

Surface Chemical Structure of Isolated Aerosols by X-ray Photoelectron Spectroscopy

Olivier Sublemontier, S. Benkoula, D. Aureau, C. Nicolas, M. Patanen, H. Kintz, X. Liu, M A Gaveau, Jean-Luc Le Garrec, E. Robert, et al.

▶ To cite this version:

Olivier Sublemontier, S. Benkoula, D. Aureau, C. Nicolas, M. Patanen, et al.. Surface Chemical Structure of Isolated Aerosols by X-ray Photoelectron Spectroscopy. European Aerosol Conference (EAC2016), Sep 2016, Tours, France. cea-02346110

HAL Id: cea-02346110 https://cea.hal.science/cea-02346110

Submitted on 4 Nov 2019

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Surface Chemical Structure of Isolated Aerosols by X-ray Photoelectron Spectroscopy

O. Sublemontier¹, S. Benkoula², D. Aureau³, C. Nicolas², M. Patanen², H. Kintz¹, X. Liu², M.A. Gaveau¹, J.L. Le Garrec⁴, E. Robert², F.A. Barreda^{1,2}, A. Etcheberry³, C. Reynaud¹, J.B. Mitchell⁴ and C. Miron²

¹NIMBE-UMR 3685/CEA/CNRS/Université Paris-Saclay, 91191 Gif-sur-Yvette cedex, France,
²Synchrotron SOLEIL, L'Orme des Merisiers, Saint-Aubin, BP 48, 91192 Gif-sur-Yvette Cedex, France,
³Institut Lavoisier de Versailles, Université Versailles-St Quentin, UMR CNRS 8180, 78035 Versailles, France,
⁴Institut de Physique de Rennes, Université Rennes 1, 35042 Rennes, France
Keywords: X-ray Photoelectron Spectroscopy, isolated nanoparticle, surface characterization, chemical structure.
Presenting author email: Olivier.sublemontier@cea.fr

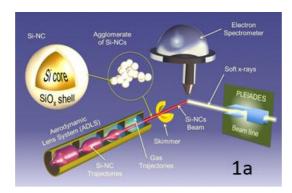
X-ray photoelectron spectroscopy (XPS) is powerful tool to investigate the surface chemical structure of any material. However, when applied to nanoobjects, this technique faces drawbacks due to interactions with a substrate, on which nanoobjects have to be deposited, and sample charging effects. We present a new experimental approach to XPS¹ based on coupling soft x-ray synchrotron radiation with an in vacuum beam of free nanoaerosols, focused by an aerodynamic lens system (Fig 1a). Two examples of experiments performed on the PLEIADES beamline at the SOLEIL Synchrotron facility are presented to illustrate the effectiveness of this approach.

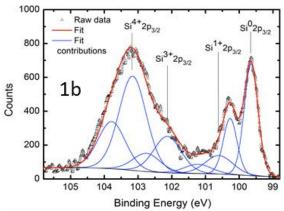
In the first example, the structure of the Si/SiO_2 interface is probed on isolated silicon nanocrystals previously oxidized with ambient air^1 (Fig. 1b) or by heat treatment under air. Full characterization of the surface has been achieved for different sizes of nanocrystals between 4 and 80 nm and with different oxidizing treatments. The technique allows probing the presence of various oxidation states at the interface and to deduce therefrom a thickness of the oxide layer. For the smaller and more oxidized nanoparticles, a relatively abrupt interface including Si=O double bonds is highlighted.

In the second example, the adsorption of water on the surface of TiO_2 nanoparticles is investigated in the gas phase³. TiO_2 free aerosols are exposed to a controlled pressure of water vapor before being analyzed directly by XPS. The technique allows here the observation of a predominantly dissociative adsorption of water on the surface of TiO_2 in its very first stage, highlighting a largely covered surface by OH groups (Fig. 1c).

- [1] Baer, D. R. et al (2010) Chem., 396, 983.
- [2] Sublemontier, O. et al (2014) *J. Phys. Chem. Lett.*, **5**, 3300
- [3] Benkoula, S. et al (2015) Sci. Rep., 5, 15088.

The experiments were performed at the PLEIADES beamline at the SOLEIL Synchrotron (Proposals No. 20110423 and 20130466). The development of the setup has received funding from the Agence Nationale de la Recherche (ANR) under Grant No. ANR-07-NANO-0031.





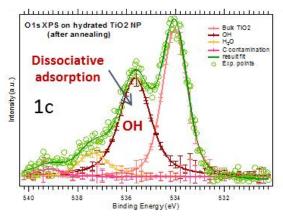


Figure 1a. XPS of isolated nanoparticles. **Figure 1b.** Si2p XPS spectrum of isolated silicon nanocrystals. **Figure 1c.** O1s XPS spectrum of hydrated TiO₂ nanoparticles.