

Physicochemical properties evolution of cathodic catalyst layer and cell performance improvement during PEMFC start-up with mitigation strategies

Timothée Drugeot, Fabrice Micoud, Eric Pinton, Jean-Philippe Poirot, Sébastien Rosini, Laure Guetaz, Lucas Poupin, Yann Bultel

▶ To cite this version:

Timothée Drugeot, Fabrice Micoud, Eric Pinton, Jean-Philippe Poirot, Sébastien Rosini, et al.. Physicochemical properties evolution of cathodic catalyst layer and cell performance improvement during PEMFC start-up with mitigation strategies. 32nd ISE Topical Meeting, Jun 2022, Stockholm, Sweden. cea-03735813

HAL Id: cea-03735813 https://cea.hal.science/cea-03735813v1

Submitted on 21 Jul2022

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.





International Society of Electrochemistry

FROM RESEARCH TO INDUSTRY

Physicochemical properties of cathodic catalyst layer and cell performance evolution during PEMFC start-up with mitigation strategies – 32nd ISE Topical Meeting

<u>Timothee Drugeot^{1,2}</u>, Fabrice Micoud¹, Eric Pinton¹, Jean-Phillippe Poirot-Crouvezier¹, Sébastien Rosini¹, Laure Guetaz¹, Lucas Poupin¹ and Yann Bultel²

¹Univ. Grenoble Alpes, CEA, LITEN, DEHT 38000 Grenoble, France

²Univ. Grenoble Alpes, Univ. Savoie Mont Blanc, CNRS, Grenoble INP, LEPMI, 38000 Grenoble, France

Email: timothee.drugeot@cea.fr



Shut-Down (SD) / Start-Up (SU) process





Oxygen consumption^[1]



Oyarce et al. "Comparing shut-down strategies for proton exchange membrane fuel cells" (2014)
Reiser et al. "A Reverse-Current Decay Mechanism for Fuel Cells" (2005)
Pei et al. « A quick evaluating method for automotive fuel cell lifetime"

cea

CONTEXT

Shut-Down (SD) / Start-Up (SU) process



Oyarce et al. "Comparing shut-down strategies for proton exchange membrane fuel cells" (2014)
Reiser et al. "A Reverse-Current Decay Mechanism for Fuel Cells" (2005)
Pei et al. « A quick evaluating method for automotive fuel cell lifetime"

cea

CONTEXT

Shut-Down (SD) / Start-Up (SU) process



[3] Pei et al. « A quick evaluating method for automotive fuel cell lifetime"

Reverse Current Decay mechanism



cea

CONTEXT

Reverse Current Decay mechanism



Reverse Current Decay mechanism



Reverse Current Decay mechanism



Reverse Current Decay mechanism



Reiser et al. "A Reverse-Current Decay Mechanism for Fuel Cells" (2005)
Radrianarizafy et al. "Modelling Carbon Corrosion during a PEMFC Startup: Simulation of Mitigation Strategies" (2020)

athode / V vs KHE

Reverse Current Decay mechanism



•

•

In-situ SU-AST protocol

Emulation of realistic cathode potential profile



• CO₂ emission rate along AST

In-situ SU-AST protocol

Emulation of realistic cathode potential profile



In-situ SU-AST protocol

Parametric study



In-situ SU-AST protocol

Parametric study



In-situ SU-AST results

E_{front} investigation – Cell voltage evolution





• Harsh voltage loss along overall polarization curve

•

In-situ SU-AST results

E_{front} investigation – Cell voltage evolution



In-situ SU-AST results

E_{front} investigation – CO₂ measurements



<u>1.0 – 1.2 V range</u>

- Unexpected improvement of cell performance
- Low CO₂ emission rate expected during the first 20 cycles

<u>1.4 – 1.6 V range</u>

- Harsh voltage loss along overall polarization curve
- **Higher CO₂ emission** rate for 1.6 V case along cycles

•

•

In-situ SU-AST results

E_{front} investigation – **CO**₂ measurments



ightarrow CSC of only poorly ordered domains

<u>1.4 – 1.6 V range</u>

- Harsh voltage loss along overall polarization curve
- Higher CO₂ emission rate for 1.6 V case along cycles

\rightarrow CSC of poorly and more ordered domains

.

•

In-situ SU-AST results

E_{front} investigation – CV evolution

In-situ SU-AST results

E_{front} investigation – Conclusion

The French Alternative Energies and Atomic Energy Commission

Timothée DRUGEOT

Post-mortem Protocol

The French Alternative Energies and Atomic Energy Commission

Cutting sample of fresh / aged MEA

Post-mortem Protocol

Post-mortem Protocol

Post-mortem Protocol

SEM observation

• No CCL structural modification from fresh to aged MEA at 1.2 V

The French Alternative Energies and Atomic Energy Commission

Timothée DRUGEOT

SEM observation

• No CCL structural modification from fresh to aged MEA at 1.2 V

Low compaction of CCL and densification of Pt nanoparticles at 1.4 V and harsher at 1.6 V

→ Linear correlation with CSC mechanisms

The French Alternative Energies and Atomic Energy Commission

Timothée DRUGEOT

Local polarization curve

Fresh - 1.0V - 1.2V - 1.4V - 1.5V - 1.6V

- At 50 %HR : No performance decay excepted for 1.6 V case
- At 100 %HR : Emergence of progressive performance decay

 $\mathbf{1}$

Local polarization curve

Fresh - 1.0V - 1.2V - 1.4V - 1.5V - 1.6V

- At 50 %HR : No performance decay excepted for 1.6 V case
- At 100 %HR : Emergence of progressive performance decay
- Improvement of performance for 1.0 V case similar as in previous in-situ SU-AST (98 cm²)

 \rightarrow Mass transport limitation is more pronounced At higher RH

- → Water management is altered due to CCL chemical and/or electrochemical modifications:
 - i. Loss of hydrophobicity
 - ii. Modification of carbon surface properties
 - iii. Higher Pt/C loading ratio due to carbon support corrosion

Local polarization curve

Fresh - 1.0V - 1.2V - 1.4V - 1.5V - 1.6V

PEIS (H₂/Air) @0.85 V

PEIS (H₂/Air) @0.85 V

PEIS (H₂/Air) @0.85 V

PEIS (H₂/Air) @0.85 V vs ECSA

PEIS (H₂/Air) @0.85 V

The French Alternative Energies and Atomic Energy Commission

Timothée DRUGEOT

- > Validation of complete *in-situ* & *post-mortem* investigation by:
 - > Electrochemical and physico-chemical analysis on large (98 cm²) and localized small (1.8 cm²) MEA areas
- > **Mitigating parameters confirmed and quantified** with represented potential profile in real flow field design:
 - **E**_{front}: Lowering cathode potential by by-passing electrical current through dummy load
 - **t**_{front}: Lowering exposure time by accelerating H₂ injection in Air-filled anode
 - **RH**_{a/c}: Flushing cell with dry gas before H₂ injection
 - T_{cell}: Lowering stack temperature before H₂ injection

Complex degradation mechanism revealed:

- Weaker water management for damaged MEA in our SU-AST beyond 1.4 V
- Better cell performance for damaged MEA at 1.0 V
- > Further insights on degradation mechanisms by using:
 - Water sorption measurements
 - Raman spectroscopy \rightarrow Investigation of amorphous and graphitized domains into CCL
 - **XPS technique** →Investigation of functional groups onto carbon surface

Thank for your attention

Timothée DRUGEOT

Overview of post-mortem investigation

ECSA loss VS CO₂ emission

Lowering E_{front} / t_{front} / HR / T_{cell} → Reduce CSC mechanism and Pt degradations → Reduce CCL thinning and thus loss of porosity → Require additional post-mortem analysis to validate water retention and degradation mechanisms

The French Alternative Energies and Atomic Energy Commission

Timothée DRUGEOT

Polarization & CV

Post-mortem RESULTS

PEIS (H₂/Air) at high RH

cea

Post-mortem RESULTS

PEIS (H₂/N₂) at low and high RH : « Fresh » vs « Aged » MEA

