

Behaviour of two products containing film forming amines (FFA) in the secondary circuit physico-chemical conditions of the pressurized water reactor (PWR)

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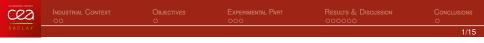
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Behaviour of Two Products Containing Film Forming Amines (FFA) in the Secondary Circuit Physico-Chemical Conditions of the Pressurized Water Reactor (PWR)



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INDUSTRIAL CONTEXT

 Lay-up of the secondary circuit of nuclear power plants (PWR)

• Secondary circuit chemistry: CHALLENGES

- Health-environment: decrease the use of hydrazine
- Safety: blockage and fouling limited
- Cost: lay up implementation and follow up simplified
- ...

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EXPECTED EFFECTS OF FFA

Protection against corrosion of all the secondary circuit (steam and liquid parts) during lay-up



Ramminger et al. 2012 Wagner et al. 2014 Anghel et al. 2014

 \Rightarrow Formation of a **protective and hydrophobic** film \Rightarrow Heat transfer performance seems to be improved

FFA INJECTION

- short duration / intermediate duration / continuous
- during normal operations, before lay up
- in the feedwater before the steam generator

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OBJECTIVES OF THIS STUDY

Study of the behaviour of 2 products containing FFA

1- Odacon[®] (Reicon) Main FFA: C₁₈H₃₇NH₂ (ODA)

2- Cetamine[®] (Kurita)

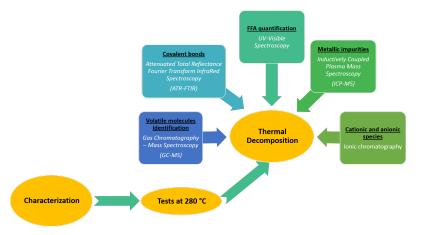
Main FFA: C₁₈H₃₅NHC₃H₆NH₂ (OLDA)

 \Rightarrow Evaluate the stability of the FFAs \Rightarrow Identify the decomposition products

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SAME METHODOLOGY FOR ODACON[®] AND CETAMINE[®]



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EXPERIMENTAL TEST DEVICE

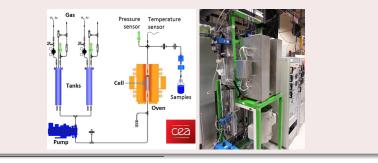
Specific device (GROZIE):

- One-pass circulation system <=> the solution is constantly renewed

 $(t_{residence} = \rho(T)V_{cell}/Q_L)$

- Or no circulation

 Designed to work in a one phase flow 25 < T(°C) < 360 and 0.1 < P(MPa) < 20



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Experimental Conditions \sim physico-chemical conditions of the secondary circuit

Product	Duration		Т	Р	pН	C _{ETA}
-	-	mg.kg ⁻¹	°C	10 ⁶ Pa	at 25 °C	mg.kg ⁻¹
Odacon®	20 min	32	280 ± 1	10 ± 0.2	9.8 ± 0.1	3.5
Ouacon	1 week	38				3.5
Cetamine®	20 min	109	280 ± 1	10 ± 0.2	9.8 ± 0.1	-
Getainine	1 week	104	200 ± 1	10 ± 0.2	9.0 <u>F</u> 0.1	-

PH is adjusted at 25 °C with diluted ammonia

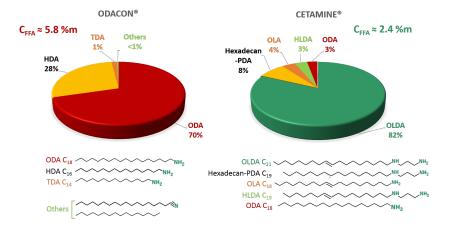
MEASURED PARAMETERS

• T, P

• C_{FFA} by UV-visible spectroscopy



Characterization of the 2 FFA mixtures



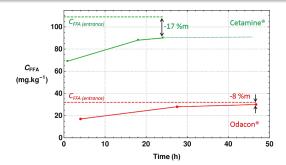
 \rightarrow No detection of metallic and ionic impurities in the 2 products except traces of Na in the Odacon $^{\textcircled{B}}$

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Thermal stability of the studied FFAs (1/2)

• After 20 min of residence time at 280 °C



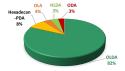
 $\Delta C_{FFA} = -8 \% m \Leftrightarrow 2 \text{ mg.kg}^{-1}$ adsorbed on the cell surface \Rightarrow no thermal decomposition of FFA in Odacon[®] $\Delta C_{FFA} = -17 \% m + 3 \text{mg.kg}^{-1}$ adsorbed on the cell surface

 \Rightarrow 14 %*m* of FFA seems to be decomposed



- For the Cetamine[®]

With the UV-visible spectroscopy analysis method: Measured absorbance \approx OLDA absorbance



 \Rightarrow 7 %*m* of FFA were thermally degraded (not detected by the analyses) OR

 $\Rightarrow \Delta C_{FFA} \approx 0 \text{ (14 \% m of FFA not detected by the analyses)}$ but OLDA was decomposed in other FFAs

Thermal stability of the studied FFAs (2/2)

• After **1 week** of residence time at 280 °C $\Rightarrow \Delta C_{FFA} = -76 \% m$ for the test with Odacon[®] $\Rightarrow \Delta C_{OLDA} = -84 \% m$ for the test with Cetamine[®]

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SACLAY	Initial products	FFA disappearance	FFA identification Covalent bonds	Ionic species	11/15

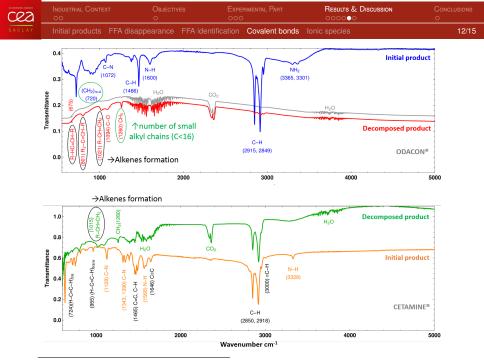
FFA IDENTIFICATION IN THE THERMALLY DEGRADED SOLUTION

Analytical method: Gas Chromatography Mass Spectroscopy (GC-MS)

- Odacon[®]
 - → Detection of ODA and HDA (∈ initial product)
 - → Disappearance of TDA
- Cetamine[®]
 - \rightarrow Disappearance of the main FFA: OLDA
 - \rightarrow \uparrow of OLA and ODA proportions
 - \rightarrow Formation of HDA

 \Rightarrow Consistent with the decomposition of OLDA in FFAs with shorter alkyl chains.

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SACLAY	Initial products	FFA disappearance	FFA identification	Covalent bonds	Ionic species	13/15

IONIC SPECIES

Analytical method: lonic chromatography

After 1 week of residence time at 280 °C

 Formation of ammonia NH⁺₄ for both products
⇒ Consistent with the detection of alkenes R-CH=CH₂ with ATR-FTIR spectroscopy analysis

NH₂ NH₂ NH₂

 Formation of carboxylates, mainly acetate CH₃COO⁻ only for Cetamine[®]

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For <u>20 min residence time</u> in the physico-chemical conditions of the secondary circuit

 \Rightarrow **No decomposition** of the Odacon[®]

- \Rightarrow Very little decomposition of the Cetamine[®]
- Significative decomposition for <u>1 week residence time</u>

⇒ the products mainly formed are FFAs which does not respond with the quantification protocol of FFAs ⇒ formation of NH_4^+ and alkenes R-CH=CH₂

PERSPECTIVES

 Determination of the *distribution coefficient* of the 2 products at steam generator temperature *i.e.* 275 °C

CES SACLAY	Objectives O		Conclusions O
			15/15



Thank you for your attention

