

# Investigating the relationships between chemical element concentrations and discharge to improve our understanding of their transport patterns in rural catchments under subtropical climate conditions

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- 1 Investigating the relationships between chemical element concentrations
- 2 and discharge to improve our understanding of their transport patterns in
- 3 rural catchments under subtropical climate conditions
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- 20 **Abstract** Solute and particulate elemental concentrations (C) exhibit different
- 21 responses to changes in discharge (Q), and those relationships are not well
- 22 understood in subtropical agricultural environments. The objective is to describe
- 23 the transport processes of different chemical elements during a set of
- 24 contrasted rainfall events (2011-2015) that occurred in a small rural catchment
- 25 under subtropical climate. The study was carried out in the Lajeado Ferreira
- 26 Creek catchment (1.23 km²), southern Brazil. To this end, the concentrations in
- 27 dissolved organic carbon (DOC), Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>-</sup>, ten chemical elements (in
- 28 either dissolved or particulate forms) and suspended sediment concentrations

(SSC) were determined. Metric indices were then calculated to characterize their transport patterns: (i) the best fit slope between log-C and log-Q (β), (ii) the coefficient of variation of C and Q, (iii) shape of the hysteresis loop and hysteresis index, and (iv) the flushing index. All particulate elements along with the dissolved inorganic phosphorus (PO<sub>4</sub>-3) were shown to be controlled by the sediment dynamics. Geogenic elements (Fe<sup>2+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Mn<sup>2+</sup>, Si<sup>4+</sup>) showed a dilution effect with increasing Q values, likely because they were mainly transported with subsurface and base flow. Dissolved elements that are mainly supplied with fertilizers (Na<sup>+</sup> and Cl<sup>-</sup>) as well as DOC showed a dilution effect, although they were mainly transported by surface runoff. Finally, a chemostatic behavior was found for those chemical elements (Mg2+, K+, Ca2+, NO3- and SO<sub>4</sub><sup>2</sup>-) that are supplied by more than one flow pathways. The results demonstrate that under subtropical climate conditions, the transport of essential nutrients including PO<sub>4</sub><sup>-3</sup> and metals (in particulate form), are mainly transported with surface runoff. Accordingly, runoff control on cultivated hillslopes should be improved to reduce the potential contaminant supply to the river and to reduce the potentially deleterious impacts that they may cause in downstream regions.

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**Keywords:** rainfall events; export regimes; solutes; metric indexes; water quality

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### 1. INTRODUCTION

The intensification of agricultural activities and the limited adoption of soil and water conservation practices have increased the formation of surface runoff and accelerated soil erosion (Deuschle et al., 2019; Londero et al., 2018),

sediment, nutrient and metal transport in river systems draining agricultural land around the world (Basu et al., 2010; Minella et al., 2018; Williams et al., 2018). Chemical elements supplied in excess to the streams may originate from anthropogenic activities, including agriculture through the supply of fertilizers and agrochemicals (Thomas et al., 2016), or from natural sources, as the result of rock weathering and leaching (Verheyen et al., 2015). The relationships observed between the chemical element concentrations (C) and water discharge (Q) result from the combined integration of biogeochemical processes, the contribution of the sources, flow pathways, and factors such as lithology, soil type and climate (Godsey et al., 2009; Knapp et al., 2020; Rumsey et al., 2017; Verheyen et al., 2015; Von Freyberg et al., 2017). Quantifying these relationships may contribute to improve our understanding of the transport of solutes and particulate chemical substances from upper catchment parts to the outlet.

The source of each element can be identified by reconstructing the hysteresis phenomenon based on the C – Q relationship (Lloyd et al., 2016; Wymore et al., 2019). Hysteresis analyses have been widely used and progressively improved during the last several decades and they provide a powerful technique for identifying the sources of both particle-bound (Duvert et al., 2010; Lawler et al., 2006; Wymore et al., 2019) and dissolved substances (Lloyd et al., 2016; Williams et al., 2018). Hysteresis occurs when concentrations of an element, observed in either dissolved or particulate form, during the rising stage of the hydrograph differ from those concentrations recorded at the same discharge rate during the falling stage. The hysteresis can be observed by the formation of a loop when plotting C *vs* Q.

There are three main potential interpretations of a given hysteresis loop (Lloyd et al., 2016). Clockwise hysteresis suggests the rapid mobilization of the chemical element of interest and the proximity of its source to the catchment outlet. This situation occurs when higher concentrations are observed during the event rising stage. However, when the loop is counterclockwise (i.e. when the concentration of an element is lower in the rising limb while increasing in the falling limb), the source of the chemical elements is expected to be located far from the catchment outlet. Furthermore, the eight-type hysteresis loop can be a sequence of clockwise and counterclockwise patterns. In addition, Lawler et al. (2006) proposed the hysteresis index (HI) which varies according to the magnitude of the rainfall event itself and, consequently, to the hydrological dynamics of the catchment.

Recently, metric indices have been used to evaluate these C-Q relationships, including (i) the best fit slope ( $\beta$ ) for the log (C) - log (Q) linear regression (Godsey et al., 2009); (ii) the ratio of the concentration and water discharge coefficients of variation ( $CV_C/CV_Q$ ) (Thompson et al., 2011); (iii) and a combination of both indicators (Musolff et al., 2015). The combination of both metric indices ( $\beta$  and  $CV_C/CV_Q$ ) results in an unique structure for the interpretation of transport patterns (Musolff et al., 2015): they can be (i) chemostatic (no change in element concentration with the increase or decrease in Q), (ii) they can show a chemodynamic dilution (as Q increases, the concentration of the chemical element is reduced), (iii) or display a chemodynamic enrichment (increase in the concentration of chemical elements with increasing Q). Besides these types of behavior, the flushing index (FI), which ranges between -1 and 1, quantifies the changes in concentrations

during a rainfall event (Rose et al., 2018; Vaughan et al., 2019). Negative FI values indicate that concentrations decrease during the hydrograph's rising limb. In contrast, positive FI values indicate increased concentrations during the hydrograph's rising limb. The interpretation of these indices may reveal the different transport patterns of the chemical elements that may be measured during the monitoring of rainfall events.

From this new approach, Rose et al. (2018) distinguished the following transport patterns: dilution for geogenic/exogenous elements ( $Ca^{2+}$ ,  $Mg^{2+}$ ,  $Si^{4+}$ ,  $Cl^-$ ,  $NO_3^-$ ); constant pattern for biologically associated solutes (carbon and dissolved nitrogen,  $NH_4^+$ ,  $K^+$ ,  $PO_4^{-3-}$ ); and enrichment for particulate substances (suspended solids and total) in a dataset covering 11 years collected in Pennsylvania (USA). Through the  $CV_C/CV_Q$  ratio and based on a set of seven-year rainfall events monitored in southern Brazil, Piazza et al. (2018) showed that the export dynamics of solutes were similar in agricultural and in native forest catchments. In South Florida, a combined analysis between the  $CV_C/CV_Q$  and  $\beta$  showed that total nitrogen (TN) presented a chemostatic behavior while total phosphorus (TP) was chemodynamic; this was due to a combination of fertilizer doses, low P retention in the soil and high storms (Wan et al., 2017).

In subtropical and tropical climate conditions, the relationship between C and Q is not well known at the event-scale. In these environments the annual average rainfall volume is high as well as the kinetic energy (Ramon et al., 2017), which can impact the C - Q relations during rainfall events. However, there is little information on the event scale for these climates. Gwenzi et al. (2017) reports the scarcity of information on the scale of hydrological (Q) and biogeochemistry events for tropical climate, more precisely in Africa. In Brazil,

water quality monitoring programs generally collect samples only four times per year following a routine schedule, due to limited human and financial resources applied in this program, where the measurements do not usually occur during events. Few research efforts sought to understand the influence of climate and land use conditions on the dynamics of solute transport under these environmental conditions (Piazza et al., 2018). Agriculture is the main economic activity in the Southern region of Brazil (CONAB, 2019), resulting in a high pressure on natural resources, especially soil and water. Consequently, the environmental impact of this activity must be a constant concern, where the limited adoption of conservationist agriculture associated with the erosive potential of the rains in the region, increase the degree of soil and water degradation (Didoné et al., 2015; Merten et al., 2015). Therefore, it is important to know the hydrological and biogeochemical behavior of rural catchments in a subtropical climate like in southern Brazil, where there is no prior knowledge on the subject.

In this study, we aim to investigate the transport processes of different chemical elements during a set of contrasted rainfall events through the analysis of their C - Q relationships. For this purpose, a program to monitor the transport of chemical elements in a small rural catchment representative of those environments found on the Southern Plateau of Brazil and in neighboring regions, under subtropical climate conditions, was established in the period between 2011 and 2015.

### 2. MATERIALS AND METHODS

### 154 2.1 Study area

The experimental catchment of the Lajeado Ferreira Creek – Arvorezinha (1.23 km²), is located on the border of the Brazilian southern plateau (Figure 1). This is a typical headwater catchment of the Jacuí River, one of the main tributaries of the Guaíba lake, which supplies drinking water to more than 4 million people (Cargnin et al., 2013). Its altitude ranges from 580 to 730 m (Figure 1). The upper third of the catchment has an undulating plateau relief with slopes up to 7%. The middle and lower thirds of the catchment have a much steeper topography with slope gradients often exceeding 15%.

The climate is classified as Cfb (subtropical super-humid with no dry season and warm summer) according to Köppen (Alvares et al., 2013). The average annual rainfall is ca. 1,938 mm (15 years, 2002 - 2016) and erosivity index (EI<sub>30</sub>) is 9,344 MJ mm ha<sup>-1</sup> yr<sup>-1</sup> (Ramon, 2017). Rainfall is well distributed over the year, although a wetter season with more rainfall events of high intensity can be identified during the spring season with the frequent occurrence of surface runoff.

The soil types in the catchment are Acrisols (57%), Cambisols (10%) and Leptosols (33%). The spatial and temporal variability of land use and soil management was determined using a combination of satellite images and field surveys using a handheld Global Positioning System device. Land use and soil management in 2011 included native forest (15% of the total catchment surface area), reforested zones (23.2%), pasture (6.5%), cropland (soybean + tobacco) under minimum tillage (26%), cropland under conventional tillage (13%), fallow (13%) and other minor uses (3.3%). By 2015, land use and soil management had significantly changed, mainly through an increase in reforested areas (35%), a decrease in cropland under minimum tillage (15%), while the cropland

surface proportion under conventional tillage increased to 24%, and that under fallow decreased to 2.2%. In contrast, the respective surface areas under native forest, pasture, and other land uses did not change.

### 2.2 Hydro-sedimentary monitoring

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The Arvorezinha catchment has been monitored since 2002, with the continuous measurement of flow discharge (Q) and suspended sediment concentrations (SSC) at the outlet of the catchment in a Parshall type flume built in concrete with a 1.83 m width in its narrowest part. Water level and turbidity were recorded through a float-type water level sensor (Thalimedes, OTT HydroMet, Germany) and a turbidimeter (Model SL-2000, Solar®, Brazil), respectively. All automatic equipment recorded data every 10 minutes. SSC data were obtained by the turbidimeter properly calibrated with samples collected manually using an isokinetic sampler (US DH-48) during rainfall-runoff events as proposed by Merten et al. (2014). The 10 min interval data was used to calculate the variations of the sediment yield throughout the year. Rainfall intensity and depth have been measured by means of tipping bucket rain gauges and pluviometers, respectively, and their location is shown in Figure 1. The rainfall of the last five days (R5) before the rainfall events analyzed was computed to reflect the antecedent soil moisture conditions (SILVEIRA et al., 2000), which is classified into three classes according to rainfall volume: 1) dry when R5 < 36 mm; 2) field capacity when R5 > 36 and < 53 mm; and 3) saturated when R5 > 53 mm.

The runoff separation (the separation of surface runoff and base flow contributions) was performed for the 16 rainfall events monitored, using the graphical analysis method of Chow et al. (1988). The onset of the surface runoff

is determined by the rise of the hydrograph until the moment when the discharge returned to values close to those observed just before the rising limb, thus characterizing the end of the rainfall event.

### 2.3 Sampling and chemical analysis

During the rainfall - runoff events, 164 water samples were collected at the catchment outlet using a isokinetic sampler. Duplicate samples were collected, one of them to calculate the sediment concentration following the evaporation method according to Shreve and Downs (2005) and the second sample was used for determining the water chemical composition, which is described in section 2.3.1.

### 2.3.1 Chemical analysis

In the laboratory, an aliquot of the sample was filtered at  $0.22~\mu m$  to characterize the dissolved fraction. The total element concentration was determined in the non-filtered samples. Then, the particulate fraction was obtained through calculating the difference between total and dissolved concentrations.

In the filtered sample, the dissolved organic carbon (DOC) content was determined after mixing an aliquot of the filtered sample with a 1:1 sulphochromic solution incubated in an oven for 4 h at 60°C, with a spectrophotometer at 580 nm. Sulphate (SO<sub>4</sub><sup>2-</sup>), chlorine (Cl<sup>-</sup>) and nitrate (NO<sub>3</sub><sup>-</sup>) contents were measured with a high-performance liquid chromatography (HPLC) - Sykam S 135 Ion Chromatography System (U.S. EPA, 1997). Finally, the concentration of dissolved Ca<sup>2+</sup>, Cu<sup>2+</sup>, Fe<sup>2+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Mn<sup>2+</sup>, Na<sup>+</sup>, PO<sub>4</sub><sup>3-</sup>, Si<sup>4+</sup>,

and Zn<sup>2+</sup> was quantified by Inductively Coupled Plasma - Optical Emission Spectrometry (ICP-OES, Perkin-Elmer) directly in the filtered water sample (U.S. EPA, 1994).

Total content in unfiltered samples of water + sediments was assessed after digestion with *aqua-regia* (U.S. EPA, 1996). In brief, a 20 mL aliquot of unfiltered sample of water + sediment was inserted into Teflon tubes together with 0.5 mL of HCl (37%) and 1 mL of HNO<sub>3</sub> (65%), followed by microwave assisted digestion for 9.5 min at a temperature of 182 °C. After that, Ca, Cu, Fe, K, Mg, Mn, Na, P, Si and Zn were quantified using Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES, Perkin-Elmer) (U.S. EPA, 1997). Then, particulate contents of these elements were obtained by calculating the difference between total and dissolved contents.

### 2.4 Data analysis

The relationship between the element concentrations and the water discharge was evaluated by means of metric indices: slope ( $\beta$ ),  $CV_C/CV_Q$ , qualitative (loop) and hysteresis index. The analyses were performed for the 16 rainfall events monitored, evaluating the SSC, DOC,  $NO_3^-$ ,  $SO_4^{2-}$ ,  $CI^-$  and the ten chemical elements analyzed ( $Ca^{2+}$ ,  $Cu^{2+}$ ,  $Fe^{2+}$ ,  $K^+$ ,  $Mg^{2+}$ ,  $Mn^{2+}$ ,  $Na^+$ ,  $Zn^{2+}$ ,  $PO_4^{-3}$ ,  $Si^{4+}$ ).

### 2.4.1 Metric indexes

To obtain the  $\beta$ , all Q and C data were log transformed. Accordingly, a  $\beta$  value was obtained for each element for the 16 rainfall events. A negative  $\beta$  value reflects the decrease of element concentration with the increase of Q,

whereas positive values indicate an increase of both C and Q. When  $\beta \sim 0$ , it indicates a chemostatic behavior of the element. The chemostatic pattern was defined when the  $\beta$  was considered to be significantly different from zero, at  $\alpha = 0.05$  if Student's t statistic | t | > 2 (Helsel and Hirsch, 2002; Rose et al., 2018).

The coefficient of variation of the concentration in each chemical element was calculated (CV<sub>C</sub>), as well as the coefficient of variation of water discharge (CV<sub>Q</sub>). The CV<sub>C</sub>/CV<sub>Q</sub> metric index indicates whether the behavior of the element is chemostatic (CV<sub>C</sub>/CV<sub>Q</sub>  $\leq$  0.5), or chemodynamic (CV<sub>C</sub>/CV<sub>Q</sub> > 0.5). According to Musolff et al. (2015) both  $\beta$  and CV<sub>C</sub>/CV<sub>Q</sub> need to be evaluated in combination.

In addition, we also examined hysteresis patterns in dissolved and particulate form of the elements during individual rainfall events, as well as the hysteresis index (HI), with the method of Lawler et al. (2006). The method is based on measuring the width of the hysteresis loop in the central water discharge ( $Q_{50}$ ), when there is a 50% increment of the Q compared to the beginning of the rainfall event. The  $Q_{50}$  is calculated with equation 1, based on the minimum ( $Q_{min}$ ) and maximum water discharge ( $Q_{max}$ ) values during the rainfall event.

$$Q_{50} = 0.5 \times (Q_{max} - Q_{min}) + Q_{min} \tag{1}$$

Thus, with the  $Q_{50}$  calculated, the chemical element concentration and SCC for the exact corresponding moment was obtained by a coumpound rule of three, using the closest sample collected before and after the  $Q_{50}$  time in the rising and falling limbs. Therefore, HI was calculated according to equations 2 and 3, when the concentration in the rising limb - RL is higher than in the falling

280 limb - FL (clockwise hysteresis) and RL < FL (counterclockwise hysteresis), 281 respectively.

$$HI = \left(\frac{RL}{FL}\right) - 1\tag{2}$$

$$HI = \left(\frac{-1}{\frac{RL}{FL}}\right) + 1\tag{3}$$

Where: RL is the element concentration in water discharge on the rising limb of the hydrograph and FL is the concentration on the falling limb.

The Flushing Index (FI) was calculated according to the method of (Vaughan et al., 2019) (Equation 4). The water discharge and the element concentration had to be normalized, according to equations 5 and 6, respectively.

$$FI = C_{Opeak,normal} - C_{initial,normal}$$
 (4)

Where:  $C_{Qpeak,normal}$  and  $C_{initial,normal}$  are the normalized elements during the peak and the beginning of each rainfall event, respectively.

$$Q_{i,normal} = \frac{Q_i - Q_{min}}{Q_{max} - Q_{min}} \tag{5}$$

$$C_{i,normal} = \frac{C_i - C_{min}}{C_{max} - C_{min}} \tag{6}$$

Where:  $Q_i$  and  $C_i$  are the water discharge and element values at time step i, respectively,  $Q_{\text{max}}$  and  $Q_{\text{min}}$  are the maximum and minimum water

discharge values during the rainfall event, respectively, and  $C_{\text{max}}$  and  $C_{\text{min}}$  are the maximum and minimum concentrations of each element during the rainfall event, respectively.

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### 3. RESULTS

### 3.1 Hydro-sedimentary monitoring and element concentration

Table 1 shows the main characteristics of the 16 rainfall events monitored between 2011 and 2015. The peak flow varied from 35 to 3,399 L s<sup>-1</sup> while the maximum suspended sediment concentrations fluctuated from 0.1 to 5.4 g L<sup>-1</sup>. The magnitude of peak flows followed the magnitude of the volume and duration of rainfall (Table 1). However, it is important to consider the effect of the previous events on the hydro-sedimentary dynamics, because even with low rainfall volume, surface runoff generation, sediment and chemical element transport may occur. The rainfall of the last five days (R5) are presented in Table 1. It can be observed that among all the events analyzed, there was no rainfall at all during the previous five days for four events and for five events the R5 was < 36 mm, corresponding to a dry condition. Five other events were classified as having field capacity humidity, and two rain events considered as saturated soil. The SSC during this period (2011-2015) was not high, except for the events that occurred on 18/09/12 and 12/07/15 (Table 1). According to studies conducted in this catchment between 2002 and 2016, the minimum and maximum SCC recorded values are 0 and 13.7 g L<sup>-1</sup>, respectively (Minella et al., 2018). However, considering the duration and the surface runoff (Q<sub>runoff</sub>) of a certain event, the sediment yield (SY) may still be high. An example is the event of 20/07/2011 that presented low average SSC (SSC<sub>ave</sub> - 0.2 g L<sup>-1</sup>) and a maximum value of 0.6 g  $L^{-1}$ , but due to high  $Q_{runoff}$ , generated high SY (24.8 ton). The opposite situation was observed for the event that occurred on 12/07/2015, where the  $SSC_{ave}$  was higher, but the SY was lower, because the  $Q_{runoff}$  was much lower. Annual and event-based hydrographs and sedimentographs during sampling can be found in the supplementary material.

Figure 2 shows the preferential form of transport of each chemical element. A significant proportion of Ca<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup>, and Mg<sup>2+</sup> transport occurred in dissolved form (47, 41, 39 and 28% respectively). In contrast, Mn, Fe, P, Cu, Si and Zn were mostly transported in particulate form (99, 97, 91, 90, 85 and 79%, respectively).

Table 1: Hydro-sedimentary variables monitored during 16 rainfall events from 2011 to 2015 in the Arvorezinha catchment.

### 3.2 Chemical element exports

Table 2 shows the number of rainfall events, the number of samples (n), minimum, average, and maximum concentrations for each element. A large number of samples were collected during the rising, peak and falling limbs of the hydrograph resulting in a total of 164 samples, although for some elements, mainly anions, only the samples of a restricted number of events (n= 7) could be analyzed (Table 2). The analyzed samples cover a wide range of concentrations, due to the variability of rainfall events in terms of magnitude and seasonality.

Table 2: Chemicals, total rainfall events, total of samples, maximum, mean and minimum concentration, log (C) – log (Q) regression slopes ( $\pm$ SE) and CV<sub>c</sub>/CV<sub>Q</sub> of dissolved and particulate elements during events from 2011 to 2015 in the Arvorezinha catchment.

### 3.2.1 Dilution patterns

Analyzing separately the metric indices  $\beta$  and  $CV_C/CV_Q$ , the following results were obtained with  $\beta$  < 0 significantly different from zero (p < 0.05): -0.32 ± 0.11 for DOC and -0.12 ± 0.03 for Na<sup>+</sup>; where their concentration decreased when Q increased (Table 2). Although the  $\beta$  value for Cl<sup>-</sup> was not significantly different from zero (p < 0.06), its value was also < 0. The  $CV_C/CV_Q$  values obtained for DOC and Cl<sup>-</sup>, were 0.83 and 0.58, respectively, which are > 0.5, indicating a chemodynamic pattern (Table 2). In contrast, for Na<sup>+</sup>, the value of  $CV_C/CV_Q$  was 0.42, indicating a chemostatic pattern.

### 3.2.2 Enrichment patterns

For SSC and for the concentrations in particulate form of all the chemical elements analyzed, the  $\beta$  values were > 0 (p < 0.05), with the exception of that obtained for Zn, which was not statistically different from zero (p < 0.13). This behavior indicates an enrichment (Table 2), and it was also observed for  $PO_4^{3-}$  and  $Zn^{2+}$ . Similarly, the  $CV_C/CV_Q$  ratio was > 0.5 for all elements in particulate form, SSC,  $PO_4^{3-}$  and  $Zn^{2+}$ , characterizing a chemodynamic pattern. In addition, for some particulate elements (Ca, K, Mg, Mn and Zn) and for  $PO_4^{3-}$ , the  $CV_C/CV_Q$  ratio is  $\geq$  1, which is classified as a strong chemodynamic pattern.

### 3.2.3 Chemostatic patterns

The concentration of NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Ca<sup>2+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Cu<sup>2+</sup>, Fe<sup>2+</sup>, Mn<sup>2+</sup> and Si<sup>4+</sup> showed little variations with water discharge, because  $\beta \sim 0$ . All these elements, except for SO<sub>4</sub><sup>2-</sup> ( $\beta = -0.09 \pm 0.03$ , p < 0.01) and Ca<sup>2+</sup> ( $\beta = -0.04 \pm 0.02$ , p < 0.03), showed no significant difference from zero, at p < 0.05. The chemostatic pattern was confirmed by CV<sub>C</sub>/CV<sub>Q</sub> ratio values  $\leq 0.5$  for NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Ca<sup>2+</sup>, K<sup>+</sup> and Mg<sup>2+</sup> (Table 2). However, Cu<sup>2+</sup>, Fe<sup>2+</sup>, Mn<sup>2+</sup> and Si<sup>4+</sup> were associated with a CV<sub>C</sub>/CV<sub>Q</sub> ratio > 0.5, which would indicate a chemodynamic pattern (Table 2).

When the two metric indices are combined, it is possible to obtain better insights into the transport patterns of the analyzed elements (Gwenzi et al., 2017) (Figure 3). A major part of the elements was characterized by an increase in their concentration with increasing water discharge ( $\beta > 0$ ;  $CV_C/CV_Q \ge 0.5$ ). Furthermore, those elements such as dissolved  $PO_4^{3^-}$  and  $Zn^{2^+}$  were among those with the strongest chemodynamic behavior ( $CV_C/CV_Q \ge 1$ ). In contrast, those elements with  $\beta \sim 0$  and  $CV_C/CV_Q < 0.5$  showed a weak relationship between their concentration and the water discharge, mainly  $Mg^{2^+}$ ,  $K^+$ ,  $Ca^{2^+}$ ,  $NO_3^-$  and  $SO_4^{2^-}$  (Figure 3).  $Na^+$ ,  $Cl^-$  and DOC exhibit a dilution behavior.

### 3.3 Hysteresis and flushing analysis

The occurrence of a clockwise pattern was more frequent (> 50%) for K<sup>+</sup>, Fe<sup>2+</sup>, Mn<sup>2+</sup> and Na<sup>+</sup>, SSC and for all particulate elements (Table 3). PO<sub>4</sub><sup>3-</sup> and particulate Na presented equal proportions of clockwise (50%) and counterclockwise (50%) hysteresis behaviors. All particulate elements and SSC, except for Zn, presented positive HI values. Fe<sup>2+</sup>, Mn<sup>2+</sup> and particulate Zn,

showed negative HI values (-1.18, -0.67 and -1.31, respectively), but the clockwise hysteresis pattern predominated. Among the dissolved chemical elements, only  $NO_3^-$ ,  $K^+$ ,  $Na^+$ ,  $PO_4^{3^-}$  and  $Si^{4^+}$  showed positive HI values, although  $NO_3^-$  did not show a well-defined hysteresis pattern (Table 3).

For Cl<sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Ca<sup>2+</sup>, Cu<sup>2+</sup>, Mg<sup>2+</sup>, Si<sup>4+</sup>, Zn<sup>2+</sup> and DOC, a dominance of counterclockwise patterns occurred (> 50%) which is supported by their negative HI values (Table 3). The only exception was observed for DOC and Si<sup>4+</sup> that showed a positive HI value, with the occurrence of a few events with high HI values, resulting in a positive average value.

Similarly, all particulate elements and SSC presented positive FI values, indicating that their concentration increased during the hydrograph's rising limb (Table 3). All the dissolved elements, except for PO<sub>4</sub><sup>3-</sup>, together with DOC showed negative values for FI (Table 3), indicating that their concentration decreased during the hydrograph's rising limb.

Table 3: Chemicals, mean Hysteresis Index (HI) and Flushing Index (FI) values and percentages of distinct hysteresis patterns during these events monitored from 2011 to 2015 in the Arvorezinha catchment.

When plotting FI *vs* HI, additional information may be derived to improve our understanding of the transport and the sources of these elements during the investigated rainfall events (Figure 4).

In Figure 4a, the elements with negative FI and positive HI are found (NO<sub>3</sub>-, Si<sup>4+</sup>, K<sup>+</sup>, Na<sup>+</sup> and DOC). These elements show a dilution effect and they are supplied by nearby sources, which is particularly well defined for DOC. The

dilution effect indicates that this element reaches the river from one of the main flow pathways, which is hypothesized to be surface runoff, since organic carbon concentrations are higher in the topsoil layers. However, the slight predominance of the counterclockwise direction for DOC may be surprising (56%, Table 3), as we expected a clockwise predominance when examining the hysteresis behavior alone without considering other indices (Figure 4a). All other elements are characterized by HI values close to zero, suggesting the absence of well-defined hysteresis patterns (clockwise, counterclockwise, or complex).

All the elements in particulate form, except Zn, SSC and dissolved PO<sub>4</sub><sup>3-</sup> are characterized by an increase in their concentrations during the rising limb (Figure 4b). Furthermore, the source of these elements is expected to be in the vicinity of the catchment outlet, being delivered during the rising limb and depleted during the falling limb (Figure 4b). Particulate Zn is an exception because its main behavior is counterclockwise according to the HI value (its source is expected to be far from the catchment outlet) (Figure 4d). This analysis suggests that at the beginning of an event there is an increase of these element during the rising limb and it is supplied by more distant sources. However, for most events (57%, Table 3), the particulate Zn shows a clockwise pattern. For most events, the HI value was positive and close to zero for Zn. For only two events the HI values were more negative (-4.69 and -5.65 for the events occurred in 01/10/2011 and 20/09/2013, respectively), which makes the result for this element inconclusive.

Cl<sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Cu<sup>2+</sup> and Mn<sup>2+</sup> showed a dilution effect and they are likely supplied by distant sources, well characterized for Ca<sup>2+</sup>, Cu<sup>2+</sup>, Mn<sup>2+</sup>,

Fe<sup>2+</sup> and Zn<sup>2+</sup> that present more negative values (-0.22, -0.36, -0.67 and -1.18, respectively) (Figure 4c). The other three elements (Cl<sup>-</sup>,  $SO_4^{2-}$ ,  $Mg^{2+}$ ) were characterized by HI values close to zero (-0.03, -0.05 and -0.06, respectively), suggesting that they are supplied by closer sources during the event, than the other four elements found in Figure 4c.

### 4. DISCUSSION

The rainfall events investigated in the current research covered a pluriannual period, including different seasons and climatic phenomena. The year 2015 was classified as El Niño (NOAA) for Southern Brazil with annual rainfall exceeding 2,498 mm in the study site (Barros, 2016). In the agricultural crop season of 2011/2012, a La Niña phenomenon was registered, which may cause periods of drought in Southern Brazil (NOAA). Therefore, different hydrological behavior occurs in the catchment and they were covered by the period of sampling.

The chemical elements that can be strongly adsorbed by clays and iron oxides, were mainly transported in particulate form, including P, Fe and Mn with 91, 97 and 99%, respectively. In contrast, large proportions of ions such as Ca, K and Na were transported in dissolved form (48, 41 and 39%, respectively). However, it should be noted that these values refer to the average value calculated for the entire set of 16 rainfall events (Figure 2). When analyzing the dissolved and particulate proportions for each individual rainfall event, these proportions can be different. Accordingly, during those rainfall events with high SSC, almost 100% of the elements with a high affinity with sediment particles, such as P, Mn, and Fe, were transported in particulate form.

For events with lower sediment concentrations, the proportion of these elements in dissolved form increased. In contrast, elements with a low affinity for sediment particles, such as Ca and K, were mainly transferred in dissolved form whatever the SSC conditions observed during the event.

Three different transport regimes are described for the analyzed elements in the next sections, namely: (i) chemodynamic - enrichment, (ii) chemodynamic - dilution and (iii) chemostatic.

### 4.1 Drivers for enrichment patterns

All the elements in particulate form (Ca, Cu, Fe, K, Mg, Mn, Na, P, Si and Zn), as well as SSC and dissolved  $PO_4^{3-}$  showed a concentration enrichment with increased discharges. Botter et al. (2019) observed a similar behavior for total phosphorous (Particulate P +  $PO_4^{3-}$ ), which shows higher concentrations with increasing discharges.

Sediment availability for transport is dependent on the kinetic energy of rainfall during previous events and on the occurrence of new sediment pulses, that may be triggered by landslides or channel bank erosion (Wymore et al., 2019). As the Arvorezinha catchment did not experience any landslide during the study period, the main sediment supply to the stream is associated with the occurrence of interill and rill erosion in cultivated areas (57% of the total sediment) (Tiecher et al., 2015). However, other potential sources were outlined as delivering a significant proportion of sediment and particle-bound elements. Unpaved roads were shown to provide 23% of the sediment (Tiecher et al., 2015) with preferential connections with the river network (Barros, 2016). This high connectivity between the roads and the stream systems allows the quick

delivery of readily available sediment to the stream (Figure 1). These results are consistent with the findings of the current research showing that the sources of particle-bound elements are close from the monitoring station (as these parameters showed a clockwise pattern in > 50% of the events), which may explain the rapid mobilization of the sediment and the elements bound to or contained in the particles (Table 3, Figure 4).

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PO<sub>4</sub><sup>3</sup>- was the only compound in dissolved form that presented the same behavior as the particulate elements (Figure 4). Bender et al. (2018) evaluating the P (dissolved and particulate form) dynamics in the same place of the present study and through eight rain events already verified a strong correspondence between the PO<sub>4</sub><sup>3-</sup> behavior with its particulate form. This is likely related to the saturation of soils with the use of high doses of phosphate fertilizers for tobacco cultivation. Pellegrini et al. (2010) showed that sediment eroded from soils cultivated with tobacco were saturated with phosphates and their desorption capacity was extremely high and fast. The dissolved form of nutrients essential to the development of agricultural crops, such as PO<sub>4</sub><sup>3</sup>, is the form that they will be available in the aqueous medium and can trigger new reactions or processes (Tang et al., 2019). In this catchment, a high availability of PO<sub>4</sub>3- is found in water and sediments, as already observed in a previous study (Bender et al., 2018). A study conducted in France and compiling 40-year records of water quality data from 293 monitoring stations showed that the mitigation of phosphorus pollution reduced the concentrations in this substance and altered the corresponding C - Q slopes (Moatar et al., 2017). Therefore, understanding how the transfer of these elements occurs is of fundamental importance, and is linked to the climatic conditions associated with the

characteristics of the catchment responsible for generating runoff and erosive processes.

### 4.2 Drivers for dilution patterns

Fe<sup>2+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup> and Mn<sup>2+</sup> exhibited an increase in their concentration in the rising limb, but during the event, other flow pathways with low concentration in these elements have reached the catchment outlet and contributed to a dilution phenomenon (Figure 3). All these elements displayed counterclockwise patterns, reflecting their supply by sources that are located far from the catchment outlet (Figure 4c).

More specifically, Fe<sup>2+</sup>, Si<sup>4+</sup> and Mn<sup>2+</sup> are released by natural weathering processes and they remain in soluble form in the soil water compartment. Previous studies that examined the hydrological dynamics in the Arvorezinha catchment revealed that subsurface and base flows may supply 60 to 80% of the streamflow during a rainfall event in autumn and winter (Barros, 2016; Robinet et al., 2018). Accordingly, the greater frequency of counterclockwise hysteresis behavior patterns may be explained by an important contribution of subsurface flow during rainfall events.

Dissolved Cu<sup>2+</sup> and Zn<sup>2+</sup> showed a similar behavior than Fe<sup>2+</sup>, Si<sup>4+</sup> and Mn<sup>2+</sup> (Figures 3 and 4). The increased concentrations measured at the beginning of the rainfall event were likely due to the release of these elements by weathering processes, as young soils (Acrisols, Cambisols and Leptsols) of basaltic origin are found in this catchment. In addition, in this area, low inputs of Cu<sup>2+</sup> and Zn<sup>2+</sup> are expected from the use of fertilizers and agrochemicals

(exogenous solutes), which reinforces the different behavior observed compared to that of DOC and dissolved Na<sup>+</sup> and Cl<sup>-</sup>.

Na<sup>+</sup>, Cl<sup>-</sup> and DOC are preferentially transported with surface runoff, due to their high concentrations in the topsoil layer. The high concentration of Na<sup>+</sup> and Cl<sup>-</sup> in water may be explained using inorganic fertilizers for tobacco cultivation (NaNO<sub>3</sub> and KCl), which occupied 15 to 20% of the total catchment surface area between 2011 to 2015. Finally, organic carbon is concentrated in the topsoil layer due to inputs from cover crops; although the levels found may vary depending on land use and soil management (Boix-Fayos et al., 2017; Endale et al., 2017).

### 4.3 Chemostatic patterns

In general, Mg<sup>2+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>-2</sup> showed a very weak C - Q relationship. Several sources (anthropogenic and geogenic) may supply these elements, which can come from the storage of these solutes in the soil and by surface runoff. Thus, elements such as K<sup>+</sup>, may not have their concentration altered with the increase of Q, because all flow pathways to the river may have similar concentrations, causing no dilution effect nor enrichment with the changes in the water discharge. In White Clay Creek, Pennsylvania – USA, the K<sup>+</sup> was the only element among those analyzed that presented a chemostatic behavior and a complex hysteresis pattern, as it can be considered a biological solute, geological weathering and additional K<sup>+</sup> inputs from exogenous sources such as fertilizer (Rose et al., 2018). In Arvorezinha catchment, the production of the tobacco requires high inputs of lime and industrial fertilizers (anthropogenic source) containing K<sup>+</sup>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>-2</sup>. Both ions have a low

bridging energy to clay, oxides, and organic matter, which may facilitate their release into the soil and possibly be transported by subsurface flow towards the catchment outlet. In addition, another fraction of these elements can be carried by surface runoff, due to their higher concentration on the topsoil layer. Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup> were already shown to be associated with geogenic sources in a study conducted under a subtropical climate in Brazil (Piazza et al., 2018). In Nevada, USA, eight small river catchments were monitored (0.5 to 2.3 km<sup>2</sup>) and during the 8-years study period, Mg<sup>2+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Cl<sup>-</sup>, Na<sup>+</sup> and SO<sub>4</sub><sup>-2</sup> revealed a chemostatic behavior under temperate climate, although an opposite situation was observed for NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-1</sup> (Hunsaker and Johnson, 2017).

### 5. CONCLUSIONS

The results show the strong correspondence between the magnitude of the rainfall event and the preferred form of transportation of the chemical elements. The water discharge is linked to the magnitude of the rainfall event and the generation of surface runoff, the main responsible for the transport of sediment and potential contaminants to the river. This can be attributed to the strong erosive potential of rainfall in subtropical region combined to use and soil management, that have an impact on the hydrological dynamics of the catchment. Practices that may reduce the supply of water (by surface runoff) and sediment to the river network will also minimize the transport of elements in particulate form (Ca, Cu, Fe, K, Mg, Mn, Na, P, Si and Zn), which is the preferred form of chemical elements transport in this catchment. In the same

way, PO<sub>4</sub><sup>3-</sup> concentration increased with the surface runoff, having the same behavior of particulate elements, preferentially transferred by surface runoff.

Most of the dissolved chemical elements are associated to both element sources (geogenic/anthropogenic) and flow pathways (base, subsurface and surface flow). Thus, elements that are not preferably transported by surface runoff, such as geogenic (Fe<sup>2+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Mn<sup>2+</sup> and Si<sup>4+</sup>), reduce their concentration with the increase in water discharge, and the dilution process occurs. DOC, Na<sup>+</sup> and Cl<sup>-</sup> also have a dilution regime export (concentrations decrease with the increase of Q). However, their source is anthropogenic through fertilization and soil organic matter and mainly transported by the surface runoff.

The chemostatic regime export is linked to chemical elements that are supplied by more than one flow pathway  $(Mg^{2+}, K^+, Ca^{2+}, NO_3^-)$  and  $SO_4^{2-}$ , due its potential contribution by both geogenic and anthropogenic source.

Overall, the results show that despite a small data set, it was possible to understand the transport of some important chemical elements (nutrients, geochemicals and metals) in a subtropical environment and it could be better understood when the different metric indices were used together. These insights are fundamental to design strategies to control surface runoff and erosive process, because as we have observed, they seem to be the main responsible for the chemical elements transport in this region.

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808	Figure 1 - Location of the study site in Brazil and the hydro-sedimentary
809	monitoring within the catchment.
810	
811	Figure 2 - Average distribution of chemical elements in dissolved and
812	particulate forms during the rainfall events monitored from 2011 to 2015.
813	
814	Figure 3: Plot of log (C) – log (Q) regression slope – $\beta$ versus $CV_C/CV_Q$ for all
815	elements determined in the current study. The plotted values correspond to the
816	mean of those measurements conducted on all the samples analyzed during
817	the 2011 – 2015 period. Error bars represent $\pm$ 1 Standard Error $\beta$ . SSC:
818	suspended sediment concentration; DOC: dissolved organic carbon.
819	
820	Figure 4: Plots of median hysteresis index versus median flushing index for
821	chemical across all sampled rainfall events from 2011 to 2015. SSC: suspended
822	sediment concentration; DOC: dissolved organic carbon. HI and FI values are
823	normalized considering their maximum value equal to 1.
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Table 1: Hydro-sedimentary variables monitored during events from 2011 to 2015 in the Arvorezinha catchment.

Doto	Т	R	R5	Q <sub>total</sub>	Q <sub>base</sub>	$Q_{runoff}$	$Q_{peak}$	С	$SSC_{peak}$	SSC <sub>ave</sub>	SY
Data	min			mm		•	Ls <sup>-1</sup>	%	g L	-1 -	ton
20/07/2011	1340	127	44	92.7	37.5	55.2	3399.4	43.4	0.6	0.2	24.8
01/10/2011	350	17	0	0.81	0.61	0.2	57.7	1.2	0.2	0.1	0.1
05/07/2012	1460	95	0	16.5	10.6	6.2	536.4	6.4	0.4	0.2	2.4
26/08/2012	2040	27	15	1.0	0.9	0.1	44.2	0.5	0.2	0.1	0.3
09/09/2012	540	26	0	9.1	8.9	0.2	35.6	0.7	0.3	0.1	0.1
18/09/2012	430	110	27	44.3	19.3	25	1470.8	22.7	5.3	1.1	20.6
12/03/2013	480	27	58	5.6	4.7	0.9	235.6	3.4	0.2	0.1	0.6
24/08/2013	*	64	16	0.9	**	**	45.5	**	0.1	0.1	0.1
20/09/2013	1980	84	18	47.6	22.2	25.4	785.3	30.2	2.2	0.7	70.6
26/10/2013	245	36	52	6.3	3.6	2.7	463.6	7.5	1.0	0.4	3.3
19/03/2014	430	34	35	2.1	1.3	0.9	125.9	2.5	0.7	0.3	0.9
23/07/2014	950	83	0	26.7	17.0	9.7	1034.5	11.7	1.1	0.2	6.4
06/09/2014	390	42	8	7.4	5.2	2.2	258.4	5.2	0.4	0.2	1.2
30/09/2014	950	42	51	20.2	12.5	7.7	520.5	18.2	0.2	0.1	1.9
08/07/2015	1370	48	36	19.7	17.8	1.9	536.4	4.0	0.3	0.1	2.2
12/07/2015	*	87	86	22.5	17.0	5.5	599.5	6.3	5.4	0.4	6.6

<sup>\*</sup> Events that lasted more than one day. \*\* This event was long, and the rainfall was of low intensity; accordingly, during the period of sampling, the flow level did not rise. T = duration of the rainfall, R = rainfall depth, R5 = antecedent rainfall of the last five days,  $Q_{\text{total}} = \text{total flow}$ ,  $Q_{\text{base}} = \text{total base flow}$ ,  $Q_{\text{runoff}} = \text{total surface runoff}$ ;  $Q_{\text{peak}} = \text{peak flow}$ , C = surface runoff coefficient,  $SSC_{\text{peak}} = \text{peak suspended sediment concentration}$ ,  $SSC_{\text{ave}} = \text{average suspended sediment concentration}$ , SY = sediment yield.

Table 2: Chemicals, total rainfall events, total of samples, maximum, mean and minimum concentration, log (C) – log (Q) regression slopes ( $\pm$ SE) and CV<sub>o</sub>/CV<sub>Q</sub> of dissolved and particulate elements during events from 2011 to 2015 in the Arvorezinha catchment.

Chemical	Events	Data	Maximum	Mean	Minimum	β	CV <sub>C</sub> /CV <sub>Q</sub>
SSC	16	164	4.4	0.3	0.0	$0.52 (\pm 0.08)^{\dagger}$	1.16
DOC	15	114	10.4	1.8	0.0	$-0.32 (\pm 0.11)^{\dagger}$	0.83
Dissolved							
Cl	7	53	16.4	3.0	0.6	-0.10 (± 0.05)	0.58
$NO_3^-$	7	53	2.7	1.2	0.4	-0.06 (± 0.03)	0.23
SO <sub>4</sub> <sup>2-</sup>	7	53	5.3	2.7	0.6	$-0.09 (\pm 0.03)^{\dagger}$	0.24
Ca <sup>2+</sup>	16	164	4333.7	2183.8	660.9	-0.04 (± 0.02) <sup>†</sup>	0.16
Cu <sup>2+</sup>	16	142	31.6	2.1	0.0	0.05 (± 0.06)	1.05
Fe <sup>2+</sup>	16	163	2034.6	303.7	0.0	0.07 (± 0.12)	0.89
K⁺	16	164	4548.3	2611.6	534.0	-0.01 (± 0.02)	0.13
Mg <sup>2+</sup>	16	164	1273.3	743.0	297.5	-0.02 (± 0.01)	0.11
Mn <sup>2+</sup>	16	126	22.0	3.6	0.0	0.08 (± 0.09)	0.74
Na⁺	16	143	4492.4	1210.7	0.0	-0.12 (± 0.03) <sup>†</sup>	0.42
PO <sub>4</sub> <sup>3-</sup>	16	150	642.5	27.7	0.0	$0.23 (\pm 0.08)^{\dagger}$	1.41
Si <sup>4+</sup>	16	163	10223.6	1621.7	0.0	0.01 (± 0.08)	0.63
Zn <sup>2+</sup>	16	164	77.7	10.6	0.3	$0.14 (\pm 0.06)^{\dagger}$	0.70
Particulate							
Ca	12	105	72709	2191.3	0.0	0.62 (± 0.11) <sup>†</sup>	2.05
Cu	11	95	98.6	12.3	0.0	$0.35 (\pm 0.14)^{\dagger}$	0.96
Fe	12	118	111452.0	13408.7	290.3	$0.66 (\pm 0.07)^{\dagger}$	0.97
K	12	103	28784.1	3218.0	0.0	$0.46 (\pm 0.09)^{\dagger}$	1.02
Mg	12	110	18021.1	1710.1	0.0	$0.51 (\pm 0.08)^{\dagger}$	1.10
Mn	12	118	4631.5	472.5	9.0	$0.77 (\pm 0.08)^{\dagger}$	1.03
Na	11	89	9438.0	1376.1	0.0	$0.45 (\pm 0.15)^{\dagger}$	0.99
Р	12	114	1936.1	203.8	0.0	$0.60 (\pm 0.07)^{\dagger}$	0.90
Si	12	112	53759.4	9383.1	0.0	$0.65 (\pm 0.08)^{\dagger}$	0.91
Zn	11	81	489.4	25.3	0.0	0.20 (± 0.13)	1.48

SSC: suspended sediment concentration - g L<sup>-1</sup>; DOC: dissolved organic carbon (mg L<sup>-1</sup>); CI, NO<sub>3</sub>, SO<sub>4</sub><sup>2-</sup> are expressed in mg L<sup>-1</sup>, whereas the other elements are expressed in  $\mu$ g L<sup>-1</sup>, SE: standard error. <sup>†</sup> Significant different from zero at p < 0.05.

Table 3: Chemicals, mean Hysteresis Index (HI) and Flushing Index (FI) values and percentages of distinct hysteresis patterns during these events monitored from 2011 to 2015 in the Arvorezinha catchment.

Chemical	HI	FI	% Clockwise	% Counterclockwise	%Complex	
SSC	1.48	0.75	85	15	0	
DOC	2.63	-0.09	44	56	0	
Dissolved						
Cl <sup>-</sup>	-0.03	-0.20	43	57	0	
NO <sub>3</sub>	0.03	-0.34	43	43	14	
SO <sub>4</sub> <sup>2-</sup>	-0.05	-0.38	43	57	0	
Ca <sup>2+</sup>	-0.22	-0.38	38	62	0	
Cu <sup>2+</sup>	-0.36	-0.08	36	64	0	
Fe <sup>2+</sup>	-1.18	-0.15	54	46	0	
K <sup>+</sup>	0.05	-0.04	54	38	8	
Mg <sup>2+</sup>	-0.06	-0.30	38	68	0	
Mn <sup>2+</sup>	-0.67	-0.04	67	33	0	
Na⁺	0.01	-0.71	55	45	0	
PO <sub>4</sub> <sup>3-</sup>	0.69	0.28	50	50	0	
Si <sup>4+</sup>	0.05	-0.67	46	54	0	
Zn <sup>2+</sup>	-1.43	-0.10	23	77	0	
Particulate						
Mg	1.16	0.50	90	20	0	
Ca	0.56	0.59 0.38	80 80	20 20	0 0	
Cu	1.02	0.36	89	20 11	0	
Mn	1.80	0.44	90	10	0	
Fe	1.18	0.67	80	20	0	
Na	0.24	0.30	50 50	50 50	0	
K	0.42	0.50	70	30	0	
P	0.42	0.63	70 80	10	10	
Si	1.36	0.59	100	0	0	
Zn	-1.31	0.59	57	43	0	
	-1.31	0.59	31	<del>1</del> 3		

SSC: suspended sediment concentration; DOC: dissolved organic carbon









