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1 **Investigating the relationships between chemical element concentrations**
2 **and discharge to improve our understanding of their transport patterns in**
3 **rural catchments under subtropical climate conditions**

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19

20 **Abstract** - Solute and particulate elemental concentrations (C) exhibit different
21 responses to changes in discharge (Q), and those relationships are not well
22 understood in subtropical agricultural environments. The objective is to describe
23 the transport processes of different chemical elements during a set of
24 contrasted rainfall events (2011-2015) that occurred in a small rural catchment
25 under subtropical climate. The study was carried out in the Lajeado Ferreira
26 Creek catchment (1.23 km²), southern Brazil. To this end, the concentrations in
27 dissolved organic carbon (DOC), Cl⁻, NO₃⁻, SO₄⁻, ten chemical elements (in
28 either dissolved or particulate forms) and suspended sediment concentrations

29 (SSC) were determined. Metric indices were then calculated to characterize
30 their transport patterns: (i) the best fit slope between log-C and log-Q (β), (ii) the
31 coefficient of variation of C and Q, (iii) shape of the hysteresis loop and
32 hysteresis index, and (iv) the flushing index. All particulate elements along with
33 the dissolved inorganic phosphorus (PO_4^{-3}) were shown to be controlled by the
34 sediment dynamics. Geogenic elements (Fe^{2+} , Zn^{2+} , Cu^{2+} , Mn^{2+} , Si^{4+}) showed a
35 dilution effect with increasing Q values, likely because they were mainly
36 transported with subsurface and base flow. Dissolved elements that are mainly
37 supplied with fertilizers (Na^+ and Cl^-) as well as DOC showed a dilution effect,
38 although they were mainly transported by surface runoff. Finally, a chemostatic
39 behavior was found for those chemical elements (Mg^{2+} , K^+ , Ca^{2+} , NO_3^- and
40 SO_4^{2-}) that are supplied by more than one flow pathways. The results
41 demonstrate that under subtropical climate conditions, the transport of essential
42 nutrients including PO_4^{-3} and metals (in particulate form), are mainly transported
43 with surface runoff. Accordingly, runoff control on cultivated hillslopes should be
44 improved to reduce the potential contaminant supply to the river and to reduce
45 the potentially deleterious impacts that they may cause in downstream regions.

46

47 **Keywords:** rainfall events; export regimes; solutes; metric indexes; water
48 quality

49

50

51 1. INTRODUCTION

52 The intensification of agricultural activities and the limited adoption of
53 soil and water conservation practices have increased the formation of surface
54 runoff and accelerated soil erosion (Deuschle et al., 2019; Londero et al., 2018),

55 sediment, nutrient and metal transport in river systems draining agricultural land
56 around the world (Basu et al., 2010; Minella et al., 2018; Williams et al., 2018).
57 Chemical elements supplied in excess to the streams may originate from
58 anthropogenic activities, including agriculture through the supply of fertilizers
59 and agrochemicals (Thomas et al., 2016), or from natural sources, as the result
60 of rock weathering and leaching (Verheyen et al., 2015). The relationships
61 observed between the chemical element concentrations (C) and water
62 discharge (Q) result from the combined integration of biogeochemical
63 processes, the contribution of the sources, flow pathways, and factors such as
64 lithology, soil type and climate (Godsey et al., 2009; Knapp et al., 2020;
65 Rumsey et al., 2017; Verheyen et al., 2015; Von Freyberg et al., 2017).
66 Quantifying these relationships may contribute to improve our understanding of
67 the transport of solutes and particulate chemical substances from upper
68 catchment parts to the outlet.

69 The source of each element can be identified by reconstructing the
70 hysteresis phenomenon based on the C – Q relationship (Lloyd et al., 2016;
71 Wymore et al., 2019). Hysteresis analyses have been widely used and
72 progressively improved during the last several decades and they provide a
73 powerful technique for identifying the sources of both particle-bound (Duvert et
74 al., 2010; Lawler et al., 2006; Wymore et al., 2019) and dissolved substances
75 (Lloyd et al., 2016; Williams et al., 2018). Hysteresis occurs when
76 concentrations of an element, observed in either dissolved or particulate form,
77 during the rising stage of the hydrograph differ from those concentrations
78 recorded at the same discharge rate during the falling stage. The hysteresis can
79 be observed by the formation of a loop when plotting C vs Q.

80 There are three main potential interpretations of a given hysteresis loop
81 (Lloyd et al., 2016). Clockwise hysteresis suggests the rapid mobilization of the
82 chemical element of interest and the proximity of its source to the catchment
83 outlet. This situation occurs when higher concentrations are observed during
84 the event rising stage. However, when the loop is counterclockwise (i.e. when
85 the concentration of an element is lower in the rising limb while increasing in the
86 falling limb), the source of the chemical elements is expected to be located far
87 from the catchment outlet. Furthermore, the eight-type hysteresis loop can be a
88 sequence of clockwise and counterclockwise patterns. In addition, Lawler et al.
89 (2006) proposed the hysteresis index (HI) which varies according to the
90 magnitude of the rainfall event itself and, consequently, to the hydrological
91 dynamics of the catchment.

92 Recently, metric indices have been used to evaluate these C - Q
93 relationships, including (i) the best fit slope (β) for the log (C) - log (Q) linear
94 regression (Godsey et al., 2009); (ii) the ratio of the concentration and water
95 discharge coefficients of variation (CV_C/CV_Q) (Thompson et al., 2011); (iii) and a
96 combination of both indicators (Musolff et al., 2015). The combination of both
97 metric indices (β and CV_C/CV_Q) results in a unique structure for the
98 interpretation of transport patterns (Musolff et al., 2015): they can be (i)
99 chemostatic (no change in element concentration with the increase or decrease
100 in Q), (ii) they can show a chemodynamic dilution (as Q increases, the
101 concentration of the chemical element is reduced), (iii) or display a
102 chemodynamic enrichment (increase in the concentration of chemical elements
103 with increasing Q). Besides these types of behavior, the flushing index (FI),
104 which ranges between -1 and 1, quantifies the changes in concentrations

105 during a rainfall event (Rose et al., 2018; Vaughan et al., 2019). Negative FI
106 values indicate that concentrations decrease during the hydrograph's rising
107 limb. In contrast, positive FI values indicate increased concentrations during the
108 hydrograph's rising limb. The interpretation of these indices may reveal the
109 different transport patterns of the chemical elements that may be measured
110 during the monitoring of rainfall events.

111 From this new approach, Rose et al. (2018) distinguished the following
112 transport patterns: dilution for geogenic/exogenous elements (Ca^{2+} , Mg^{2+} , Si^{4+} ,
113 Cl^- , NO_3^-); constant pattern for biologically associated solutes (carbon and
114 dissolved nitrogen, NH_4^+ , K^+ , PO_4^{3-}); and enrichment for particulate substances
115 (suspended solids and total) in a dataset covering 11 years collected in
116 Pennsylvania (USA). Through the CV_C/CV_Q ratio and based on a set of seven-
117 year rainfall events monitored in southern Brazil, Piazza et al. (2018) showed
118 that the export dynamics of solutes were similar in agricultural and in native
119 forest catchments. In South Florida, a combined analysis between the CV_C/CV_Q
120 and β showed that total nitrogen (TN) presented a chemostatic behavior while
121 total phosphorus (TP) was chemodynamic; this was due to a combination of
122 fertilizer doses, low P retention in the soil and high storms (Wan et al., 2017).

123 In subtropical and tropical climate conditions, the relationship between C
124 and Q is not well known at the event-scale. In these environments the annual
125 average rainfall volume is high as well as the kinetic energy (Ramon et al.,
126 2017), which can impact the C - Q relations during rainfall events. However,
127 there is little information on the event scale for these climates. Gwenzi et al.
128 (2017) reports the scarcity of information on the scale of hydrological (Q) and
129 biogeochemistry events for tropical climate, more precisely in Africa. In Brazil,

130 water quality monitoring programs generally collect samples only four times per
131 year following a routine schedule, due to limited human and financial resources
132 applied in this program, where the measurements do not usually occur during
133 events. Few research efforts sought to understand the influence of climate and
134 land use conditions on the dynamics of solute transport under these
135 environmental conditions (Piazza et al., 2018). Agriculture is the main economic
136 activity in the Southern region of Brazil (CONAB, 2019), resulting in a high
137 pressure on natural resources, especially soil and water. Consequently, the
138 environmental impact of this activity must be a constant concern, where the
139 limited adoption of conservationist agriculture associated with the erosive
140 potential of the rains in the region, increase the degree of soil and water
141 degradation (Didoné et al., 2015; Merten et al., 2015). Therefore, it is important
142 to know the hydrological and biogeochemical behavior of rural catchments in a
143 subtropical climate like in southern Brazil, where there is no prior knowledge on
144 the subject.

145 In this study, we aim to investigate the transport processes of different
146 chemical elements during a set of contrasted rainfall events through the
147 analysis of their C - Q relationships. For this purpose, a program to monitor the
148 transport of chemical elements in a small rural catchment representative of
149 those environments found on the Southern Plateau of Brazil and in neighboring
150 regions, under subtropical climate conditions, was established in the period
151 between 2011 and 2015.

152

153 **2. MATERIALS AND METHODS**

154 2.1 Study area

155 The experimental catchment of the Lajeado Ferreira Creek – Arvorezinha
156 (1.23 km²), is located on the border of the Brazilian southern plateau (Figure 1).
157 This is a typical headwater catchment of the Jacuí River, one of the main
158 tributaries of the Guaíba lake, which supplies drinking water to more than 4
159 million people (Cargnin et al., 2013). Its altitude ranges from 580 to 730 m
160 (Figure 1). The upper third of the catchment has an undulating plateau relief
161 with slopes up to 7%. The middle and lower thirds of the catchment have a
162 much steeper topography with slope gradients often exceeding 15%.

163 The climate is classified as Cfb (subtropical super-humid with no dry
164 season and warm summer) according to Köppen (Alvares et al., 2013). The
165 average annual rainfall is ca. 1,938 mm (15 years, 2002 - 2016) and erosivity
166 index (EI₃₀) is 9,344 MJ mm ha⁻¹ yr⁻¹ (Ramon, 2017). Rainfall is well distributed
167 over the year, although a wetter season with more rainfall events of high
168 intensity can be identified during the spring season with the frequent occurrence
169 of surface runoff.

170 The soil types in the catchment are Acrisols (57%), Cambisols (10%) and
171 Leptosols (33%). The spatial and temporal variability of land use and soil
172 management was determined using a combination of satellite images and field
173 surveys using a handheld Global Positioning System device. Land use and soil
174 management in 2011 included native forest (15% of the total catchment surface
175 area), reforested zones (23.2%), pasture (6.5%), cropland (soybean + tobacco)
176 under minimum tillage (26%), cropland under conventional tillage (13%), fallow
177 (13%) and other minor uses (3.3 %). By 2015, land use and soil management
178 had significantly changed, mainly through an increase in reforested areas
179 (35%), a decrease in cropland under minimum tillage (15%), while the cropland

180 surface proportion under conventional tillage increased to 24%, and that under
181 fallow decreased to 2.2%. In contrast, the respective surface areas under native
182 forest, pasture, and other land uses did not change.

183 2.2 Hydro-sedimentary monitoring

184 The Arvorezinha catchment has been monitored since 2002, with the
185 continuous measurement of flow discharge (Q) and suspended sediment
186 concentrations (SSC) at the outlet of the catchment in a Parshall type flume
187 built in concrete with a 1.83 m width in its narrowest part. Water level and
188 turbidity were recorded through a float-type water level sensor (Thalimedes,
189 OTT HydroMet, Germany) and a turbidimeter (Model SL-2000, Solar®, Brazil),
190 respectively. All automatic equipment recorded data every 10 minutes. SSC
191 data were obtained by the turbidimeter properly calibrated with samples
192 collected manually using an isokinetic sampler (US DH-48) during rainfall-runoff
193 events as proposed by Merten et al. (2014). The 10 min interval data was used
194 to calculate the variations of the sediment yield throughout the year. Rainfall
195 intensity and depth have been measured by means of tipping bucket rain
196 gauges and pluviometers, respectively, and their location is shown in Figure 1.
197 The rainfall of the last five days (R5) before the rainfall events analyzed was
198 computed to reflect the antecedent soil moisture conditions (SILVEIRA et al.,
199 2000), which is classified into three classes according to rainfall volume: 1) dry
200 when $R5 < 36$ mm; 2) field capacity when $R5 > 36$ and < 53 mm; and 3)
201 saturated when $R5 > 53$ mm.

202 The runoff separation (the separation of surface runoff and base flow
203 contributions) was performed for the 16 rainfall events monitored, using the
204 graphical analysis method of Chow et al. (1988). The onset of the surface runoff

205 is determined by the rise of the hydrograph until the moment when the
206 discharge returned to values close to those observed just before the rising limb,
207 thus characterizing the end of the rainfall event.

208

209 2.3 Sampling and chemical analysis

210 During the rainfall - runoff events, 164 water samples were collected at
211 the catchment outlet using a isokinetic sampler. Duplicate samples were
212 collected, one of them to calculate the sediment concentration following the
213 evaporation method according to Shreve and Downs (2005) and the second
214 sample was used for determining the water chemical composition, which is
215 described in section 2.3.1.

216

217 2.3.1 Chemical analysis

218 In the laboratory, an aliquot of the sample was filtered at 0.22 μm to
219 characterize the dissolved fraction. The total element concentration was
220 determined in the non-filtered samples. Then, the particulate fraction was
221 obtained through calculating the difference between total and dissolved
222 concentrations.

223 In the filtered sample, the dissolved organic carbon (DOC) content was
224 determined after mixing an aliquot of the filtered sample with a 1:1
225 sulphochromic solution incubated in an oven for 4 h at 60°C, with a
226 spectrophotometer at 580 nm. Sulphate (SO_4^{2-}), chlorine (Cl^-) and nitrate (NO_3^-)
227) contents were measured with a high-performance liquid chromatography
228 (HPLC) - Sykam S 135 Ion Chromatography System (U.S. EPA, 1997). Finally,
229 the concentration of dissolved Ca^{2+} , Cu^{2+} , Fe^{2+} , K^+ , Mg^{2+} , Mn^{2+} , Na^+ , PO_4^{3-} , Si^{4+} ,

230 and Zn^{2+} was quantified by Inductively Coupled Plasma - Optical Emission
231 Spectrometry (ICP-OES, Perkin-Elmer) directly in the filtered water sample
232 (U.S. EPA, 1994).

233 Total content in unfiltered samples of water + sediments was assessed
234 after digestion with *aqua-regia* (U.S. EPA, 1996). In brief, a 20 mL aliquot of
235 unfiltered sample of water + sediment was inserted into Teflon tubes together
236 with 0.5 mL of HCl (37%) and 1 mL of HNO_3 (65%), followed by microwave
237 assisted digestion for 9.5 min at a temperature of 182 °C. After that, Ca, Cu, Fe,
238 K, Mg, Mn, Na, P, Si and Zn were quantified using Inductively Coupled Plasma
239 Optical Emission Spectrometry (ICP-OES, Perkin-Elmer) (U.S. EPA, 1997).
240 Then, particulate contents of these elements were obtained by calculating the
241 difference between total and dissolved contents.

242

243 2.4 Data analysis

244 The relationship between the element concentrations and the water
245 discharge was evaluated by means of metric indices: slope (β), CV_C/CV_Q ,
246 qualitative (loop) and hysteresis index. The analyses were performed for the 16
247 rainfall events monitored, evaluating the SSC, DOC, NO_3^- , SO_4^{2-} , Cl^- and the ten
248 chemical elements analyzed (Ca^{2+} , Cu^{2+} , Fe^{2+} , K^+ , Mg^{2+} , Mn^{2+} , Na^+ , Zn^{2+} , PO_4^{-3} ,
249 Si^{4+}).

250

251 2.4.1 Metric indexes

252 To obtain the β , all Q and C data were log transformed. Accordingly, a β
253 value was obtained for each element for the 16 rainfall events. A negative β
254 value reflects the decrease of element concentration with the increase of Q,

255 whereas positive values indicate an increase of both C and Q. When $\beta \sim 0$, it
256 indicates a chemostatic behavior of the element. The chemostatic pattern was
257 defined when the β was considered to be significantly different from zero, at $\alpha =$
258 0.05 if Student's t statistic $|t| > 2$ (Helsel and Hirsch, 2002; Rose et al., 2018).

259 The coefficient of variation of the concentration in each chemical element
260 was calculated (CV_C), as well as the coefficient of variation of water discharge
261 (CV_Q). The CV_C/CV_Q metric index indicates whether the behavior of the element
262 is chemostatic ($CV_C/CV_Q \leq 0.5$), or chemodynamic ($CV_C/CV_Q > 0.5$). According
263 to Musolff et al. (2015) both β and CV_C/CV_Q need to be evaluated in
264 combination.

265 In addition, we also examined hysteresis patterns in dissolved and
266 particulate form of the elements during individual rainfall events, as well as the
267 hysteresis index (HI), with the method of Lawler et al. (2006). The method is
268 based on measuring the width of the hysteresis loop in the central water
269 discharge (Q_{50}), when there is a 50% increment of the Q compared to the
270 beginning of the rainfall event. The Q_{50} is calculated with equation 1, based on
271 the minimum (Q_{min}) and maximum water discharge (Q_{max}) values during the
272 rainfall event.

273

$$Q_{50} = 0.5 \times (Q_{max} - Q_{min}) + Q_{min} \quad (1)$$

274

275 Thus, with the Q_{50} calculated, the chemical element concentration and
276 SCC for the exact corresponding moment was obtained by a compound rule of
277 three, using the closest sample collected before and after the Q_{50} time in the
278 rising and falling limbs. Therefore, HI was calculated according to equations 2
279 and 3, when the concentration in the rising limb - RL is higher than in the falling

280 limb - FL (clockwise hysteresis) and $RL < FL$ (counterclockwise hysteresis),
281 respectively.

$$HI = \left(\frac{RL}{FL}\right) - 1 \quad (2)$$

282

$$HI = \left(\frac{-1}{\frac{RL}{FL}}\right) + 1 \quad (3)$$

283

284 Where: RL is the element concentration in water discharge on the rising
285 limb of the hydrograph and FL is the concentration on the falling limb.

286

287 The Flushing Index (FI) was calculated according to the method of (Vaughan et
288 al., 2019) (Equation 4). The water discharge and the element concentration had
289 to be normalized, according to equations 5 and 6, respectively.

$$FI = C_{Q_{peak},normal} - C_{initial,normal} \quad (4)$$

290

291 Where: $C_{Q_{peak},normal}$ and $C_{initial,normal}$ are the normalized elements during
292 the peak and the beginning of each rainfall event, respectively.

293

$$Q_{i,normal} = \frac{Q_i - Q_{min}}{Q_{max} - Q_{min}} \quad (5)$$

294

$$C_{i,normal} = \frac{C_i - C_{min}}{C_{max} - C_{min}} \quad (6)$$

295

296 Where: Q_i and C_i are the water discharge and element values at time
297 step i , respectively, Q_{max} and Q_{min} are the maximum and minimum water

298 discharge values during the rainfall event, respectively, and C_{\max} and C_{\min} are
299 the maximum and minimum concentrations of each element during the rainfall
300 event, respectively.

301

302 3. RESULTS

303 3.1 Hydro-sedimentary monitoring and element concentration

304 Table 1 shows the main characteristics of the 16 rainfall events
305 monitored between 2011 and 2015. The peak flow varied from 35 to 3,399 L s⁻¹
306 while the maximum suspended sediment concentrations fluctuated from 0.1 to
307 5.4 g L⁻¹. The magnitude of peak flows followed the magnitude of the volume
308 and duration of rainfall (Table 1). However, it is important to consider the effect
309 of the previous events on the hydro-sedimentary dynamics, because even with
310 low rainfall volume, surface runoff generation, sediment and chemical element
311 transport may occur. The rainfall of the last five days (R5) are presented in
312 Table 1. It can be observed that among all the events analyzed, there was no
313 rainfall at all during the previous five days for four events and for five events the
314 R5 was < 36 mm, corresponding to a dry condition. Five other events were
315 classified as having field capacity humidity, and two rain events considered as
316 saturated soil. The SSC during this period (2011-2015) was not high, except for
317 the events that occurred on 18/09/12 and 12/07/15 (Table 1). According to
318 studies conducted in this catchment between 2002 and 2016, the minimum and
319 maximum SCC recorded values are 0 and 13.7 g L⁻¹, respectively (Minella et
320 al., 2018). However, considering the duration and the surface runoff (Q_{runoff}) of a
321 certain event, the sediment yield (SY) may still be high. An example is the event
322 of 20/07/2011 that presented low average SSC (SSC_{ave} - 0.2 g L⁻¹) and a

323 maximum value of 0.6 g L^{-1} , but due to high Q_{runoff} , generated high SY (24.8
324 ton). The opposite situation was observed for the event that occurred on
325 12/07/2015, where the SSC_{ave} was higher, but the SY was lower, because the
326 Q_{runoff} was much lower. Annual and event-based hydrographs and
327 sedimentographs during sampling can be found in the supplementary material.

328 Figure 2 shows the preferential form of transport of each chemical
329 element. A significant proportion of Ca^{2+} , K^+ , Na^+ , and Mg^{2+} transport occurred
330 in dissolved form (47, 41, 39 and 28% respectively). In contrast, Mn, Fe, P, Cu,
331 Si and Zn were mostly transported in particulate form (99, 97, 91, 90, 85 and
332 79%, respectively).

333

334 Table 1: Hydro-sedimentary variables monitored during 16 rainfall events from
335 2011 to 2015 in the Arvorezinha catchment.

336

337 3.2 Chemical element exports

338 Table 2 shows the number of rainfall events, the number of samples (n),
339 minimum, average, and maximum concentrations for each element. A large
340 number of samples were collected during the rising, peak and falling limbs of
341 the hydrograph resulting in a total of 164 samples, although for some elements,
342 mainly anions, only the samples of a restricted number of events (n= 7) could
343 be analyzed (Table 2). The analyzed samples cover a wide range of
344 concentrations, due to the variability of rainfall events in terms of magnitude and
345 seasonality.

346

347 Table 2: Chemicals, total rainfall events, total of samples, maximum, mean and
348 minimum concentration, $\log(C) - \log(Q)$ regression slopes ($\pm SE$) and CV_C/CV_Q
349 of dissolved and particulate elements during events from 2011 to 2015 in the
350 Arvorezinha catchment.

351

352 3.2.1 Dilution patterns

353 Analyzing separately the metric indices β and CV_C/CV_Q , the following
354 results were obtained with $\beta < 0$ significantly different from zero ($p < 0.05$): -
355 0.32 ± 0.11 for DOC and -0.12 ± 0.03 for Na^+ ; where their concentration
356 decreased when Q increased (Table 2). Although the β value for Cl^- was not
357 significantly different from zero ($p < 0.06$), its value was also < 0 . The CV_C/CV_Q
358 values obtained for DOC and Cl^- , were 0.83 and 0.58, respectively, which are $>$
359 0.5, indicating a chemodynamic pattern (Table 2). In contrast, for Na^+ , the value
360 of CV_C/CV_Q was 0.42, indicating a chemostatic pattern.

361

362 3.2.2 Enrichment patterns

363 For SSC and for the concentrations in particulate form of all the chemical
364 elements analyzed, the β values were > 0 ($p < 0.05$), with the exception of that
365 obtained for Zn, which was not statistically different from zero ($p < 0.13$). This
366 behavior indicates an enrichment (Table 2), and it was also observed for PO_4^{3-}
367 and Zn^{2+} . Similarly, the CV_C/CV_Q ratio was > 0.5 for all elements in particulate
368 form, SSC, PO_4^{3-} and Zn^{2+} , characterizing a chemodynamic pattern. In addition,
369 for some particulate elements (Ca, K, Mg, Mn and Zn) and for PO_4^{3-} , the
370 CV_C/CV_Q ratio is ≥ 1 , which is classified as a strong chemodynamic pattern.

371

372 3.2.3 Chemostatic patterns

373 The concentration of NO_3^- , SO_4^{2-} , Ca^{2+} , K^+ , Mg^{2+} , Cu^{2+} , Fe^{2+} , Mn^{2+} and
374 Si^{4+} showed little variations with water discharge, because $\beta \sim 0$. All these
375 elements, except for SO_4^{2-} ($\beta = -0.09 \pm 0.03$, $p < 0.01$) and Ca^{2+} ($\beta = -0.04 \pm$
376 0.02 , $p < 0.03$), showed no significant difference from zero, at $p < 0.05$. The
377 chemostatic pattern was confirmed by CV_C/CV_Q ratio values ≤ 0.5 for NO_3^- ,
378 SO_4^{2-} , Ca^{2+} , K^+ and Mg^{2+} (Table 2). However, Cu^{2+} , Fe^{2+} , Mn^{2+} and Si^{4+} were
379 associated with a CV_C/CV_Q ratio > 0.5 , which would indicate a chemodynamic
380 pattern (Table 2).

381 When the two metric indices are combined, it is possible to obtain better
382 insights into the transport patterns of the analyzed elements (Gwenzi et al.,
383 2017) (Figure 3). A major part of the elements was characterized by an increase
384 in their concentration with increasing water discharge ($\beta > 0$; $\text{CV}_C/\text{CV}_Q \geq 0.5$).
385 Furthermore, those elements such as dissolved PO_4^{3-} and Zn^{2+} were among
386 those with the strongest chemodynamic behavior ($\text{CV}_C/\text{CV}_Q \geq 1$). In contrast,
387 those elements with $\beta \sim 0$ and $\text{CV}_C/\text{CV}_Q < 0.5$ showed a weak relationship
388 between their concentration and the water discharge, mainly Mg^{2+} , K^+ , Ca^{2+} ,
389 NO_3^- and SO_4^{2-} (Figure 3). Na^+ , Cl^- and DOC exhibit a dilution behavior.

390

391 3.3 Hysteresis and flushing analysis

392 The occurrence of a clockwise pattern was more frequent ($> 50\%$) for K^+ ,
393 Fe^{2+} , Mn^{2+} and Na^+ , SSC and for all particulate elements (Table 3). PO_4^{3-} and
394 particulate Na presented equal proportions of clockwise (50%) and
395 counterclockwise (50%) hysteresis behaviors. All particulate elements and SSC,
396 except for Zn, presented positive HI values. Fe^{2+} , Mn^{2+} and particulate Zn,

397 showed negative HI values (-1.18, -0.67 and -1.31, respectively), but the
398 clockwise hysteresis pattern predominated. Among the dissolved chemical
399 elements, only NO_3^- , K^+ , Na^+ , PO_4^{3-} and Si^{4+} showed positive HI values,
400 although NO_3^- did not show a well-defined hysteresis pattern (Table 3).

401 For Cl^- , SO_4^{2-} , Ca^{2+} , Cu^{2+} , Mg^{2+} , Si^{4+} , Zn^{2+} and DOC, a dominance of
402 counterclockwise patterns occurred (> 50%) which is supported by their
403 negative HI values (Table 3). The only exception was observed for DOC and
404 Si^{4+} that showed a positive HI value, with the occurrence of a few events with
405 high HI values, resulting in a positive average value.

406 Similarly, all particulate elements and SSC presented positive FI values,
407 indicating that their concentration increased during the hydrograph's rising limb
408 (Table 3). All the dissolved elements, except for PO_4^{3-} , together with DOC
409 showed negative values for FI (Table 3), indicating that their concentration
410 decreased during the hydrograph's rising limb.

411

412 Table 3: Chemicals, mean Hysteresis Index (HI) and Flushing Index (FI) values
413 and percentages of distinct hysteresis patterns during these events monitored
414 from 2011 to 2015 in the Arvorezinha catchment.

415

416 When plotting FI vs HI, additional information may be derived to improve
417 our understanding of the transport and the sources of these elements during the
418 investigated rainfall events (Figure 4).

419 In Figure 4a, the elements with negative FI and positive HI are found
420 (NO_3^- , Si^{4+} , K^+ , Na^+ and DOC). These elements show a dilution effect and they
421 are supplied by nearby sources, which is particularly well defined for DOC. The

422 dilution effect indicates that this element reaches the river from one of the main
423 flow pathways, which is hypothesized to be surface runoff, since organic carbon
424 concentrations are higher in the topsoil layers. However, the slight
425 predominance of the counterclockwise direction for DOC may be surprising
426 (56%, Table 3), as we expected a clockwise predominance when examining the
427 hysteresis behavior alone without considering other indices (Figure 4a). All
428 other elements are characterized by HI values close to zero, suggesting the
429 absence of well-defined hysteresis patterns (clockwise, counterclockwise, or
430 complex).

431 All the elements in particulate form, except Zn, SSC and dissolved PO_4^{3-}
432 are characterized by an increase in their concentrations during the rising limb
433 (Figure 4b). Furthermore, the source of these elements is expected to be in the
434 vicinity of the catchment outlet, being delivered during the rising limb and
435 depleted during the falling limb (Figure 4b). Particulate Zn is an exception
436 because its main behavior is counterclockwise according to the HI value (its
437 source is expected to be far from the catchment outlet) (Figure 4d). This
438 analysis suggests that at the beginning of an event there is an increase of these
439 element during the rising limb and it is supplied by more distant sources.
440 However, for most events (57%, Table 3), the particulate Zn shows a clockwise
441 pattern. For most events, the HI value was positive and close to zero for Zn. For
442 only two events the HI values were more negative (-4.69 and -5.65 for the
443 events occurred in 01/10/2011 and 20/09/2013, respectively), which makes the
444 result for this element inconclusive.

445 Cl^- , SO_4^{2-} , Ca^{2+} , Mg^{2+} , Cu^{2+} and Mn^{2+} showed a dilution effect and they
446 are likely supplied by distant sources, well characterized for Ca^{2+} , Cu^{2+} , Mn^{2+} ,

447 Fe^{2+} and Zn^{2+} that present more negative values (-0.22, -0.36, -0.67 and -1.18,
448 respectively) (Figure 4c). The other three elements (Cl^- , SO_4^{2-} , Mg^{2+}) were
449 characterized by HI values close to zero (-0.03, -0.05 and -0.06, respectively),
450 suggesting that they are supplied by closer sources during the event, than the
451 other four elements found in Figure 4c.

452

453 **4. DISCUSSION**

454 The rainfall events investigated in the current research covered a
455 pluriannual period, including different seasons and climatic phenomena. The
456 year 2015 was classified as El Niño (NOAA) for Southern Brazil with annual
457 rainfall exceeding 2,498 mm in the study site (Barros, 2016). In the agricultural
458 crop season of 2011/2012, a La Niña phenomenon was registered, which may
459 cause periods of drought in Southern Brazil (NOAA). Therefore, different
460 hydrological behavior occurs in the catchment and they were covered by the
461 period of sampling.

462 The chemical elements that can be strongly adsorbed by clays and iron
463 oxides, were mainly transported in particulate form, including P, Fe and Mn
464 with 91, 97 and 99%, respectively. In contrast, large proportions of ions such as
465 Ca, K and Na were transported in dissolved form (48, 41 and 39%,
466 respectively). However, it should be noted that these values refer to the average
467 value calculated for the entire set of 16 rainfall events (Figure 2). When
468 analyzing the dissolved and particulate proportions for each individual rainfall
469 event, these proportions can be different. Accordingly, during those rainfall
470 events with high SSC, almost 100% of the elements with a high affinity with
471 sediment particles, such as P, Mn, and Fe, were transported in particulate form.

472 For events with lower sediment concentrations, the proportion of these
473 elements in dissolved form increased. In contrast, elements with a low affinity
474 for sediment particles, such as Ca and K, were mainly transferred in dissolved
475 form whatever the SSC conditions observed during the event.

476 Three different transport regimes are described for the analyzed
477 elements in the next sections, namely: (i) chemodynamic - enrichment, (ii)
478 chemodynamic – dilution and (iii) chemostatic.

479

480 4.1 Drivers for enrichment patterns

481 All the elements in particulate form (Ca, Cu, Fe, K, Mg, Mn, Na, P, Si and
482 Zn), as well as SSC and dissolved PO_4^{3-} showed a concentration enrichment
483 with increased discharges. Botter et al. (2019) observed a similar behavior for
484 total phosphorous (Particulate P + PO_4^{3-}), which shows higher concentrations
485 with increasing discharges.

486 Sediment availability for transport is dependent on the kinetic energy of
487 rainfall during previous events and on the occurrence of new sediment pulses,
488 that may be triggered by landslides or channel bank erosion (Wymore et al.,
489 2019). As the Arvorezinha catchment did not experience any landslide during
490 the study period, the main sediment supply to the stream is associated with the
491 occurrence of interill and rill erosion in cultivated areas (57% of the total
492 sediment) (Tiecher et al., 2015). However, other potential sources were outlined
493 as delivering a significant proportion of sediment and particle-bound elements.
494 Unpaved roads were shown to provide 23% of the sediment (Tiecher et al.,
495 2015) with preferential connections with the river network (Barros, 2016). This
496 high connectivity between the roads and the stream systems allows the quick

497 delivery of readily available sediment to the stream (Figure 1). These results are
498 consistent with the findings of the current research showing that the sources of
499 particle-bound elements are close from the monitoring station (as these
500 parameters showed a clockwise pattern in > 50% of the events), which may
501 explain the rapid mobilization of the sediment and the elements bound to or
502 contained in the particles (Table 3, Figure 4).

503 PO_4^{3-} was the only compound in dissolved form that presented the same
504 behavior as the particulate elements (Figure 4). Bender et al. (2018) evaluating
505 the P (dissolved and particulate form) dynamics in the same place of the
506 present study and through eight rain events already verified a strong
507 correspondence between the PO_4^{3-} behavior with its particulate form. This is
508 likely related to the saturation of soils with the use of high doses of phosphate
509 fertilizers for tobacco cultivation. Pellegrini et al. (2010) showed that sediment
510 eroded from soils cultivated with tobacco were saturated with phosphates and
511 their desorption capacity was extremely high and fast. The dissolved form of
512 nutrients essential to the development of agricultural crops, such as PO_4^{3-} , is
513 the form that they will be available in the aqueous medium and can trigger new
514 reactions or processes (Tang et al., 2019). In this catchment, a high availability
515 of PO_4^{3-} is found in water and sediments, as already observed in a previous
516 study (Bender et al., 2018). A study conducted in France and compiling 40-year
517 records of water quality data from 293 monitoring stations showed that the
518 mitigation of phosphorus pollution reduced the concentrations in this substance
519 and altered the corresponding C – Q slopes (Moatar et al., 2017). Therefore,
520 understanding how the transfer of these elements occurs is of fundamental
521 importance, and is linked to the climatic conditions associated with the

522 characteristics of the catchment responsible for generating runoff and erosive
523 processes.

524

525 4.2 Drivers for dilution patterns

526 Fe^{2+} , Zn^{2+} , Cu^{2+} and Mn^{2+} exhibited an increase in their concentration in
527 the rising limb, but during the event, other flow pathways with low concentration
528 in these elements have reached the catchment outlet and contributed to a
529 dilution phenomenon (Figure 3). All these elements displayed counterclockwise
530 patterns, reflecting their supply by sources that are located far from the
531 catchment outlet (Figure 4c).

532 More specifically, Fe^{2+} , Si^{4+} and Mn^{2+} are released by natural weathering
533 processes and they remain in soluble form in the soil water compartment.
534 Previous studies that examined the hydrological dynamics in the Arvorezinha
535 catchment revealed that subsurface and base flows may supply 60 to 80% of
536 the streamflow during a rainfall event in autumn and winter (Barros, 2016;
537 Robinet et al., 2018). Accordingly, the greater frequency of counterclockwise
538 hysteresis behavior patterns may be explained by an important contribution of
539 subsurface flow during rainfall events.

540 Dissolved Cu^{2+} and Zn^{2+} showed a similar behavior than Fe^{2+} , Si^{4+} and
541 Mn^{2+} (Figures 3 and 4). The increased concentrations measured at the
542 beginning of the rainfall event were likely due to the release of these elements
543 by weathering processes, as young soils (Acrisols, Cambisols and Leptsols) of
544 basaltic origin are found in this catchment. In addition, in this area, low inputs of
545 Cu^{2+} and Zn^{2+} are expected from the use of fertilizers and agrochemicals

546 (exogenous solutes), which reinforces the different behavior observed
547 compared to that of DOC and dissolved Na^+ and Cl^- .

548 Na^+ , Cl^- and DOC are preferentially transported with surface runoff, due
549 to their high concentrations in the topsoil layer. The high concentration of Na^+
550 and Cl^- in water may be explained using inorganic fertilizers for tobacco
551 cultivation (NaNO_3 and KCl), which occupied 15 to 20% of the total catchment
552 surface area between 2011 to 2015. Finally, organic carbon is concentrated in
553 the topsoil layer due to inputs from cover crops; although the levels found may
554 vary depending on land use and soil management (Boix-Fayos et al., 2017;
555 Endale et al., 2017).

556

557 4.3 Chemostatic patterns

558 In general, Mg^{2+} , K^+ , Ca^{2+} , NO_3^- and SO_4^{2-} showed a very weak C - Q
559 relationship. Several sources (anthropogenic and geogenic) may supply these
560 elements, which can come from the storage of these solutes in the soil and by
561 surface runoff. Thus, elements such as K^+ , may not have their concentration
562 altered with the increase of Q, because all flow pathways to the river may have
563 similar concentrations, causing no dilution effect nor enrichment with the
564 changes in the water discharge. In White Clay Creek, Pennsylvania – USA, the
565 K^+ was the only element among those analyzed that presented a chemostatic
566 behavior and a complex hysteresis pattern, as it can be considered a biological
567 solute, geological weathering and additional K^+ inputs from exogenous sources
568 such as fertilizer (Rose et al., 2018). In Arvorezinha catchment, the production
569 of the tobacco requires high inputs of lime and industrial fertilizers
570 (anthropogenic source) containing K^+ , NO_3^- and SO_4^{2-} . Both ions have a low

571 bridging energy to clay, oxides, and organic matter, which may facilitate their
572 release into the soil and possibly be transported by subsurface flow towards the
573 catchment outlet. In addition, another fraction of these elements can be carried
574 by surface runoff, due to their higher concentration on the topsoil layer. Na⁺,
575 NH₄⁺, K⁺, Mg²⁺, Ca²⁺ were already shown to be associated with geogenic
576 sources in a study conducted under a subtropical climate in Brazil (Piazza et al.,
577 2018). In Nevada, USA, eight small river catchments were monitored (0.5 to 2.3
578 km²) and during the 8-years study period, Mg²⁺, K⁺, Ca²⁺, Cl⁻, Na⁺ and SO₄⁻²
579 revealed a chemostatic behavior under temperate climate, although an opposite
580 situation was observed for NH₄⁺ and NO₃⁻ (Hunsaker and Johnson, 2017).

581

582 **5. CONCLUSIONS**

583

584 The results show the strong correspondence between the magnitude of
585 the rainfall event and the preferred form of transportation of the chemical
586 elements. The water discharge is linked to the magnitude of the rainfall event
587 and the generation of surface runoff, the main responsible for the transport of
588 sediment and potential contaminants to the river. This can be attributed to the
589 strong erosive potential of rainfall in subtropical region combined to use and soil
590 management, that have an impact on the hydrological dynamics of the
591 catchment. Practices that may reduce the supply of water (by surface runoff)
592 and sediment to the river network will also minimize the transport of elements in
593 particulate form (Ca, Cu, Fe, K, Mg, Mn, Na, P, Si and Zn), which is the
594 preferred form of chemical elements transport in this catchment. In the same

595 way, PO_4^{3-} concentration increased with the surface runoff, having the same
596 behavior of particulate elements, preferentially transferred by surface runoff.

597 Most of the dissolved chemical elements are associated to both element
598 sources (geogenic/anthropogenic) and flow pathways (base, subsurface and
599 surface flow). Thus, elements that are not preferably transported by surface
600 runoff, such as geogenic (Fe^{2+} , Zn^{2+} , Cu^{2+} , Mn^{2+} and Si^{4+}), reduce their
601 concentration with the increase in water discharge, and the dilution process
602 occurs. DOC , Na^+ and Cl^- also have a dilution regime export (concentrations
603 decrease with the increase of Q). However, their source is anthropogenic
604 through fertilization and soil organic matter and mainly transported by the
605 surface runoff.

606 The chemostatic regime export is linked to chemical elements that are
607 supplied by more than one flow pathway (Mg^{2+} , K^+ , Ca^{2+} , NO_3^- and SO_4^{2-}), due
608 its potential contribution by both geogenic and anthropogenic source.

609 Overall, the results show that despite a small data set, it was possible to
610 understand the transport of some important chemical elements (nutrients,
611 geochemicals and metals) in a subtropical environment and it could be better
612 understood when the different metric indices were used together. These
613 insights are fundamental to design strategies to control surface runoff and
614 erosive process, because as we have observed, they seem to be the main
615 responsible for the chemical elements transport in this region.

616

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618

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625

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808 Figure 1 – Location of the study site in Brazil and the hydro-sedimentary
809 monitoring within the catchment.

810

811 Figure 2 – Average distribution of chemical elements in dissolved and
812 particulate forms during the rainfall events monitored from 2011 to 2015.

813

814 Figure 3: Plot of $\log(C) - \log(Q)$ regression slope – β versus CV_C/CV_Q for all
815 elements determined in the current study. The plotted values correspond to the
816 mean of those measurements conducted on all the samples analyzed during
817 the 2011 – 2015 period. Error bars represent ± 1 Standard Error β . SSC:
818 suspended sediment concentration; DOC: dissolved organic carbon.

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820 Figure 4: Plots of median hysteresis index versus median flushing index for
821 chemical across all sampled rainfall events from 2011 to 2015. SSC: suspended
822 sediment concentration; DOC: dissolved organic carbon. HI and FI values are
823 normalized considering their maximum value equal to 1.

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Table 1: Hydro-sedimentary variables monitored during events from 2011 to 2015 in the Arvorezinha catchment.

Data	T min	R	R5 -----mm-----	Q _{total}	Q _{base}	Q _{runoff}	Q _{peak} L s ⁻¹	C %	SSC _{peak}	SSC _{ave}	SY ton
20/07/2011	1340	127	44	92.7	37.5	55.2	3399.4	43.4	0.6	0.2	24.8
01/10/2011	350	17	0	0.81	0.61	0.2	57.7	1.2	0.2	0.1	0.1
05/07/2012	1460	95	0	16.5	10.6	6.2	536.4	6.4	0.4	0.2	2.4
26/08/2012	2040	27	15	1.0	0.9	0.1	44.2	0.5	0.2	0.1	0.3
09/09/2012	540	26	0	9.1	8.9	0.2	35.6	0.7	0.3	0.1	0.1
18/09/2012	430	110	27	44.3	19.3	25	1470.8	22.7	5.3	1.1	20.6
12/03/2013	480	27	58	5.6	4.7	0.9	235.6	3.4	0.2	0.1	0.6
24/08/2013	*	64	16	0.9	**	**	45.5	**	0.1	0.1	0.1
20/09/2013	1980	84	18	47.6	22.2	25.4	785.3	30.2	2.2	0.7	70.6
26/10/2013	245	36	52	6.3	3.6	2.7	463.6	7.5	1.0	0.4	3.3
19/03/2014	430	34	35	2.1	1.3	0.9	125.9	2.5	0.7	0.3	0.9
23/07/2014	950	83	0	26.7	17.0	9.7	1034.5	11.7	1.1	0.2	6.4
06/09/2014	390	42	8	7.4	5.2	2.2	258.4	5.2	0.4	0.2	1.2
30/09/2014	950	42	51	20.2	12.5	7.7	520.5	18.2	0.2	0.1	1.9
08/07/2015	1370	48	36	19.7	17.8	1.9	536.4	4.0	0.3	0.1	2.2
12/07/2015	*	87	86	22.5	17.0	5.5	599.5	6.3	5.4	0.4	6.6

* Events that lasted more than one day. ** This event was long, and the rainfall was of low intensity; accordingly, during the period of sampling, the flow level did not rise. T = duration of the rainfall, R = rainfall depth, R5 = antecedent rainfall of the last five days, Q_{total}= total flow, Q_{base}= total base flow, Q_{runoff}= total surface runoff; Q_{peak}= peak flow, C = surface runoff coefficient, SSC_{peak}= peak suspended sediment concentration, SSC_{ave}= average suspended sediment concentration, SY= sediment yield.

Table 2: Chemicals, total rainfall events, total of samples, maximum, mean and minimum concentration, log (C) – log (Q) regression slopes (\pm SE) and CV_C/CV_Q of dissolved and particulate elements during events from 2011 to 2015 in the Arvorezinha catchment.

Chemical	Events	Data	Maximum	Mean	Minimum	β	CV_C/CV_Q
SSC	16	164	4.4	0.3	0.0	0.52 (\pm 0.08) [†]	1.16
DOC	15	114	10.4	1.8	0.0	-0.32 (\pm 0.11) [†]	0.83
<i>Dissolved</i>							
Cl ⁻	7	53	16.4	3.0	0.6	-0.10 (\pm 0.05)	0.58
NO ₃ ⁻	7	53	2.7	1.2	0.4	-0.06 (\pm 0.03)	0.23
SO ₄ ²⁻	7	53	5.3	2.7	0.6	-0.09 (\pm 0.03) [†]	0.24
Ca ²⁺	16	164	4333.7	2183.8	660.9	-0.04 (\pm 0.02) [†]	0.16
Cu ²⁺	16	142	31.6	2.1	0.0	0.05 (\pm 0.06)	1.05
Fe ²⁺	16	163	2034.6	303.7	0.0	0.07 (\pm 0.12)	0.89
K ⁺	16	164	4548.3	2611.6	534.0	-0.01 (\pm 0.02)	0.13
Mg ²⁺	16	164	1273.3	743.0	297.5	-0.02 (\pm 0.01)	0.11
Mn ²⁺	16	126	22.0	3.6	0.0	0.08 (\pm 0.09)	0.74
Na ⁺	16	143	4492.4	1210.7	0.0	-0.12 (\pm 0.03) [†]	0.42
PO ₄ ³⁻	16	150	642.5	27.7	0.0	0.23 (\pm 0.08) [†]	1.41
Si ⁴⁺	16	163	10223.6	1621.7	0.0	0.01 (\pm 0.08)	0.63
Zn ²⁺	16	164	77.7	10.6	0.3	0.14 (\pm 0.06) [†]	0.70
<i>Particulate</i>							
Ca	12	105	72709	2191.3	0.0	0.62 (\pm 0.11) [†]	2.05
Cu	11	95	98.6	12.3	0.0	0.35 (\pm 0.14) [†]	0.96
Fe	12	118	111452.0	13408.7	290.3	0.66 (\pm 0.07) [†]	0.97
K	12	103	28784.1	3218.0	0.0	0.46 (\pm 0.09) [†]	1.02
Mg	12	110	18021.1	1710.1	0.0	0.51 (\pm 0.08) [†]	1.10
Mn	12	118	4631.5	472.5	9.0	0.77 (\pm 0.08) [†]	1.03
Na	11	89	9438.0	1376.1	0.0	0.45 (\pm 0.15) [†]	0.99
P	12	114	1936.1	203.8	0.0	0.60 (\pm 0.07) [†]	0.90
Si	12	112	53759.4	9383.1	0.0	0.65 (\pm 0.08) [†]	0.91
Zn	11	81	489.4	25.3	0.0	0.20 (\pm 0.13)	1.48

SSC: suspended sediment concentration - g L⁻¹; DOC: dissolved organic carbon (mg L⁻¹); Cl⁻, NO₃⁻, SO₄²⁻ are expressed in mg L⁻¹, whereas the other elements are expressed in μ g L⁻¹, SE: standard error. [†] Significant different from zero at $p < 0.05$.

Table 3: Chemicals, mean Hysteresis Index (HI) and Flushing Index (FI) values and percentages of distinct hysteresis patterns during these events monitored from 2011 to 2015 in the Arvorezinha catchment.

Chemical	HI	FI	% Clockwise	% Counterclockwise	%Complex
SSC	1.48	0.75	85	15	0
DOC	2.63	-0.09	44	56	0
<i>Dissolved</i>					
Cl ⁻	-0.03	-0.20	43	57	0
NO ₃ ⁻	0.03	-0.34	43	43	14
SO ₄ ²⁻	-0.05	-0.38	43	57	0
Ca ²⁺	-0.22	-0.38	38	62	0
Cu ²⁺	-0.36	-0.08	36	64	0
Fe ²⁺	-1.18	-0.15	54	46	0
K ⁺	0.05	-0.04	54	38	8
Mg ²⁺	-0.06	-0.30	38	68	0
Mn ²⁺	-0.67	-0.04	67	33	0
Na ⁺	0.01	-0.71	55	45	0
PO ₄ ³⁻	0.69	0.28	50	50	0
Si ⁴⁺	0.05	-0.67	46	54	0
Zn ²⁺	-1.43	-0.10	23	77	0
<i>Particulate</i>					
Mg	1.16	0.59	80	20	0
Ca	0.56	0.38	80	20	0
Cu	1.02	0.44	89	11	0
Mn	1.80	0.66	90	10	0
Fe	1.18	0.67	80	20	0
Na	0.24	0.30	50	50	0
K	0.42	0.51	70	30	0
P	0.86	0.63	80	10	10
Si	1.36	0.59	100	0	0
Zn	-1.31	0.59	57	43	0

SSC: suspended sediment concentration; DOC: dissolved organic carbon

Figure 1

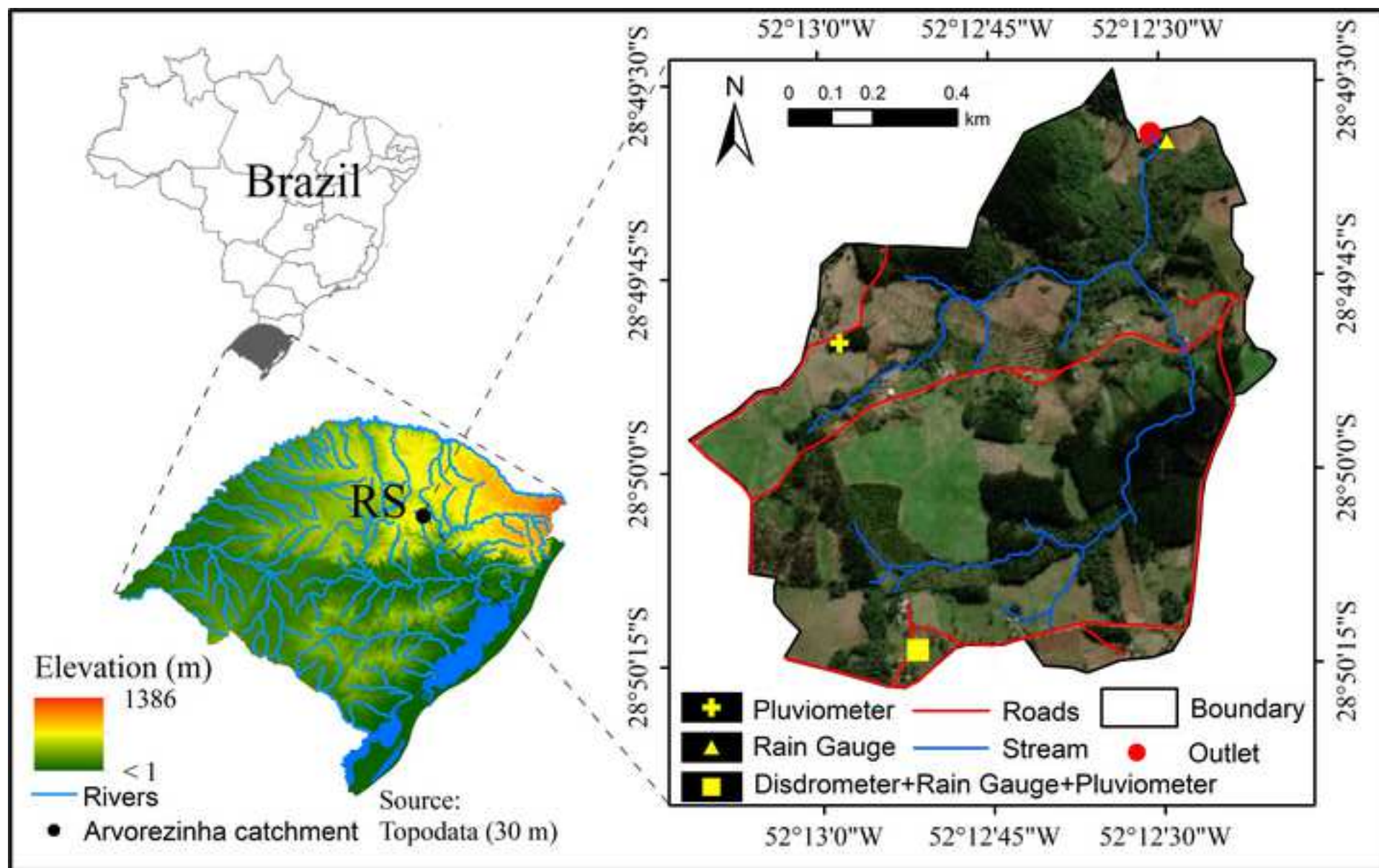


Figure 2

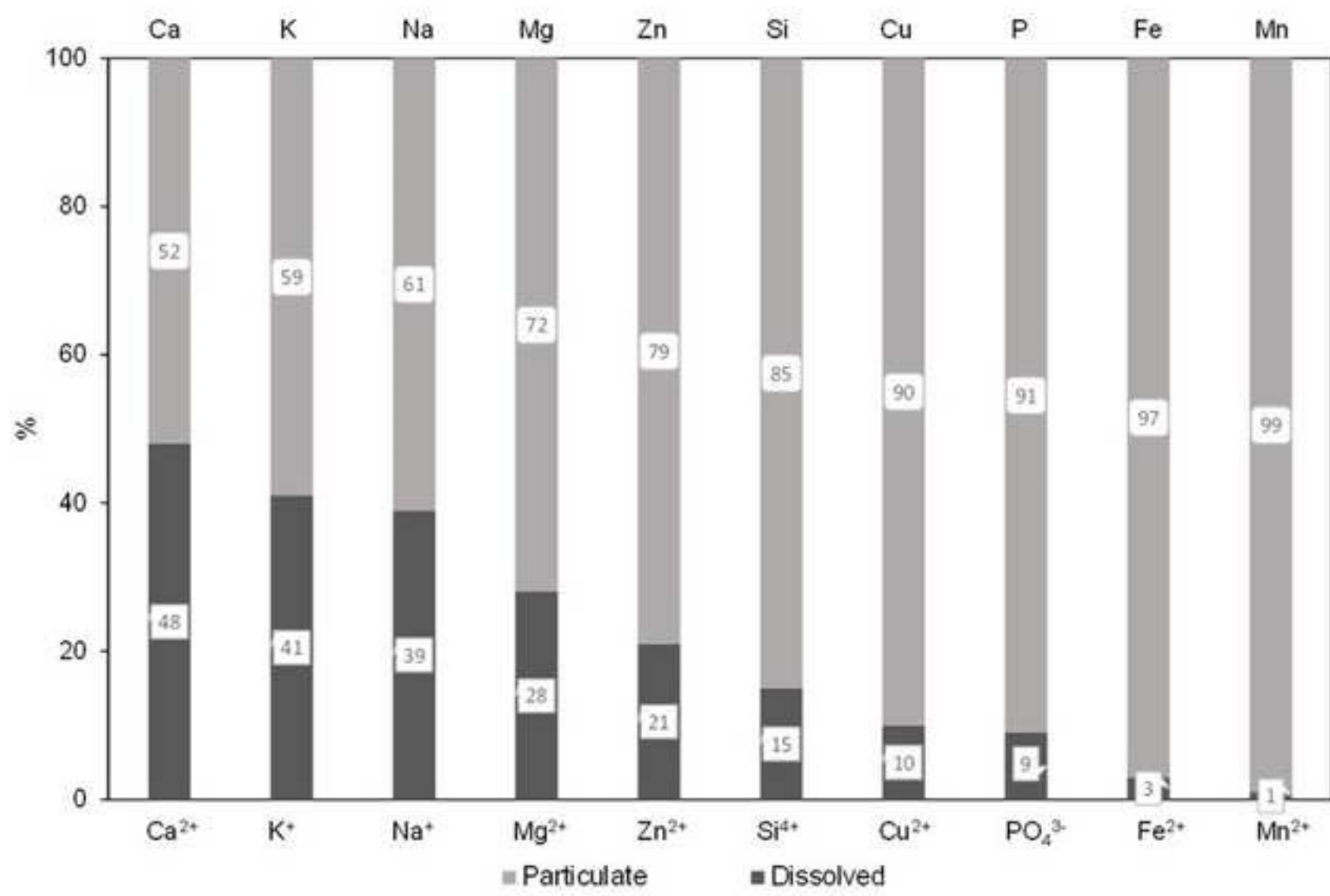


Figure 3

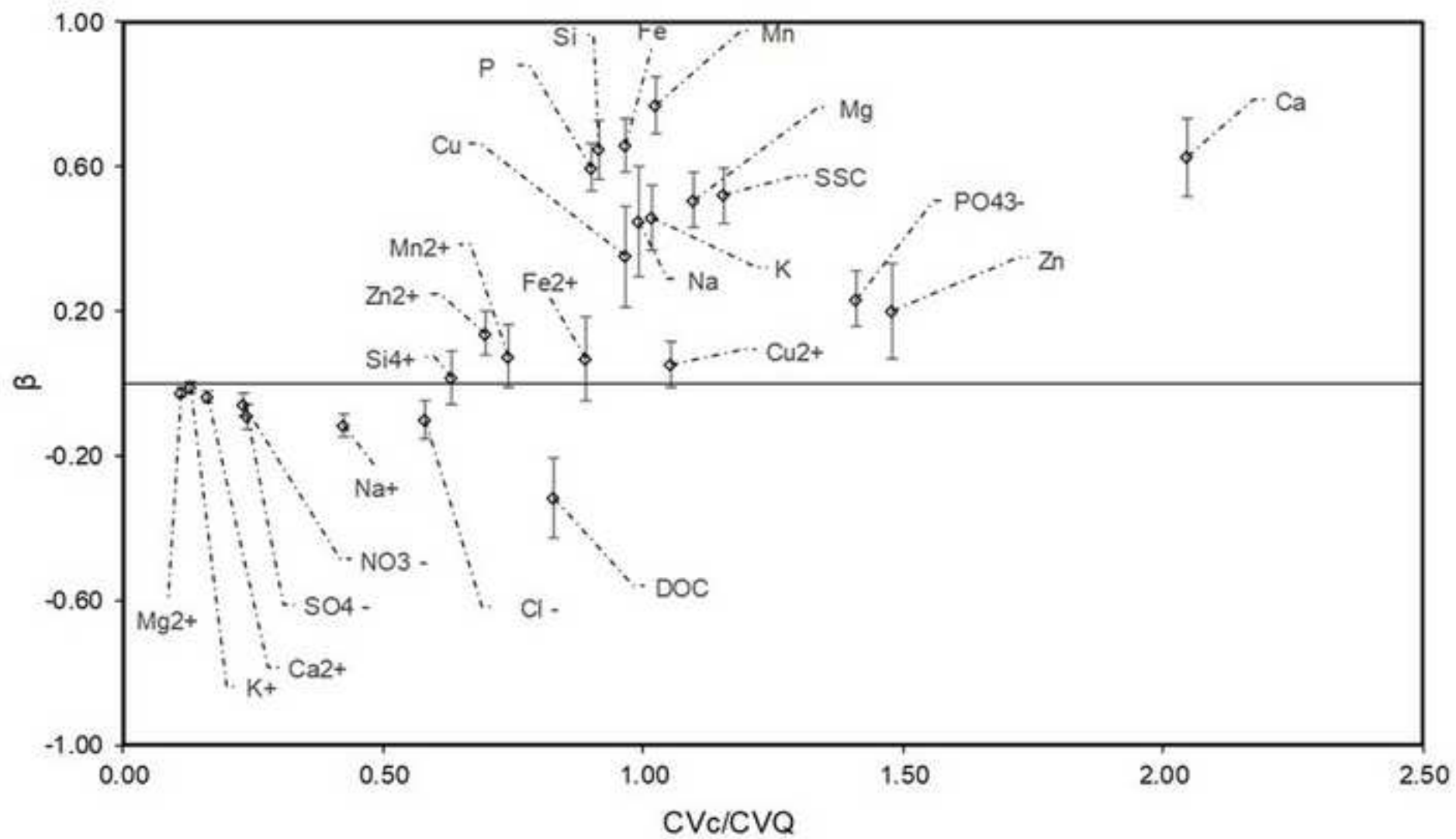


Figure 4

