



Formic acid and CO as key ‘power molecules’ in the catalytic conversion of CO₂ to chemicals

Thibault Cantat

► To cite this version:

Thibault Cantat. Formic acid and CO as key ‘power molecules’ in the catalytic conversion of CO₂ to chemicals. The Netherlands’ Catalysis and Chemistry Conference (N3C), May 2022, Noordwijkerhout, Netherlands. cea-03664843

HAL Id: cea-03664843

<https://cea.hal.science/cea-03664843>

Submitted on 11 May 2022

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L’archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d’enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Formic acid and CO as key ‘power molecules’ in the catalytic conversion of CO₂ to chemicals

T. Cantat, NIMBE, CEA, CNRS, Université Paris-Saclay, Université Paris-Saclay

With 95 % of organic chemical commodities deriving from fossil resources, the chemical industry is currently exploring novel and renewable carbon feedstocks for the production of both bulk and fine chemicals.¹ In this context, the utilization of CO₂ or products derived from biomass wastes is an attractive strategy to access value-added products. Because these carbon sources feature carbon atoms in an oxidized state, the development of reduction methods is needed and they call for the design of efficient catalysts able to break strong C–O and C=O bonds.

Over the last years, our group has developed novel catalytic reactions for the conversion of CO₂ to formamides, N-heterocycles, methylamines and methanol, using hydroboranes, hydrosilanes or formic acid as reductants.²⁻¹⁶ Extension of this methodology to SO₂ enabled the facile conversion of this gas to sulfones, under metal-free conditions.¹⁷ These new catalytic transformations rely on the use of simple organocatalysts, including nitrogen and phosphorus bases as well as Frustrated Lewis Pairs. The mechanisms at play in these transformations will be presented, based on DFT calculations and isolation of reactive catalytic intermediates.



- [1] C. Chauvier, T. Cantat. (**2017**). *ACS Cat.*, **7**, 2107.
- [2] C. D. Gomes, O. Jacquet, C. Villiers, P. Thuery, M. Ephritikhine, T. Cantat. (**2012**). *Angew. Chem. Int. Ed.*, **51**, 187.
- [3] O. Jacquet, X. Frogneux, C. D. Gomes, T. Cantat. (**2013**). *Chem. Sci.*, **4**, 2127.
- [4] O. Jacquet, C. D. Gomes, M. Ephritikhine, T. Cantat. (**2013**). *Chemcatchem*, **5**, 117.
- [5] X. Frogneux, O. Jacquet, T. Cantat. (**2014**). *Cat. Sci. Technol.*, **4**, 1529.
- [6] C. D. Gomes, E. Blondiaux, P. Thuery, T. Cantat. (**2014**). *Chem. Eur. J.*, **20**, 7098.
- [7] A. Tlili, X. Frogneux, E. Blondiaux, T. Cantat. (**2014**). *Angew. Chem. Int. Ed.*, **53**, 2543.
- [8] C. Chauvier, A. Tlili, C. D. Gomes, P. Thuery, T. Cantat. (**2015**). *Chem. Sci.*, **6**, 2938.
- [9] X. Frogneux, E. Blondiaux, P. Thuery, T. Cantat. (**2015**). *ACS Cat.*, **5**, 3983.
- [10] A. Tlili, E. Blondiaux, X. Frogneux, T. Cantat. (**2015**). *Green Chem.*, **17**, 157.
- [11] A. Aloisi, J. C. Berthet, C. Genre, P. Thuery, T. Cantat. (**2016**). *Dalton Trans.*, **45**, 14774.
- [12] X. Frogneux, N. von Wolff, P. Thuery, G. Lefevre, T. Cantat. (**2016**). *Chem. Eur. J.*, **22**, 2930.
- [13] C. Lescot, S. Savourey, P. Thuery, G. Lefevre, J. C. Berthet, T. Cantat. (**2016**). *C. R. Chimie*, **19**, 57.
- [14] A. Tlili, A. Voituriez, A. Marinetti, P. Thuery, T. Cantat. (**2016**). *Chem. Commun.*, **52**, 7553.
- [15] N. von Wolff, G. Lefevre, J. C. Berthet, P. Thuery, T. Cantat. (**2016**). *ACS Cat.*, **6**, 4526.
- [16] N. von Wolff, C. Villiers, P. Thuery, G. Lefevre, M. Ephritikhine, T. Cantat. (**2017**). *Eur. J. Org. Chem.*, **2017**, 676.
- [17] N. von Wolff, J. Char, X. Frogneux, T. Cantat. (**2017**). *Angew. Chem. Int. Ed.*, **56**, 5616.