

Redox Flow batteries at CEA-Liten: Test facilities, some examples of characterization and feedback

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REDOX FLOW BATTERIES AT CEA-LITEN: TEST FACILITIES, SOME EXAMPLES OF CHARACTERIZATION AND FEEDBACK

Annual Scientific Meeting of the GDR RFB, 3 and 4 November 2021 - Paris

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| Nicolas Guillet

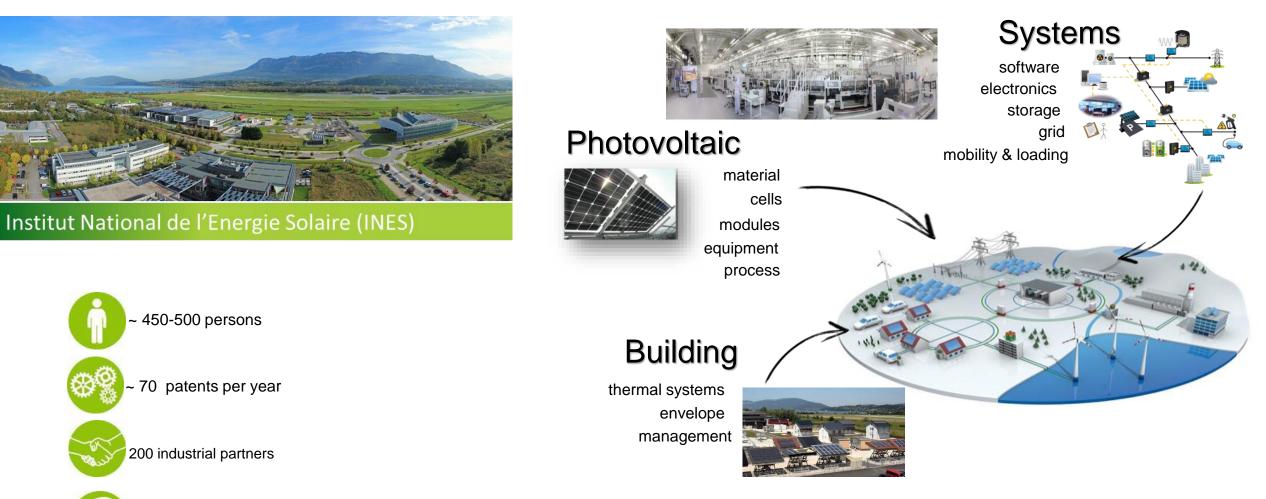
Nicolas GUILLET & Jean-Marie KLEIN Univ. Grenoble Alpes, CEA, Liten, Campus Ines, 73375 Le Bourget du Lac, France

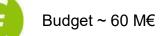


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INES CAMPUS : PRESENTATION & TESTS FACILITIES

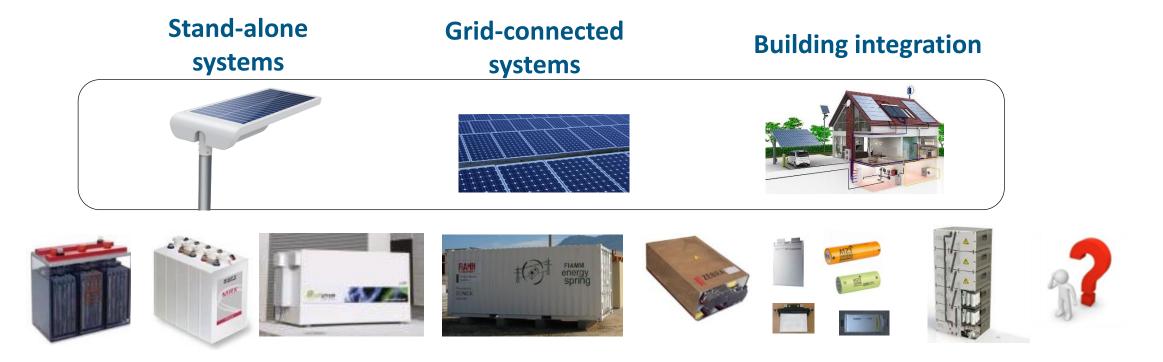








• Three mains domains adressed by the laboratory



- Selection of the most appropriate battery
- Definition of advanced management strategies

TESTS FACILITIES AT INES CAMPUS

• Tests & characterization of batteries



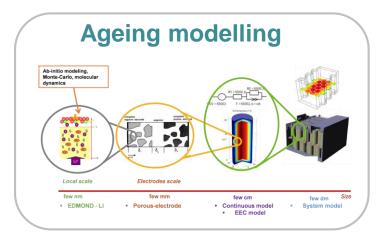
- Small size (up to 20 V, 10 to 800 A) \rightarrow 280 channels
- Module size (up to 150 V and 800 A) → 40 channels
- Pack size (up to 800 V and 600 A 250 kW max) \rightarrow 12 channels

• Battery monitoring & management

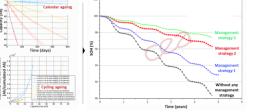


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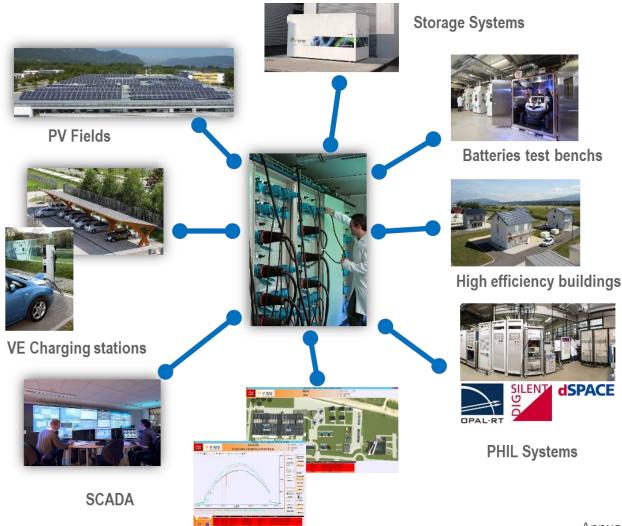






TESTS FACILITIES AT INES CAMPUS

EMS microgrid platform



Objectives

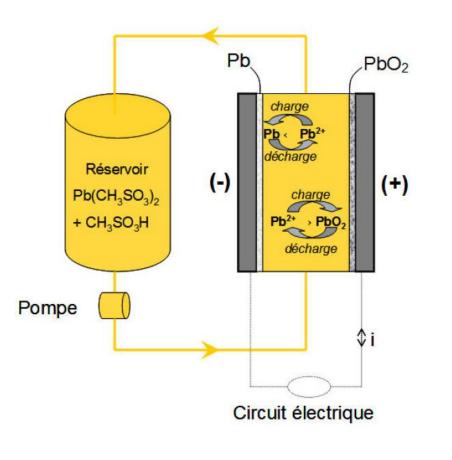
- ✓ Functional tests on equipments
- EMS algorithms development and validation
- ✓ Smart Charging tests/ V2G / V2H
- 170 kVA connection to "Public Grid"
- 6 AC LV Bus bars (400 VAC)
- 45 kVA AC Grid simulation
 - 3 Ph 120V-690V
 - 0-2500 Hz fundamental / Up to 50 kHz Harmonics
 - PHIL Simulations
- BESS characterization and control development
 - 3 test channels for BESS (on/off grid)
 - Up to 250A @400V
 - 600A Bus bar with islanding capabilities
- 110 kWc configurable PV fields
- 120 kWc PV simulation

. . .

100 kW batteries simulation

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• Lead soluble - methanesulfonic acid type redox battery – Pb-MSA (Lab. scale)



A. Oury , A. Kirchev, Y. Bultel	https://tel.archives-ouvertes.fr/tel-00962125/document
(2012 - 2014)	

$Pb^{2+}_{(aq)} + 2 H_2O \leftrightarrow PbO_{2(s)} + 4 H^+ + 2e^-$	(E ₀ =1.46 V vs. SHE)
$Pb^{2+}_{(aq)} + 2e^{-} \leftrightarrow Pb_{(s)}$	(E ₀ =-0.13 V vs. SHE)

 $Pb^{2+}_{(aq)} + 2 H_2O \leftrightarrow PbO_{2(s)} + Pb_{(s)} + 4 H^+$

OCV ~ 1.6 V, solubility of $Pb(CH_3SO_2)_2$ in MSA ~ 2 mol/L

Advantages :

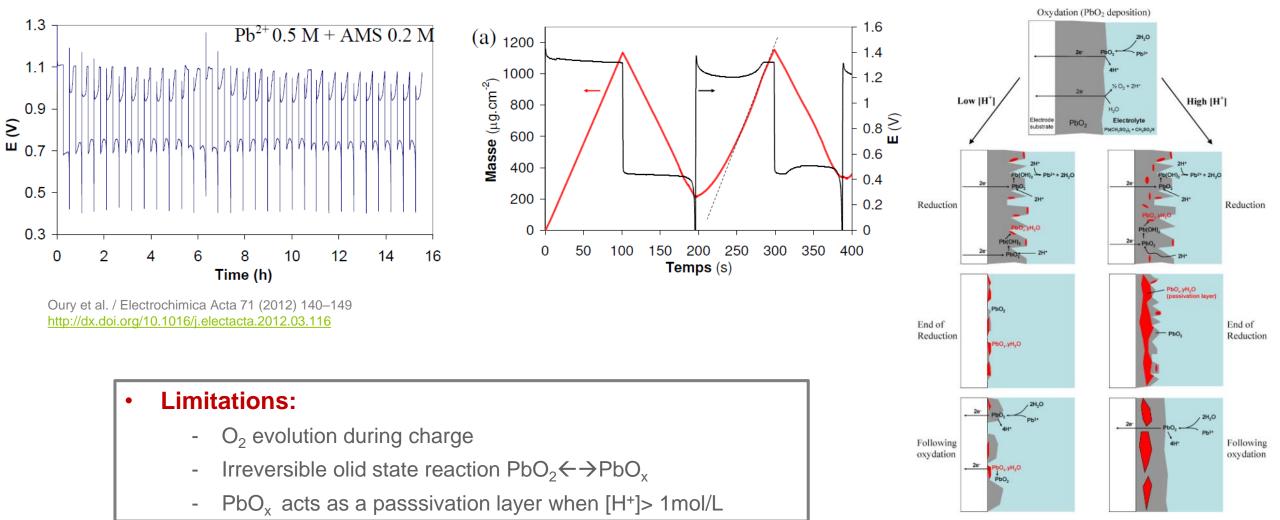
- High theoretical active lead fraction (> 80 %)
- Membraneless (low cost)
- Solid $\leftarrow \rightarrow$ aqueous species reaction : no solid phase evolution

Hurdles :

•

- Low kinetic of PbO₂/Pb²⁺ couple; high overpotentials
- Incomplete reduction of PbO₂; build-up of PbO2 at both electrodes
- Poor PbO₂ adherence ; loss of PbO₂particles in the electrolyte

• Lead soluble - methanesulfonic acid type redox battery – Pb-MSA (Lab. scale)



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Positive :

Negative :

Zinc – Bromine (Redflow)



Power / energy capacity: 5 kW (7.5 kW peak) / 11 kWh

Voltage application : 53 Vdc

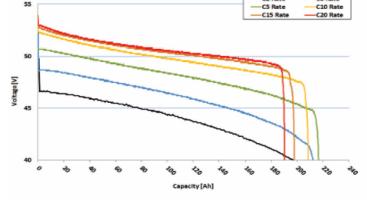
Net energy efficiency : 80% DC-DC Max

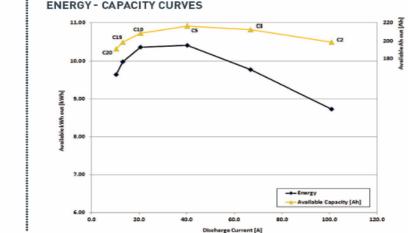
Dimensions / weight

845 L x 823 H x 400 W (mm) 90 kg without electrolyte 240 kg with electrolyte (100L)

Internal operating temperature : 10–50 °C

 $Br_2 + 2e^- \leftrightarrow 2Br^ Q(Br^-)_{ag} + nBr_2 \leftrightarrow Q(Br_2)_nBr^-_{ag}$ $Zn \leftarrow \rightarrow Zn^{2+} + 2e^{-}$ $Br_2 + Zn \rightarrow 2 Br + Zn^{2+}$ DISCHARGE - CAPACITY CURVES ENERGY - CAPACITY CURVES 11.00 -C2 Rate -C3 Rate -C5 Rate C10 Rate C15 Rate





Advantages : ٠

- Good specific energy density and round trip efficiecy
- 100 % depth discharge available

Hurdles : ٠

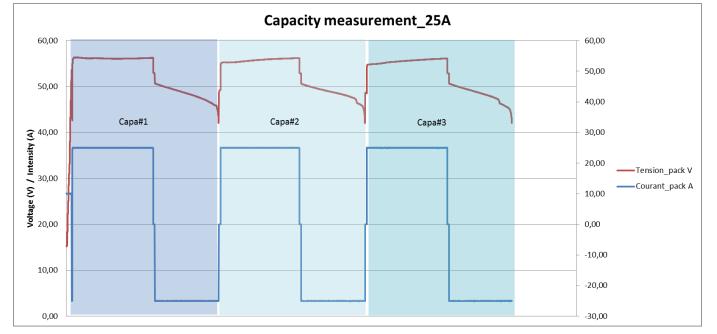
- Self discharge
- Zinc dendrites formation

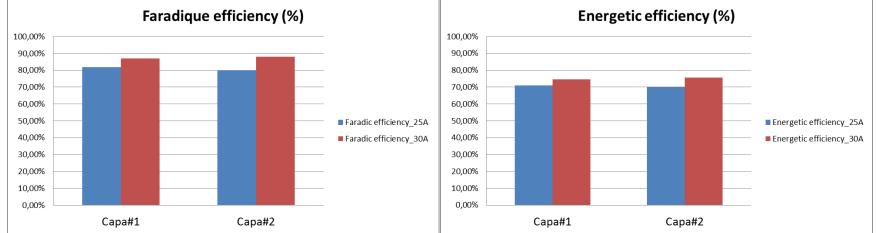
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Zinc – Bromine (Redflow)







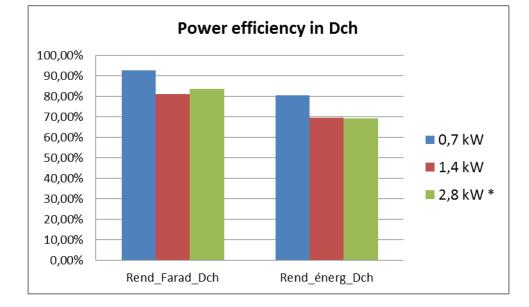


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Zinc – Bromine (Redflow)



• Efficiency:



Self discharge : SoC decreases linearly with time > 1 %/h at 25 °C

Power / energy capacity: 5 kW (7.5 kW peak) / 11 kWh

• Limitations: works very well but

- Balance of système is a major cost element
- System design and energy management system had to be improved
- Periodic self maintenance (minimum once every 72 hours of zinc pump operation)

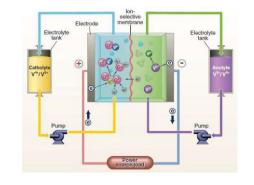
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• All Vanadium redox flow battery (VRB)



Positive : Negative : VO_2^+ + 2 H⁺ + e⁻ ← → VO^{2+} + H₂O V^{2+} ← → V^{3+} + e⁻

 $\mathsf{VO}_2^+ + 2 \mathsf{H}^+ + \mathsf{V}^{2+} \longleftrightarrow \mathsf{VO}^{2+} + \mathsf{H}_2\mathsf{O} + \mathsf{V}^{3+}$





Noak et al., 2020

• Advantages :

- Simple system
- 100 % depth discharge available
- Hurdles :
 - Self-discharge
 - Low specific density
 - Cost of vanadium

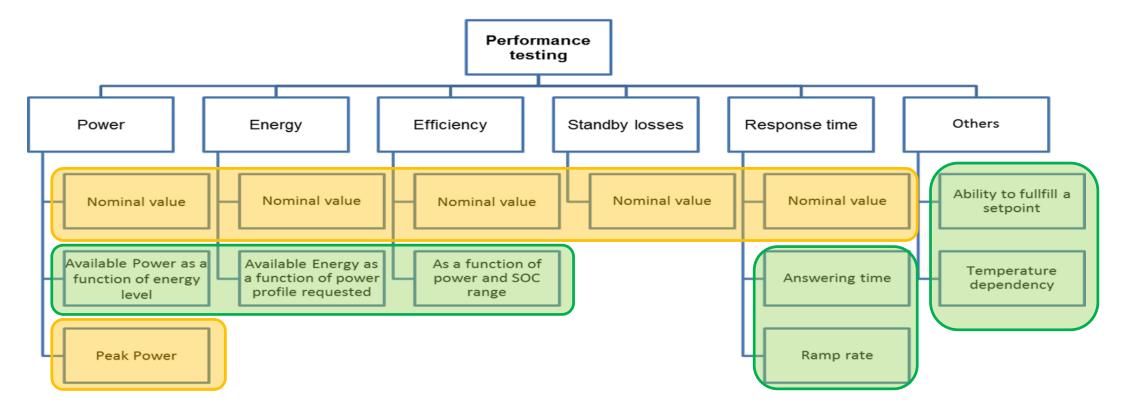
• All Vanadium redox flow battery (VRB)

	Parameter	Data from supplier
Performance	P charge nominale 15 - 80% SOC	10 kW DC 15 kVA onduleur
	Nominal energy	100 kWh
Voltage output	AC voltage	400 V _{AC}
	DC voltage range	36V _{DC} - 63V _{DC}
Efficiency	Round trio DC efficiency	70-80%
Self-discharge	Within Cells	0,07 kW
	Within tankks	Not significant
Weight & size	Size (mm)	4500*2200*2403
	Empty mass	3500 kg
	Mass with electrolyte	10300 kg



- Cell: 1,2 V / 50 W
- Module (20 cells 20S) : 24 V / 1 kW
- Battery (10 modules 2S5P) : 48 V / 10 kW

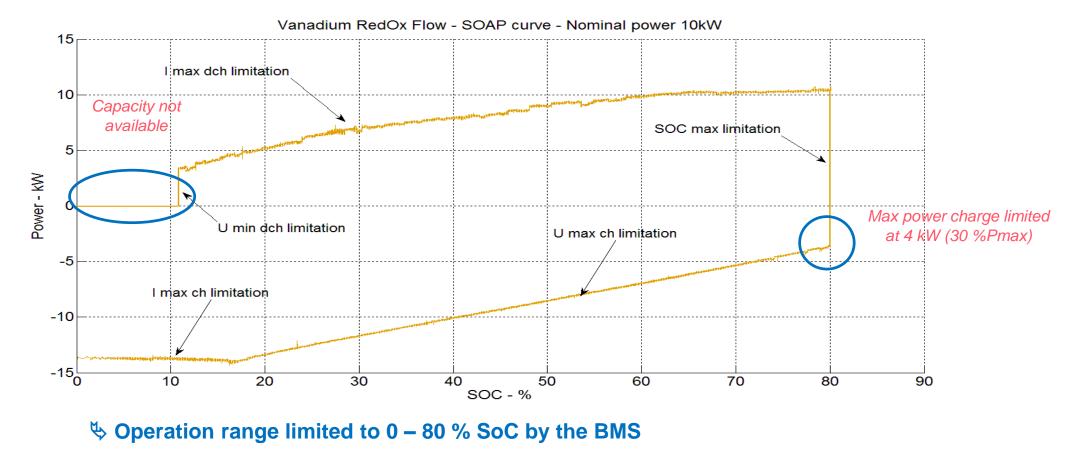
- All Vanadium redox flow battery (VRB)
 - Tests according to the DERri protocol <u>http://www.der-lab.net/downloads/noe_003_grid_connected_storage.pdf</u>



Data often given by the supplier but have to be confirmed on the whole operation range

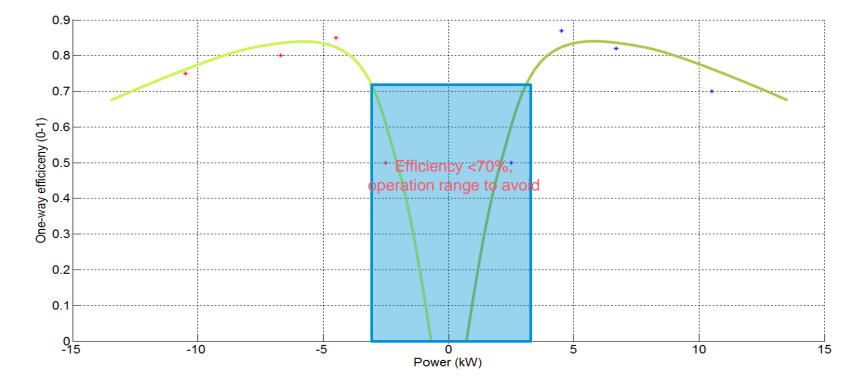
Not or rarely available ; test to be conducted

- All Vanadium redox flow battery (VRB)
 - Example of DERri test : State of Available Power (SoAP)



Solution with SoC (Umax)

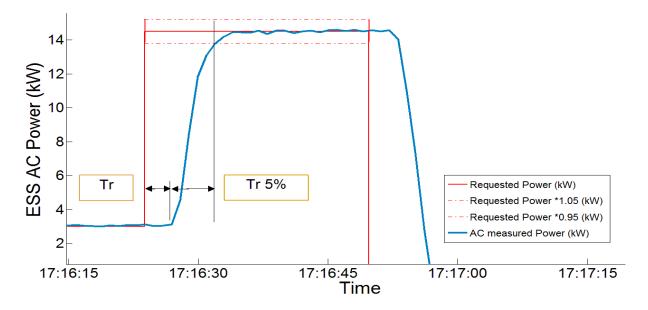
- All Vanadium redox flow battery (VRB)
 - Example of DERri test : one-way AC efficiency



b Low AC efficiency at low rcharge and discharge rates (auxiliaries)

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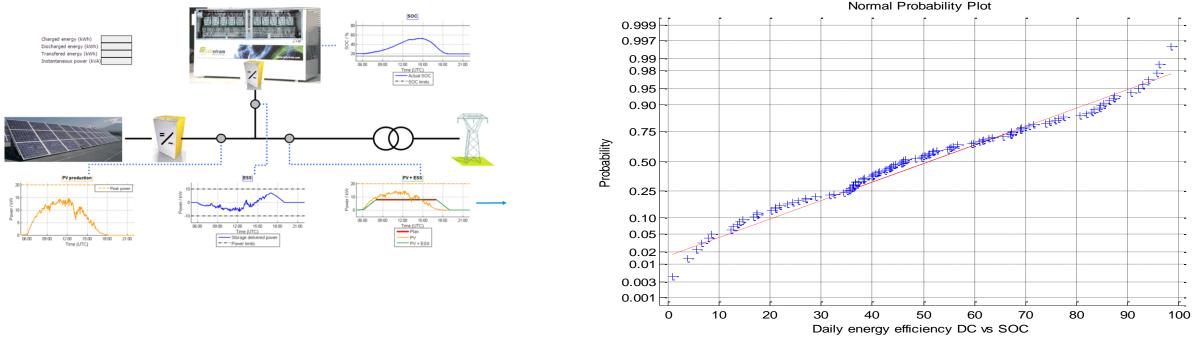
- All Vanadium redox flow battery (VRB)
 - Example of DERri test : time response



From / to	10 kW charge	10 kW discharge
10 kW charge	-	10s
2kW charge	5s	7s
STOP	8s	8s
2 kW discharge	6s	5s
10 kW discharge	9s	-

b Time constants of few seconds

- All Vanadium redox flow battery (VRB)
 - Onsite experiment (Projet AO CRE 2012 PV ZNI) : 500 days of operation



- No evolution of performance during 500 days of operation
- No evolution of the DC efficiency but slight increase of the internal impedance
- 50% of operation points are at an efficiency lower than 50% → operation at low power

All Vanadium redox flow battery (VRB)

Conclusions over 3 years of evaluation

- Strong deviation between standard datasheet and the real life
- Power limitations are mostly related to electrochemical phenomena
- Efficiency is lower than expected and is highly depending on the power request
- OPEX higher than expected (maintenance, electrolyte leak, pump failure...)
- Stable performance over 3 years while there was no temperature regulation of the battery

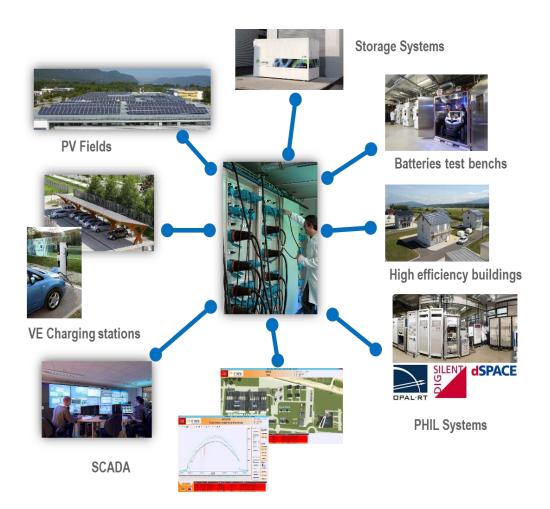








- Stationary application storage and power grid management
- A complete ecosystem and experts in the field of electrochemical storage
- Characterization and evaluation of redox flow batteries of different levels of maturity





Thank you for your attention !

Questions ?

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