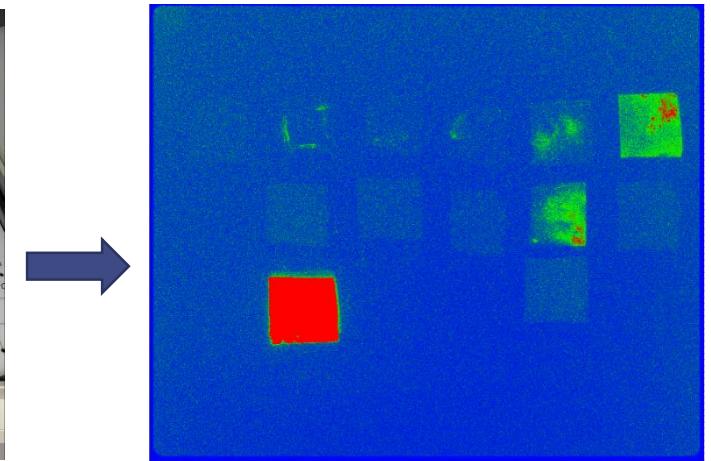


Digital autoradiography  
(Amersham™ Typhoon™)



Sources on the screen

- Close contact with screen for exposure



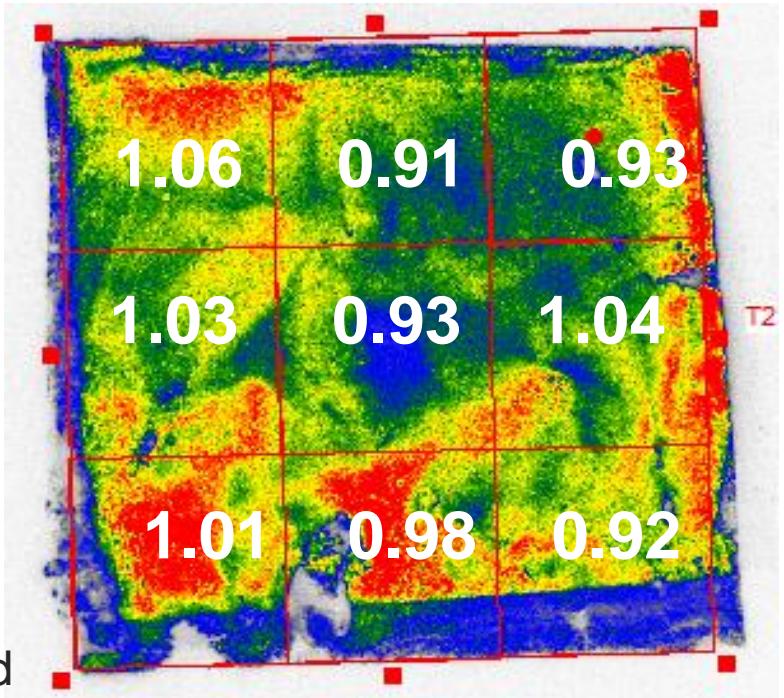
Autoradiography images

# Autoradiography imaging

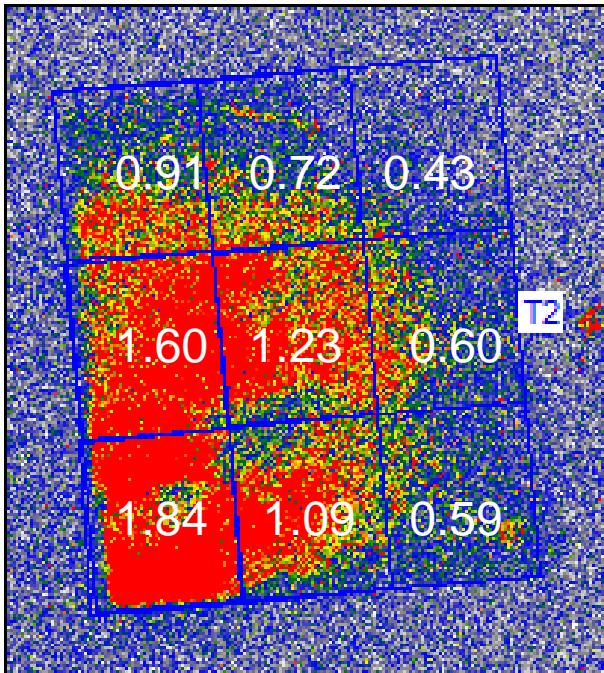
$^{241}\text{Am}$  source  
18 Bq/cm<sup>2</sup>

$^{152}\text{Eu}$  source  
5 Bq/cm<sup>2</sup>

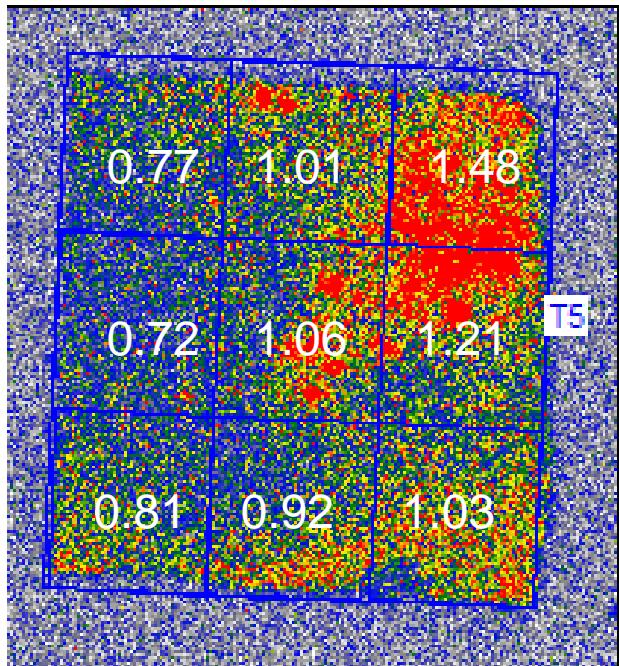
$^{152}\text{Eu}$  source  
2 Bq/cm<sup>2</sup>



Manganese oxide layer  
Uniformity (standard dev.): 95%



Chromium oxide layer  
Uniformity: ~50%

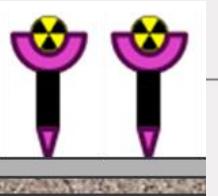
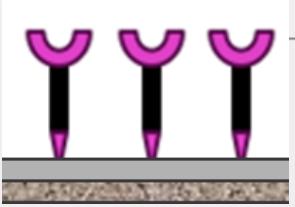


Sulfonation  
Uniformity: ~80%

# Last step

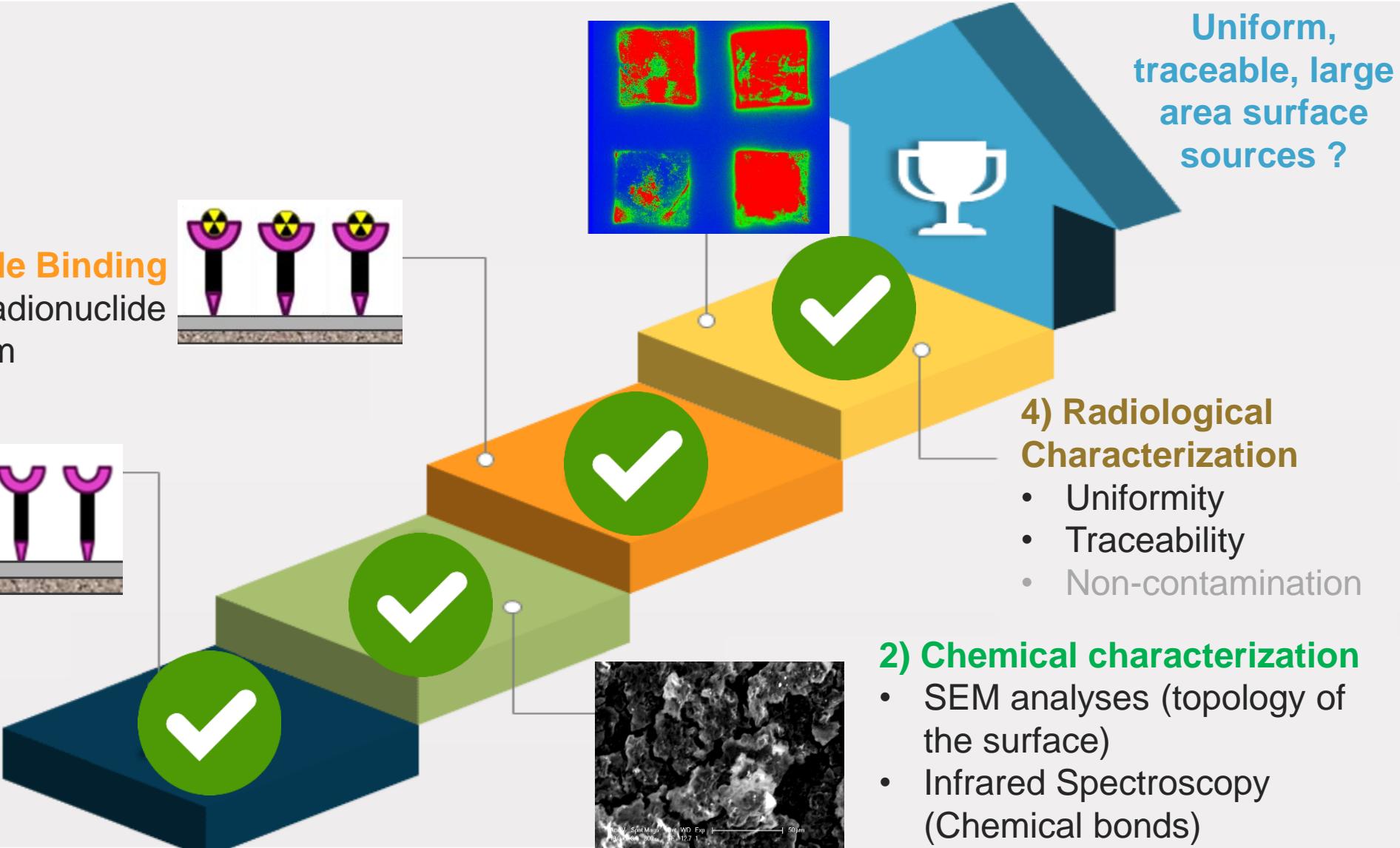
## 1) Functionalisation

- Attach functions to the surface
- 3 main methods



## 3) Radionuclide Binding

- Attach the radionuclide  $^{152}\text{Eu}$  and  $^{241}\text{Am}$



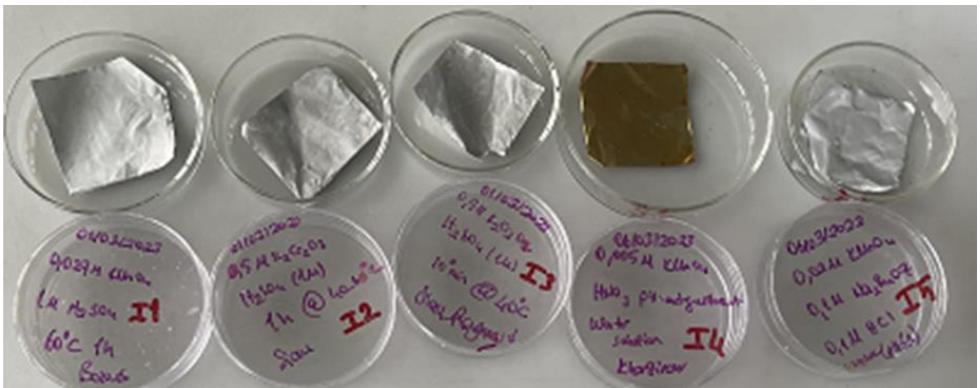
- ## 2) Chemical characterization
- SEM analyses (topology of the surface)
  - Infrared Spectroscopy (Chemical bonds)
- ## 4) Radiological Characterization
- Uniformity
  - Traceability
  - Non-contamination

# Conclusions & future work



## ACHIEVEMENTS

- 3 different source production methods
- Surface activities from 2-20 Bq/cm<sup>2</sup>: **tunable**
- <sup>241</sup>Am and <sup>152</sup>Eu sources
- Some uniform sources (>90%)



## IMPROVEMENTS

- Uniformity: by using 0.5 mm Al foils
- <sup>244</sup>Cm binding ( $\alpha$ )
- Non-contamination tests to be performed

