

Characterization of oxide scales formed on Alloy 82 in nominal PWR primary water at 340 degrees C and in hydrogenated steam at 400 degrees C

C. Duhamel, M. Sennour, Frédéric Georgi, C. Guerre, E. Chaumun, J. Crepin,

E. Heripre, I. de Curieres

▶ To cite this version:

C. Duhamel, M. Sennour, Frédéric Georgi, C. Guerre, E. Chaumun, et al.. Characterization of oxide scales formed on Alloy 82 in nominal PWR primary water at 340 degrees C and in hydrogenated steam at 400 degrees C. USNRC 2015 Meeting – International Cooperative Group for Environmentally Assisted Cracking, May 2015, Ann Harbor, United States. cea-02500813

HAL Id: cea-02500813 https://cea.hal.science/cea-02500813v1

Submitted on 12 Jun 2020

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés. Characterization of oxide scales formed on Alloy 82 in nominal PWR primary water at 340°C and in hydrogenated steam at 400°C

<u>C. Duhamel¹</u>, M. Sennour¹, F. Georgi², C. Guerre³, E. Chaumun^{1,3}, J. Crépin¹, E. Héripré⁴, I. de Curières⁵

¹ MINES ParisTech, PSL Research University, Centre des Matériaux, Evry, France
² MINES ParisTech, PSL Research University, CEMEF, Evry, France
³ CEA, DEN, DPC, SCCME, LECA, Gif-sur-Yvette, France
⁴ Ecole Polytechnique, Laboratoire de Mécanique des Solides, Palaiseau, France
⁵ IRSN, Fontenay-aux-Roses, France







SCC of Alloy 82

Initiation tests performed in hydrogenated steam at 400°C

- SCC initiation is difficult in Alloy 82 \rightarrow test acceleration
- Previous results on Alloy 600 show that :
 - Stress corrosion cracks exhibit similar features
 - Trend between heats is respected
 - Same activation energy for initiation
- 1. Representativeness of SCC tests in hydrogenated steam
- 2. Description of the oxide scale formed on Alloy 82
 - Effect of weld (composition and/or welding process)
 - Effect of thermal treatment

Hydrogenated steam 400°C vs Primary water 360°C

Materials and oxidation conditions

	С	Si	Mn	Ρ	S	Cu	Мо	Ni	Cr	Со	Nb	Ti	Fe
RCC-M	<0.1	<0.5	2.5/3.5	<0.03	<0.015	<0.5	-	>67	18/22	<0.1	2/3	<0.75	<3
Weld A	0.025	0.07	2.57	0.004	<0.001	<0.01	-	71.7	19.12	0.04	2.41	0.1	3.07
Weld B	0.014	0.17	2.88	0.002	0.017	<0.01	0.05	72.9	18.15	0.01	2.83	<0.01	2.3



Weld		В		
Welding process	Gas tungsten	arc welding (GTAW)	Flux core arc welding (FCAW)	
Metallurgical state	As-welded	Thermally-treated 7h - 600°C	As-welded	

Oxidation tests performed on coupons with 1 μ m mirror finish in stainless steel autoclaves.

HYDROGENATED STEAM

- 400°C
- P_{tot} = 200 bar, P(H₂) = 800 mbar
- 1500h / 2500h

PRIMARY WATER

- 340°C
- [H₂] = 37 mL / kg H₂O
- 1500h / 2500h

Oxidation in hydrogenated steam at 400°C

Weld		В		
Welding process	Gas tungsten	arc welding (GTAW)	Flux core arc welding (FCAW)	
Metallurgical state	As-welded	Thermally-treated 7h - 600°C	As-welded	
Oxidation duration (h)	2500	1500/2500	2500	

SEM observations





Duplex structure:

- Discontinuous outer layer with pyramidal and platelet-like crystallites
- Continuous inner layer

Oxide penetrations (depth < 500 nm)

XPS depth profiles

Hydrogenated steam Weld A – AW – 2500h



XPS spectra: Ni 2p_{3/2}

Hydrogenated steam Weld A – AW – 2500h





- Shape and binding energies consistent with NiFe₂O₄
- Ni(OH)₂ cannot be excluded



XPS spectra: Cr 2p_{3/2}



- Before sputtering: hard to distinguish NiCr₂O₄ from Cr₂O₃
- Cr with the oxidation state +III is detected for the longest sputtering time



Marchetti et al. (2015)

XPS spectra: Fe 2p

Fe 2p 2p_{3/2} 2p_{1/2} Intensity (arbitrary units) **30 s** 10 10 s Ð 11.2 () S 740 730 720 710 Binding Energy (eV)

• Before sputtering:

consistent with $NiFe_xCr_{2-x}O_4$ or $Fe_xCr_{3-x}O_4$

- Slight shift towards lower energies:
 - Fe/Cr \searrow in NiFe_xCr_{2-x}O₄? Allen et al (1989)

McIntyre at al. (1977)

• Fe (III) \rightarrow Fe (II)?

Hydrogenated steam Weld A – AW – 2500h





XPS depth profiles

Hydrogenated steam Weld A – AW – 2500h



XPS depth profiles

Hydrogenated steam Weld A – AW – 2500h Weld A – TT – 2500h



at. %



Hydrogenated steam Weld A – AW – 2500h

Outer layer: Ni- and Fe-rich cristallite

Inner layer: Cr-rich Small amounts of Mn

Alloy: Fluctuation of composition Strong localized Cr depletion

Hydrogenated steam Weld A – AW – 2500h



Outer layer: NiFe₂O₄ / NiFe_xCr_{2-x}O₄

outer cristallites STEM-HAADF Inner layer MnS particles 100 nm (c) Cr_2O_3 1120 0112 1012 **ī104** Alloy ī012 5 nm [2201] Hydrogenated steam Weld A – AW – 2500h

Outer layer: NiFe₂O₄ / NiFe_xCr_{2-x}O₄

} Inner layer:

- NiCr₂O₄

- Cr₂O₃

outer cristallites STEM-HAADF Inner layer Oxide penetration MnS particles 100 nm TEM-BF

Hydrogenated steam Weld A – AW – 2500h

Outer layer: NiFe₂O₄ / NiFe_xCr_{2-x}O₄

- Inner layer:

- NiCr₂O₄

- Cr_2O_3



Alloy:

Recrystallized sub-layer Thickness: 150 – 300 nm Nanograins: varying composition

Oxidation in primary water at 340°C

Weld	Α			
Welding process	Gas tungsten arc welding (GTAW)			
Metallurgical state	As-welded	Thermally-treated 7h - 600°C		
Oxidation duration (h)	1500	1500		

SEM observations

As-welded

Heat-treated



Discontinuous outer layer with pyramidal crystallites and filaments

Heat-treated sample

Primary water Weld A – TT – 1500h



Heat-treated sample

Primary water Weld A – AW – 1500h



Heat-treated samples

Primary water Weld A – TT – 1500h





As-welded sample

Primary water – 1500h





Summary



HYDROGENATED STEAM – 400°C



Heat-treated

- Similar oxidation kinetics
- Duplex Cr-rich inner layer:
 - NiCr₂O₄ in primary water
 - Cr₂O₃ in hydrogenated steam
- Oxide penetrations



Conclusions (1/2)

- Environment:
 - → Surface oxide layer not strictly identical but consistent with literature data



Marchetti et al. (2015)

→ Similar oxidation tests performed on 304L SS show a difference in the oxidation kinetics but not in the oxide structure

Cissé et al., JNM (2012)

Conclusions (2/2)

- Alloy microstructure:
 - For a given environment, no effect of weld composition, welding process or thermal treatment on the oxidation kinetics and oxide structure
 - Different sub-surface states: recrystallized or high density of dislocations
 - No effect on the continuous inner layer
 - But the morphology of oxide penetration strongly differs

Origin of the recrystallized layer?

As-prepared, non oxidized sample





 \rightarrow Machining by EDM + grinding + mirror polish

Thank you for your attention