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## Perspectives in using Raman spectroscopy for characterizing the microstructure of plutonium-bearing materials

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### INTRODUCTION

In the frame of the development of uranium-plutonium mixed oxide fuels for Sodium-cooled Fast Reactors (SFRs), characterizing nuclear materials by various techniques is paramount.

These fast neutron reactors imply the use of a (U,Pu)O<sub>2-x</sub> ceramic fuel with a Pu/(U+Pu) content between 19 and 30 mol.%. Furthermore, the physico-chemical and microstructural properties of such fuels, such as chemical homogeneity, oxygen stoichiometry (O/(U+Pu) ratio) and crystallographic structure, have to meet precise criteria for being introduced in the reactor core.

As evidenced in numerous studies by various experimental techniques, a supplementary difficulty with such high plutonium content is the existence of a miscibility gap comprised within the UO<sub>2</sub>-PuO<sub>2</sub>-Pu<sub>2</sub>O<sub>3</sub> region<sup>1,2</sup>. Its presence is conditioned by the difference in the possible oxidation states adoptable by the two constituting cations. The miscibility gap itself is composed of three sub-domains, consisting in closely related cubic-type phases and its extent is a function of Pu content and temperature. The temperature of phase separation, *i.e.* the temperature at which this bi- (or tri-) phasic domain appears, depends on the oxygen to metal ratio (hereinafter O/M ratio) of the material. Recent high temperature X-ray diffraction (HT-XRD) studies allowed observing *in situ* the phase separation occurring in oxygen-hypostoichiometric uranium-plutonium mixed oxides with high plutonium contents<sup>3-5</sup>.

Nevertheless, HT-XRD characterizations on (U,Pu)O<sub>2-x</sub> were performed on powder and thus did not allow observing *in situ* the microstructure changes induced by the phase separation phenomenon<sup>3,4</sup>.

As highlighted by recent studies by Talip *et al*<sup>6</sup> and Elorrietta *et al*<sup>7</sup>, Raman microscopy is a promising tool for characterizing the physico-chemical properties such as, among many others, the cation distribution homogeneity, the grain size, the crystal defects that are of main interest for the production of nuclear fuels.

The development of a new *in situ* Raman device dedicated to handling transplutonium-bearing materials is currently in

progress in our laboratory (ATALANTE facility, CEA Marcoule, France).

We propose to present at the Plutonium Futures 2018 conference our first results obtained on U<sub>0.75</sub>La<sub>0.25</sub>O<sub>2-x</sub> and *in situ* high temperature measurements on CeO<sub>2-x</sub> as the authorization of handling plutonium-bearing materials is not obtained yet. By the time of the conference and thanks to the fruitful collaboration existing between JRC Karlsruhe and our laboratory, we will be able to present our first results on (U,Pu)O<sub>2-x</sub> samples as well.

### RESULTS

The results below were obtained on a U<sub>0.75</sub>La<sub>0.25</sub>O<sub>2-x</sub> sintered pellet. The sample was prepared by gel-supported precipitation, also referred as sol-gel external gelation<sup>8</sup>. The whole preparation route is presented elsewhere<sup>9</sup>. The following SEM picture (Fig. 1.) shows the microstructure of the polished U<sub>0.75</sub>La<sub>0.25</sub>O<sub>2-x</sub> sintered pellet. The estimated mean grain size is <5 μm.

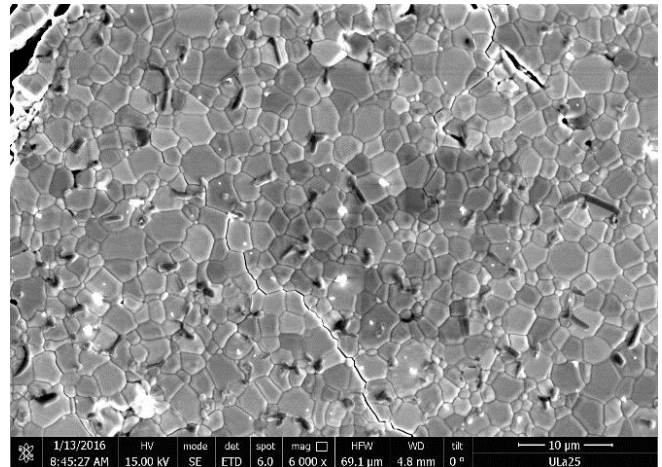


Fig. 1. SEM picture of the U<sub>0.75</sub>La<sub>0.25</sub>O<sub>2-x</sub> sintered pellet

The following Raman imaging measurements were performed at CEMHTI, Orléans, France using an InVia Reflex Renishaw system with a 514 nm LASER source (5mW and x50 objective). A 1 μm step size was used to

obtain 100x100  $\mu\text{m}$  Raman maps. The Figure 2. represents a fake colored intensity map of the  $T_{2g}$  line superimposed with the optical image.

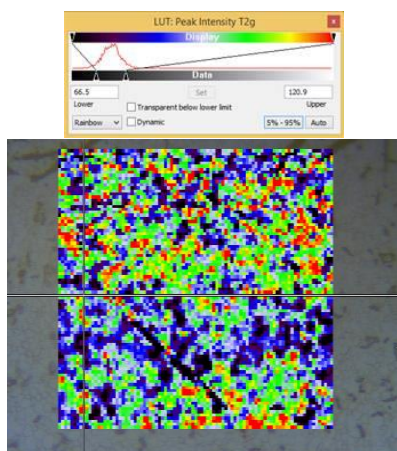


Fig. 2. Map of the Raman  $T_{2g}$  intensity line (fake colors) obtained on  $\text{U}_{0.75}\text{La}_{0.25}\text{O}_{2-x}$  (100x100  $\mu\text{m}$ )

As shown by Maslova *et al.*<sup>10</sup> on  $\text{UO}_2$  sample, a part of the  $T_{2g}$  intensity inhomogeneities is due to different orientations of the ceramic grains that allows observing the microstructure itself (grain boundaries, microstructural defects, *etc*). Thus, as revealed here, the methodology established on  $\text{UO}_2$  can be applied to mixed oxides.

In a second part, we will present our results obtained by *in situ* Raman (532 nm laser source at LGL of ENS Lyon, France) on  $\text{CeO}_{2-x}$  samples using a nuclearized version of Raman micro furnace developed by Montagnac *et al.*<sup>11</sup>. Furthermore, this unique device is associated with a setup allowing measuring, imposing and monitoring the variations in the oxygen partial pressure during a given heat treatment. As shown in Figure 3, a significant shift in the  $T_{2g}$  line is observed as a function of temperature related to the lattice thermal expansion. Moreover, both the intensity loss and the line broadening are induced by thermal agitation of anionic sublattice.

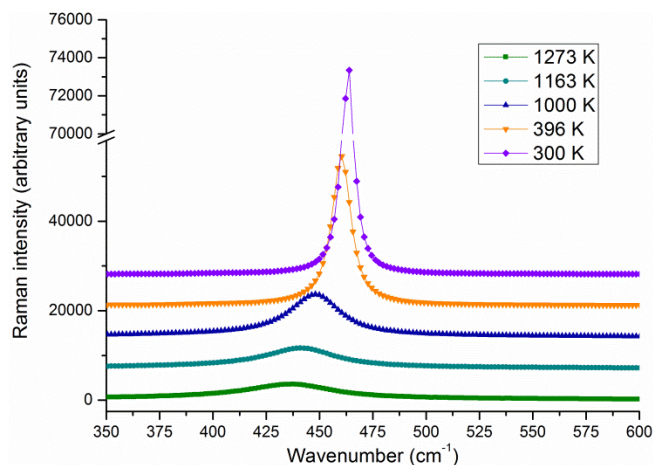


Fig. 2. Variation in the Raman spectra ( $T_{2g}$  line) of  $\text{CeO}_2$  with temperature in air

The same experiments were performed on a similar sample in reducing conditions ( $p\text{O}_2 = 1.10^{-15}$  bar) up to 1420 K. the magnitude of the shift previously observed on the  $T_{2g}$  line position was larger under these conditions. This phenomenon is explained by the *in situ* reduction of stoichiometric  $\text{CeO}_2$  to oxygen-hypostoichiometric  $\text{CeO}_{2-x}$ . We hope this promising result could be useful for determining the O/M ratio of the materials studied as a function of the thermodynamic conditions (T and  $p\text{O}_2$ ).

Finally, uranium-plutonium mixed oxide samples with 19 mol.% Pu are being characterized at JRC Karlsruhe with the experimental setup described in ref<sup>12</sup>. Experiments are performed on sintered pellets, manufactured by powder metallurgy, and show variations in the  $T_{2g}$  line position as a function of the local plutonium concentration. Raman mapping of such samples are planned in the near future.

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